12-5 Current Condition and Problems with Energy

12-5-1 Energy Management and Conservation

Consumption of electric power, oxygen gas and fuel oil — all representing different types of energy — is measured by instruments installed at the control pulpit; and daily, weekly, monthly and annual reports are prepared for management review together with other operation parameters by computer processing. IDC has organized, since the beginning of 1994, an Energy Saving Committee consisting of experts of the production and maintenance sections. The organization chart of the Energy Saving Committee is shown in Figure 12-6. Concerning energy saving, this committee examines and evaluates proposals by the workers, and reports the results of such evaluations to the top management. Daily operation results are reported and discussed in the morning meeting every day, represented by section managers including the Personnel Affairs Section, chaired by Mr. Eldem, Maintenance and Utilities Director.

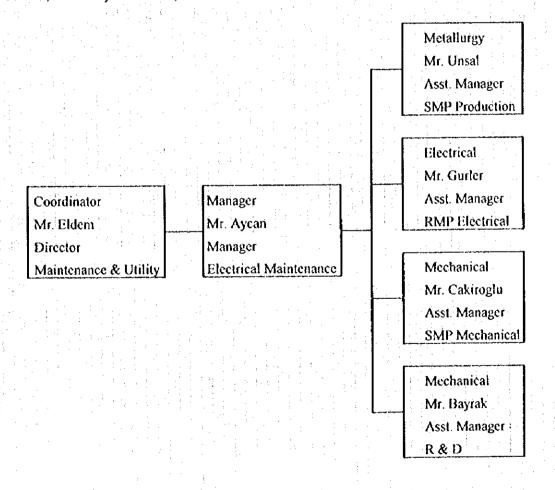


Figure 12-6 Organization Chart of Energy Saving Committee

12-5-2 Results and Plan for Rationalization of Energy Use

The following measures for energy saving are taken.

- Scrap preheating utilizing exhaust gas from the EAF: Saves about 20 kWh per ton, compared with operation without scrap preheating.
- 2. Preheating of combustion air to the ladle using its own exhaust gas is being tested.
- 3. Hot charge of billets to the reheating furnace: About 10 kilograms of fuel is saved per ton of product compared with cold charge.
- 4. The exhaust heat from the reheating furnace is used to preheat the combustion air to the reheating furnace, produce hot water and steam for heating the buildings, and preheat fuel oil.

In addition to the above measures, IDC has employed the following state-of-the-art technologies. This contributes to energy saving.

- 1. Use of large capacity transformer for EAF
- 2. Application of high power and long are operation to the EAF, resulting in reducing melting time, or short tap-to-tap time.
- 3. Use of oxy-fuel burners in the EAF makes it possible to reduce melting time and ensures uniform temperature distribution, or elimination of cold spots.
- 4. Foamy slag operation is applied to the EAF to use the arc heat more efficiently and to prevent radiation of heat from damaging the furnace wall enveloping the arcs.
- 5. The EBT system (slag-free tapping) of the EAF results in shorter tapping intervals and heat loss.
- 6. The ladle furnace (LF) contributes to increasing production of steel by relieving the EAF of the refining load. The EAF is operated for the sole purpose of melting scrap and the LF for refining the heat.
- 7. Slit rolling contributes to increasing production and saving energy.
- 8. Application of frequency converters to controlling the speed of large motors.

As shown in Table 12-9, IDC has been improving productivity by expansion of the capacities of the relevant pieces of equipment and by adopting state-of-the-art technologies. IDC furthermore plans to improve productivity. Tap-to-tap time has been remarkably improved from 80 to between 52 and 55 minutes; this will be further reduced to 45 minutes in the near future.







Consequently, electric power consumption has also been reduced from 490 to 420 kWh per ton. This is expected to be further reduced to 360 kWh per ton.

Table 12-10 shows some examples of electric power consumption in Japan. These figures range from 330 to 510 kWh per ton. In light of these figures, IDC will certainly achieve electric power consumption of 360 kWh per ton.

Table 12-9 Improvement of SMP Operation

	Design basis (*87)	'93 - '95	Future
Specifications of EAF	13 60 18 11 11 11 11 11 11 11 11 11 11 11 11		
Capacity (tons/heat)	60	70	80
Shell diameter (meters)	5.3	5,5	5.5
Transformer (MVA)	45	72	72
Operation Parameters			
Tap-to-tap time (minutes)	80	50 - 55	45 - 50
Electric power (kWh/ton)	490	405-430	360
Oxygen gas (Nm³/ton)	23	30 -35	45
Electrode (kg/ton)	3.5	1.6 - 2.3	1.5
Refractories (kg/ton)	20	10	8
Burnt lime (kg/ton)		30	25
Steel yield (%)		89	92
Technologies	1. UHP transformer		1 Additional oxy-
	2. Scrap preheating		fuel/oxy-carbon
	3. Water cooled wall a	nd roof	burner
	4. Spray water cooled	electrodes	2. Scrap upgrading
	5. Oxy-fuel burners		3. Ca-Si injection
	6. Oxy-lancing	•	4 Electromagnetic
	7. Foamy slag practice	;	stirring
	8. Long are practice		5. Ladle shrouding
	9 EBT (slag free tapp	ing)	6. Increased oxygen
	10. Ladle furnace		injection
	11. Sequence casting		7. EAF bottom
	12. Mold level control		blowing
			8. SPH improving

Source: IDC

Table 12-10 Example of Electric Power Consumption in Japan

Com-		Nomina eapacity		Trans- former	Burner (Y/N)			Actual Ton/ht	Power con-	con-	Electrode con-
		· (t)	meter (m)	(MVA)					sump- tion (kWh/t)	sump- tion (Nm3/t)	sump- tion (kg/t)
KO		100	6.4	35	7	N	7	109.4	432,5	: 8,5	3.4
KO	2	20	υ.4 4	7.5	N	N	11	109.4	513.5	29.2	4.6
	3	20	4	12.5	2	Y	14	21.6	345.1	17.3	4.6
V A		1.		41	2	N	21	73.5	397.1	33.1	2.1
KA :	1	60	5.8		3		22	126.2	363.6	33.6	2.1
GO	1	100	6.7	60		N	22 31	60.3	383.2	30.1	2
* !	2	70 70	6.3	50	2	Y					2
	3	70	5.8	35	3	T	25	74.4	410.7	27.9	
AS	1	60	5.7	45	N	Y	24	57.5	342.5	32.8	1.8
TY	}	200	8	70	10	N	19	170	341	36.4	1.3
TO	1.	140	7	60	3	N	21	132.6	380.4	24.3	1.9
	2	140	7	60	3	N	21	132.2	378.2	22.1	1.9
	3	150	8	140	N	N	22	215.8	379.9	30.6	1.6
	4	60	5.2	27.5	. 3	N	22	62.8	337.1	40.9	1.8
	5	60	5.2	27.5	. 5	N	22	62.9	341.2	39.1	1.5
	6	130	7	100	N .	Y	28	119.4	325	27	0.9
TOA	1	110	7	58	. 4	Y	26	124.5	400.2	30.6	1.6
	2	70	5.8	30	3	N	19	77.2	410.6	31.4	2.1
	3	70	5.8	30	3	N	18	77.4	405	30.7	2.4
	4	50	5.1	22	3	Y	30	43.8	440.1	20.1	2.6
	5	150	7	55	4	Υ	28	133.6	353.3	21.5	1.7
TOP	1	30	4.6	15	N	N	17	31.7	472	25.1	4.5
	2	30	4.6	15	Ν	N	17	31.8	433.3	25.2	
	3	120	6.5	56	7	Υ	19	136.4	329.1	25.1	1.6
NI	1	25	4	8.5	2	Υ	10	20.1	446.9	39.7	4.4

12-6 Current Condition and Problems with Facilities

12-6-1 Major Energy Consuming Facilities

Major energy consuming facilities are as follows:

1. Electric arc furnace: 630,000 tons per year of molten steel

- Ladle furnace: 630,000 tons per year of molten steel
 Electric power consumption: 35 kWh per ton, 22,000 MWh per year
- Rolling mill: 510,000 tons per year of rebar
 Electric power consumption: 80 kWh per ton, 41,000 MWh per year
- Reheating furnace: 510,000 tons per year of rebar
 Fuel oil consumption: 25 kilograms per ton, 12,800 kilograms per year

12-6-2 Identification of the Current Problems

(1) Problem with Major Energy Consuming Facilities

As mentioned in section 12-5-2, IDC has also been improving productivity and energy saving by expansion of the capacities and introduction of state-of-the art technologies. IDC has achieved superb operation results, of which the following are particularly outstanding:

- 1. Electric power consumption of the electric arc furnace (400 to 420 kWh/ton-MS)

 Table 12-10 indicates that there is some room to further reduce electric power consumption.
- 2. Utilization of the scrap preheater

 Use of scrap preheater is limited; half the exhaust gas is used and the equipment capacity, time available and equipment configuration limit the preheating to 70 percent of the scrap.
- 3. Hot charge
 Application is limited; Application is limited only to low temperature billets due to metallurgical problems.
- (2) Problems in Energy Consumption Already Recognized, Items Requested for Auditing IDC has envisaged reduction of electric power consumption for the electric are furnace from 420 kWh/ton-MS to 360 kWh/ton-MS. Expansion of the scrap preheater is now under study as one of the means to achieve this objective.

To set a definite target for energy saving and confirm the results of energy saving efforts, it is important to determine the heat balance of the arc furnace operation regarding it as an integral system. Determination of the exact heat balance involves enormous amounts of measurement

and calculation. Nevertheless, IDC finds it necessary to determine the heat balance of the electric arc furnace.

(3) Major Items and Points of Factory Audit

The study team understands IDC's objective and agrees with IDC that determination of the heat balance is the cornerstone for promotion of energy saving. Consequently, major items for audit concern development of a heat balance of the are furnace.

12-7 Method and Procedure of Energy Audit

12-7-1 General

As a basis for total energy control, determination of the energy balance is important. When the energy balance sheet are properly prepared, energy efficiency of an individual plant can be obtained. This enables one to compare his plants with those of other companies and helps identify their problems.

A properly prepared balance gives the entire picture of consumption of energy and comparison with other companies. The daily management of energy can be done more easily by smaller-scaled measurement, or measurements that can be made by existing on-line instruments may be enough.

The measurement of energy flow of an entire minimill plant is very difficult, and a great deal of work is required. Even measurement done only once or twice requires a project organization: many pieces of equipment and many experts, to be mobilized under well-planned preparation.

Generally, input energy flows are not very difficult to measure but output flows are. Even when they are measured, accuracy cannot easily be obtained within 20 percent. Determination of a heat balance requires a great deal of work, time, and precaution for possible dangers.

(1) Premises of the Analysis

The entire minimil! plant is a rather complicated system, the main equipment being the EAF (Electric Arc Furnace). Related pieces of equipment include: an LF (Ladle Furnace) and auxiliary systems (pumps, heat-exchangers, dust collectors, etc.). To determine the energy balance of the entire system is really difficult; and as matter of fact the energy balance alone is not enough to understand the state of energy consumption of the EAF; therefore, a measurement plan



was developed not only for the EAF but for surrounding pieces of equipment.

The EAF alone required a great deal of preparation: equipment and personnel. IDC and the study team seriously considered and prepared for equipment and personal allocation.

The preparation IDC had to make was not limited to the measurement project and the above mentioned equipment and personnel. IDC had to do some preparation work on site, on the equipment, and other necessary work (analysis, weighing, etc.).

(2) Formulation of Detailed Plan

1) Facilities, Studied and Purpose of the Audit

(a) Facilities

Ideally, the energy balance should be determined for the entire IDC factory. As already mentioned above and as the block flow diagram of Figure 12-7 below shows, the EAF itself has many inputs and outputs and hence required very elaborate work.

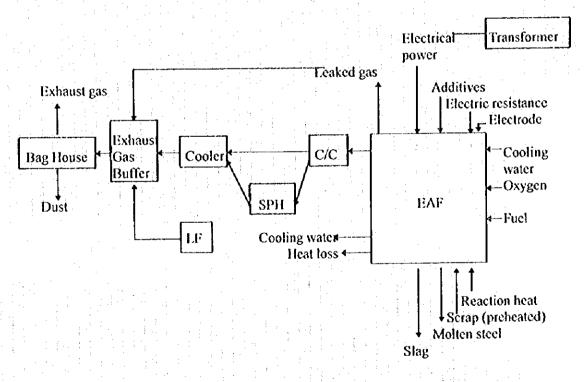


Figure 12-7 Block Diagram of Heat Inputs to and Heat Outputs from EAF

The EAF itself is a reactor: raw materials (scrap and pig iron) and additives melt at high temperatures accompanied by generation and absorption of heat.

(b) Purpose of the Audit

IDC has already attained energy efficiencies of relatively high levels. The following purposes of the audit were tailored to the conditions of IDC:

- 1. Confirmation of energy consumption levels and comparison with international levels
- 2. Management's grasp of the entire picture and basis for policy making
- 3. Enhancement of employees' morale by being exposed to the positive attitude of management
- 4. Acquisition of measuring skills throughout the whole company

12-7-2 Measuring Items, Method and Frequency

The basic measurements are limited to the EAF and its immediately neighboring equipment, as mentioned above and energy balance should be determined as clearly, simply and easily as possible for the sake of making the difficulties with measurement easier.

(1) Measurement and Estimation Items

The measurement for heat balance of the EAF was done on both inputs and outputs of energy in the following five categories:

- 1. Measurement by existing instruments: by IDC
 - 1) Consumption of electric power
 - 2) Consumption of raw materials (scrap and pig iron)
 - 3) Consumption of fuel oil and oxygen gas for each of 4 furnace burners and 1 door burner (4 furnace/1 door burners) and oxygen gas for 2 lancings
 - 4) Composition of molten steel
 - 5) Composition of slag
 - 6) Consumption of additives (burnt lime and carbon injection)
 - 7) Temperature of molten steel
 - 8) How rate of cooling water for furnace
 - 9) Inlet temperature of cooling water for furnace
 - 10) Power-on-to-power-off time
- 2. Measurement by newly prepared instruments: by JICA study team
 - 1) Temperature of scrap
 - 2) Temperature, CO/CO₂, and O₂ of the exhaust gas at C/C inlet
 - 3) Temperature, static pressure, flow rate, CO/CO₂, and O₂ of the exhaust gas at C/C

- 4) Outlet temperature of cooling water for furnace
- 5) Temperature of the roof, shell and bottom of furnace
- 6) Surrounding conditions
- 3. Analysis: by IDC
 - 1) Calorie of fuel
 - 2) Composition of injected carbon
 - 3) Composition of additives (burnt lime and carbon injection)
 - 4) Composition of scrap
- 4. Calculation/estimation from existing data: by IDC/the study team
 - 1) Weight and temperature of hot heel --- estimation by experience
 - 2) Electrode consumption --- statistic estimation
 - 3) N₂ of the exhaust gas at C/C inlet and outlet --- calculation
 - 4) Weight of output (molten steel)---statistic estimation
 - 5) Weight and temperature of slag---calculation and estimation
 - 6) Surface area of the roof, shell and bottom of furnace---IDC data
- 5. Analogy: by IDC/the study team
 - 1) Heat loss at the secondary conductors and transformer

These works were done in close cooperation between IDC and the study team.

(2) Measuring Methods

The original basic plan was proposed by the study team. Through repeated discussions and cooperation between IDC and the study team, the plan was modified and finalized to fit the actual situations. The proposed methods for measurement and analysis were based on NKK's experience which is commensurate with JIS (the Japanese Industrial Standards) including the processing of data after the measurement.

The plan for analysis and measurement for energy audit is summarized in Table 12-11. Figures 12-8 and 12-9 are the flow diagram around EAF and layout around EAF, respectively.

The measurements were done in the following method,

- 1. Continuous measurement
 - 1) Consumption of electric power
 - 2) Temperature, CO/CO2 and O2 of exhaust gas at C/C inlet

- 3) Temperature, static pressure, flow rate, CO/CO₂ and O₂ of exhaust gas at C/C outlet
- 4) Outlet temperature of cooling water for furnace
- 5) Temperature of shell and bottom of furnace
- 2. Total amount/integration
 - 1) Consumption of raw materials (scrap and pig iron)
 - 2) Consumption of fuel oil and oxygen gas for each of 4 furnace /1 door burners and oxygen 2 lancings oxygen gas for two lances
 - 3) Consumption of additives (burnt lime and carbon injection)
- 3. Instantaneous value
 - 1) Temperature of molten steel
 - 2) Flow rate of cooling water for furnace
 - 3) Inlet temperature of cooling water for furnace
 - 4) Power-on-to-power-off time
 - 5) Temperature of raw materials (scrap and pig iron)
 - 6) Surrounding conditions
- 4. Analysis values
 - 1) Composition of molten steel:
 - 2) Composition of slag
 - 3) Composition of additives (burnt lime and carbon injection)
 - 4) Composition of scrap and pig iron
 - 5) Temperature of roof of EAF
 - 6) Calorie of fuel oil
- (3) Measuring Manual of Exhaust Gas, Temperature of Cooling Water and Temperature of Furnace Body

Details are shown in Appendix-1.

Table 12-11 Plan of Analysis and Measurement for Energy Audit (IDC) 1/3

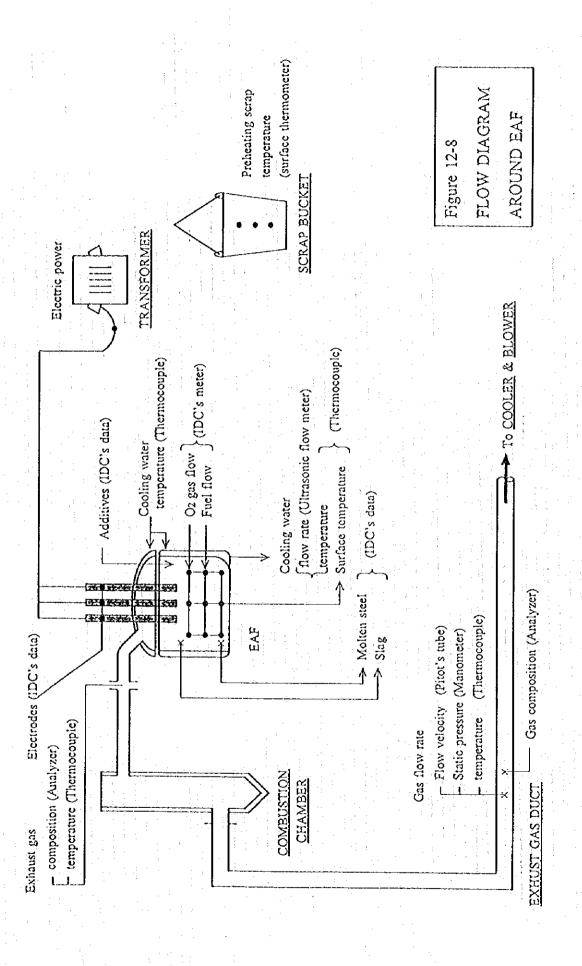
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11.4	2. Oz. for cach of 4 furnaces/1 door burners				В		
	1) Time (hr-min)	×	Clock			Start & Finish	Control room
+* 4	2) Consumption (Nm³)	×	Integrator			Integration	Control room
	3. Oil, for each of 4 furnaces/1 door burners	-			£ 1		
	1) Time (hr-min)	·				• • • • • • • • • • • • • • • • • • •	
. •		×	Clock			Start & Firush	Control room
4.4	3) Calorie (kcal)	ΣÞ	Integrator X			Integration	Control room
9		1	(виринс)			1,54	
•	4. Carbon injection	. ,			∢	; ;	
•	1) Time (hr-min)	>	Clock			Start & rimsh	Control room
	2) Consumption (kg)	X	(Calculation)		1	Heat	
	3) Composition (%)	11	(Standard)				
	S. Additives, each				∢		
	1) Time (hr-min)	×	Clock			Start & Finnish	Control room
•	2) Consumption (kg)	×	Weigher			Heat	Неат героп
	 Composition (%) 	ய	(Standard)			Heat	
_	6. Scrap (Scrap bucket)				o O		:
	 Consumption (kg) 	Z	Weigher		•	Each charge	Неат героп
	2) Composition (%)	ω	(Standard)		•		
:	 Temperature (°C) 	×	Thermocoupie	×		Selore charge	Scrap bucket
	7. Hot heel						:
	1) weight (kg)	u >	· fuc				Hear resour
	 Composition (%) Temperature (%C) 	ន់ ពោ	(Estimation)			•	
	8. Electric power				ш		
	1) Time (hr-min)	×	Clock			Start & Finish	Control room
	1) Consumption (RWh)	×	kWh meter			Every I mm. & integration	Control room
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ı	- 1	щ	(Standard)			Acat	•
		•					VI
	1) Weight (vheat)	ม)	(Calculation)		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	near S	ricat rocord
	2) Composition (%)	2	Analyzer			Champion neats	neat record

Table 12-11 Detailed Plan of Analysis and Measurement for Energy Audit (IDC) 2/3

Audit Electric Arc (Heat output) Furnace 1. Exhaust gas, C/C inlet 1) Time (hr-min) 2) Temperature (°C) 3) CO/CO, in gas (%) 4) O; in gas (%) 2) N; in gas (%) 2) Temperature (°C) 3) Static prossure (mmH-O) 4) Flow rate (Nm³/min) 5) CO/CO, in gas (%) 6) O; in gas (%) 7) N; in gas (%) 7) Cooling water 1) Time (hr-min) 2) Temperature, outlet (°C) 4) Flow rate (m³/min) 2) Flow rate (m³/min) 2) Flow rate (m³/min) 3) Temperature, inlet (°C) 4) Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 4) Temperature of roof (°C) 5) Temperature of coof (°C) 5) Temperature of coof (°C) 5) Temperature of coof (°C) 7) Temperature of coof (°C) 8) Mosten steel including bot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)	Measurement or Estimate or Estimate NA	Equipment of Analysis and Measurement Required Equipment Factory EIE JICA Local Labo Additional Automatic meter	Equipment of Analysis and Measurement Pactory FIE JICA Local Labo	1 1	Personnel Allocation JICA EIE Factory		Measuring Points
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2) Temperature (°C) 3) Static pressure (mmH ₂ O) 4) Flow rate (Nm ³ /min) 5) CO/CO ₂ in gas (%) 6) O ₂ in gas (%) 7) N ₂ in gas (%) 3. Cooling water 1) Time (hr-min) 2) Flow rate (m ³ /min) 2) Flow rate (m ³ /min) 3) Temperature, outlet (°C) 4) Temperature, outlet (°C) 4) Temperature of 2 points of wall (°C) 5) Temperature of 2 points of bottom (°C) 6) Temperature of 2 points of bottom (°C) 7) Temperature of 2 points of bottom (°C) 8) Temperature of Cool (°C) 7) Temperature (°C) 8) Motten steel including bot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%) 3) Composition (%)		Automatic moter				Continuously	C/Courler
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4) Flow rate (Nm²/man) 5) CO/CO₂ in gas (%) 6) O₂ in gas (%) 7) N₂ in gas (%) 3. Cooling water 1) Time (hr-man) 2) Flow rate (m²/man) 2) Flow rate (m²/man) 2) Temperature, outlet (°C) 4) Temperature, outlet (°C) 4. Furnace body 1) Time (hr-man) 2) Temperature of 12 points of wall (°C) 3) Temperature of 12 points of bottom (°C) 4) Temperature of Coof (°C) 5) Motten steel including bot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%) 3) Composition (%)		Digital manometer	× .			Continuously	C/Contlet
5) CO/CO- in gas (%) 6) O- in gas (%) 7) N- in gas (%) 3. Cooling water 1) Time (hr-min) 2) Flow rate (m²/min) 3) Temperature, outlet (°C) 4) Temperature, outlet (°C) 4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 2 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%) 3)	1.	Pitot, tube	×			Continuously	C/Courier
6) O: in gas (%) 7) N: in gas (%) 3. Cooling water 1) Time (hr-min) 2) Flow rate (m³/min) 3) Temperature, outlet (°C) 4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 24 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%) 5)		Automatic meter				Continuously	C/Coutlet
7) No in gas (%) 3. Cooling water 1) Time (hr-min) 2) Flow rate (m²/min) 3) Temperature, inlet (°C) 4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 2 points of bottom (°C) 4) Temperature of roof (°C) 5. Motten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%) 5.	➣	Automatic meter	· · · · · · · · · · · · · · · · · · ·		¥	Continuously	C/Courlet
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1) Time (hr-min) 2) Flow rate (m³/min) 3) Temperature, inlet (°C) 4) Temperature, outlet (°C) 4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 4 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (°S) 3))	F		
2) Flow rate (m²/min) 3) Temperature, inlet (°C) 4) Temperature, outlet (°C) 6. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of roof (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (°S)	×	Clock	1		:	One time a heat	Control room
3) Temperature, inlet (°C) 4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 4 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)	×	Magnetic flow meter x			`	One time a heat	Control room
4) Temperature, outlet (°C) 4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 4 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)		Thermocouple N				One time a heat	Inlet
4. Furnace body 1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 4 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)	įψ	Thermometer	X)	Continuously	Outlet
1) Time (hr-min) 2) Temperature of 12 points of wall (°C) 3) Temperature of 4 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)				Ω	•		
 Temperature of 12 points of wall (°C) Temperature of 4 points of bottom (°C) Temperature of roof (°C) Molten steel including hot beel Temperature (°C) Weight (kg) Composition (%) 	×	Clock	×			Continuously	Heat report
3) Temperature of 4 points of bottom (°C) 4) Temperature of roof (°C) 5. Molten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)	×	Thermocoupic -				Continuously	Hear report
4) Temperature of roof (°C) 5. Mosten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)		Thermocoupie .	×) *************************************	Continuously.	Heat report
5. Motten steel including hot beel 1) Temperature (°C) 2) Weight (kg) 3) Composition (%)	<u>ម</u>	(Standard)	X):/ :	One time during test	
1) Temperature (°C) 2) Weight (kg) 3) Composition (%)							
2) Weight (kg) (3) Composition (%)		Thermocouple x				Heat	Heat report
3) Composition (%)	ш	(Calculation)				Heat	Heat report
	×	Analyzer x				Heat	•
6. Slag					•		
1) Temperature (°C)	េជ	(Estimation)	The second secon			Heat	Heat report
2) Weight (kg)		(Calculation)			-1-4	Heat	Heat report

Table 12-11 Detailed Plan of Analysis and Measurement for Energy Audit (IDC) 3/3

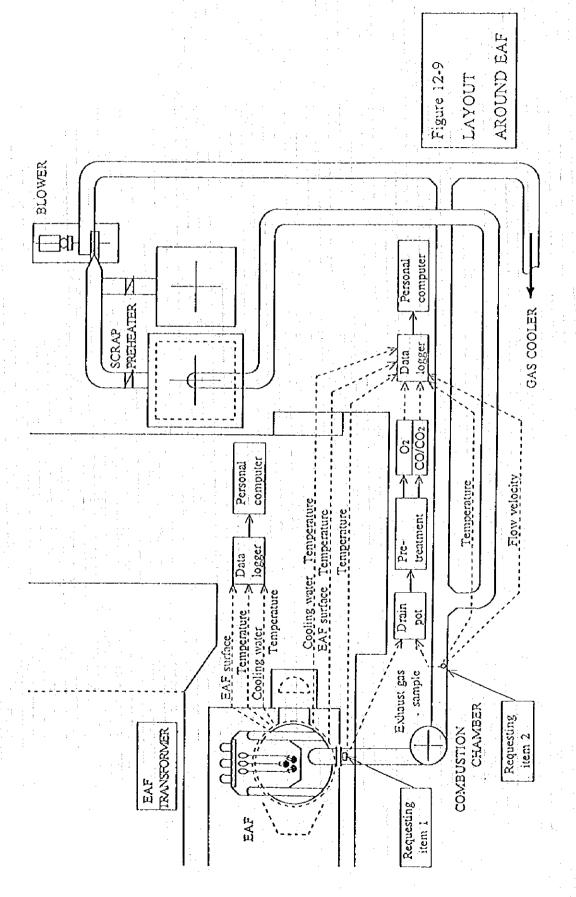
Major Items		Met	Methods of Analysis and Measurement	Kemarks
of Fromov	Subject Items and points	Measurement	nont Equipment of Analysis and Measurement Personnel Allocation	
Audit		 or Estimate	ured Eq	nterval Measuring Points
lectric Arc	Electric Arc (Others)			
'~ }	1. Operation results		9. С 3	
	2. Surrounding condition			
	1) Weather			
. •	 Atmospheric pressure (hpa) 		N Start of Acat Operation	
••	 Outdoor temperature (°C) 		N Start of near operation	non Cutsiac
•	4) Indoor temperature (°C)		N. Start of boat operation	
	5) Humidiry (%)		x Start of heat operation	tion Around EAF



13

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8



(3) Measuring Frequency

- 1. 3 days including trial measurement 1
- 2. I heat of trial measurement and 5 heats for official measurement a day (measuring time of one heat: 1 hr), Total 11 heats
- 3. Total 11 heats

12-7-3 Measuring Equipment

- 1. Study team: mainly energy outputs with the newly prepared instruments.
- 2. IDC: mainly energy inputs with the existing instruments.

Table 12-12 shows additional items, numbers and descriptions of the equipment brought by the study team. The measurements by these pieces of equipment were carried out by the study team, while measurements by existing equipment were all carried out by IDC.

Table 12-12 Equipment List of the Study Team

	ITEM	DESCRIPTION
a-1)	CO/CO ₂ analyzer 1 set	To measure CO and CO ₂ contents of the exhaust gas at C/C inlet
a-2)	CO/CO ₂ analyzer 1 set	To measure CO and CO ₂ contents of the exhaust gas at C/C outlet
b-1)	O ₂ analyzer 1 set	To measure O ₂ content of the exhaust gas at C/C inlet
b-2)	O₂ analyzer 1 set	To measure O ₂ content of the exhaust gas at C/C outlet
c)	Pretreatment unit (filter, drain pot, cooler) 1 set	To remove dusts and moisture in the exhaust gas
d-1)	Data logger 2 sets	To input into the personal computer the output data (analog signals) from the measuring devices after conversion into digital signals
d-2)	Data logger 1 set	Stand-by
e-1)	Personal computer 2 sets	To record and exhibit on the monitoring screen the output data after being converted to the digital signals by the data logger

e-2)	Personal computer 1 set	Stand-by. To be used in case of trouble of one of e-1). Actually this computer was used.
f-1)	Thermocouple PR type (JIS type R) 4 sets	To measure the temperature of the exhaust gas at the C/C inlet
f-2)	Thermocouple CA type (JIS type K) 24 sets: 1 set for outlet, 16 sets for furnace, 7 sets for spare	To measure the temperature of the exhaust gas at the C/C outlet and temperature of the furnace shell and bottom
f-3)	Thermocouple CC type (JIS type T) 10 sets (including 4 sets of spare)	To measure the temperature of the cooling water for the furnace
g-1)	Cable (100 m) for thermocouple for CA type	To connect the thermocouple to the data logger
g-2)		To connect the thermocouple to the data logger
g-3)	Cable (300 m) for thermocouple for CA type	To connect the thermocouple to the data logger
h)	Ultrasonic flow meter 1 set	To measure the flow rate of cooling water for the furnace. Not used. IDC's instrument was used.
i)	Gas sampling unit 4 pieces	Sampler for the exhaust gas at C/C inlet. Water cooled. Stainless-steel, for C/C outlet.
j-l)	Digital manometer 2 sets	To measure the dynamic and static pressure of the exhaust gas for measurement of the flow rate at the C/C outlet
i-2)	Digital manometer 2 sets	Stand-by
k)	Pitot' tube 4 sets	Used for measurement of the flow rate of the exhaust gas at the C/C outlet
i)	Surface thermometer 2 sets	To measure the surface temperature of the scrap bucket for scrap temperature
m)	Printer 1 set	To print out the results of measurement
	Transformer 3 sets	Step down transformer (200 V to 100V) and stabilizer for instruments
n-2)	Transformer 1 set	Stand-by
0)	Pyrometer (Thermometer) 2 sets	To measure the roof surface temperature and for back-up use for measurement of surface temperature of the furnace shell and bottom.

One is for temperatures higher than 800°C and the other is for temperatures lower than 500 °C

p) Equipment for moisture 1 set

To measure moisture in the exhaust gas.

12-8 Measurement Execution Procedure

12-8-1 Site Survey

(1) General

In order to plan measuring methods in detail, a preparatory site survey was done before the field survey (measuring for heat balance of Electric Are Furnace (EAF) in IDC). This site survey was done by two experts. Their activities included explanation and discussions with IDC on the plant operation, site conditions, IDC's assistance to the measurement, and measuring methods.

Site survey was carried out by the measuring experts of the study team on the investigation for:

- 1. Proper locations for installation of instruments and protection shelter,
- 2. Utility supply for field instruments (electric wiring and water supply)
- 3. Existing on-line instruments (kinds and location)

(2) Surveyors

This preparatory field survey was done by the following two experts.

Mr. Isamu Kawakami (Process D, Iron and Steel)

Mr. Tokuyoshi Kawai (Measurement, Iron and Steel)

(3) Schedule of Site Survey

1) Overall Schedule

July 8th 1996	Narita to Frankfurt	
9th	Frankfurt to Ankara	ŧ
10th	Visit to the IICA office and the EIE office	
11th	Ankara to Izmir	
	Visit to IDC	•
12th	Visit to IDC	

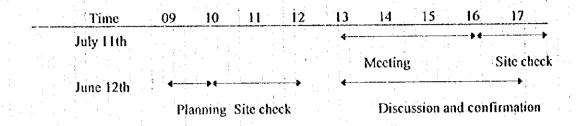
13th

Izmir to Frankfurt

14th

to Narita

2) Schedule at IDC



(4) Site Check

1) Location of Measuring Points

1. Elbow (Combustion Chamber (C/C) inlet): Fitting point of sampling probe

Approach to the fitting point

Area for preparation and maintenance for sampling probe

Feeding of cooling water for probe

Outlet of cooling water for probe

2. Duct (C/C outlet):

Probe inserting hole

Approach to measuring point for maintenance work

3. EAF cooling water:

Measuring points

Approach to measuring points

4. EAF wall and bottom:

Fitting thermocouples

Approach to fitting points

5. Scrap bucket:

Approach to bucket with hand-carried thermometers

6. Measuring instruments:

Space for 2 sets for exhaust gas and EAF cooling water

7. Others:

Operation room

2) Utility on Site

1. Electric source:

2-3 points for instruments, 200 V, 20-30 A

2. Cooling water:

For sampling probe, 3/4" diameter

3. Extension bar;

For thermometer for measuring bucket temperature, 2-3 meters

4. Others:

If necessary

3) Modifications for Measurement

1. Rag plate:

Welding a rag plate of 200 x 200 x 10 thickness to sliding duct

for U-bolt to fit sampling probe

2. Thermocouple:

Welding thermocouples to the furnace shell and bottom

3. Water tap:

Installation of a water tap for cooling water of sampling probe

4. Stage:

Preparation for measurement and maintenance

5. Others:

Prepared as necessary

4) Data Collection from Plant Instruments

1. Control room:

Timer

Watt meter

Weighing of scrap

Weighing of molten steel, etc.

2. Inlet energy:

O₂ meter

Fuel meter, etc.

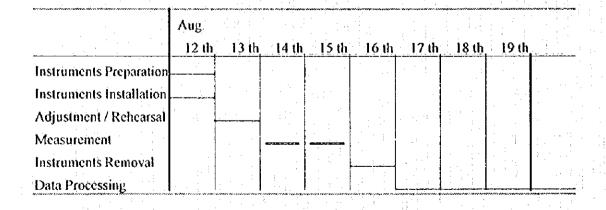
3. Others:

Ás necessary

12-8-2 Execution of Measurements

(1) Schedule of Measurements

The schedule of the measurements is shown below.



1) Aug. 12th

- 1. Meeting with EIE and IDC from 8:30 to 9:00
- 2. Unpacking of a wooden equipment and instruments box from 9:00 to 12:30
- 3. Welding of a rag plate to the sliding duct from 13:30 to 15:00
- 4. Fitting of thermocouples to the pipe for measuring outlet temperature of cooling water for furnace from 13:30 to 15:00
- 5. Welding of thermocouples to the furnace from 15.00 to 21:00
- 6. Preparation of instruments in AM and PM
- 7. Some work is not finished

This work was done during the furnace brick work shutdown time.

2) Aug. 13th

- 1. Meeting with EIE and IDC from 8:30 to 9:00.
- 2. Some work of connection not completed in Aug.12th in AM.
- 3. Adjustment of instruments in AM.
- 4. Trial measurement of one heat in PM.

3) Aug. 14th

- 1. Meeting with EIE and IDC from 8:30 to 9:00.
 - 2. Official measurement of five heats.

4) Aug. 15th

Same as Aug. 14th

5) Aug. 16th

- 1. Removing of instruments and connections in AM.
- 2. Packing instruments in the wooden box.

12-8-3 Preparatory Work

Preparatory work was carried out on Aug. 12 and 13 is as follows:

(1) Welding of a Rag Plate to the Sliding Duct (C/C inlet)

A rag plate supplied by the JICA Study Team was welded to the end of the sliding duct of EAF

(C/C inlet, combustion chamber inlet) on August 12 in the presence of the JICA Study Team.

Note: The rag plate with U-bolts is to hold a sampling probe for analysis of the exhaust gas and a thermocouple for measuring the exhaust gas temperature.

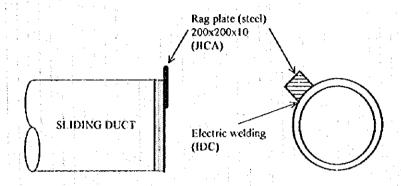
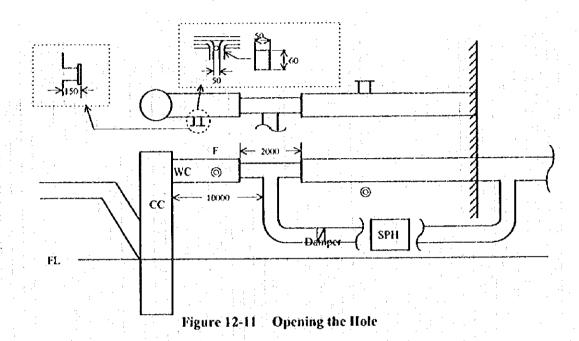


Figure 12-10 Welding of the Rag Plate

(2) Opening the Hole (C/C outlet)

Before August 12, a hole of 50×60 mm was opened at the tubular cooling duct behind C/C (combustion chamber).

Note: The 50 x 60 mm hole is for measuring the flow rate, composition and temperature of the exhaust gas.



(3) Welding Thermocouples to the Furnace Shell and Bottom

Steel plates containing thermocouple supplied by the JICA study team were welded at the furnace shell and bottom on August 12 in the presence of the JICA study team.

Shell: 12 points on the cooling pipe of center of the cooling panel. 2 points for each panel of Nos. 2, 4, 6, 8, 10 and 12

Bottom: 4 points on the bottom equally spaced

Note: Thermocouples are to measure the furnace temperature.

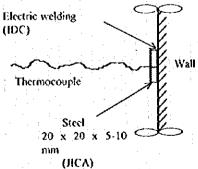


Figure 12-12 Welding Thermocouples to the Furnace Shell and Bottom

(4) Water Supply

Water tap device and basin for return water was installed near the C/C before August 12.

Note: Water is used for cooling the sampling probe at the C/C inlet.

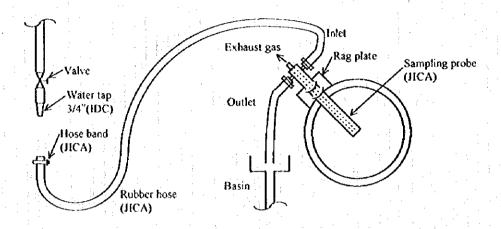


Figure 12-13 Water Supply

(5) Installation of the Holder for Cooling Water Outlet Temperature

A pipe was installed at each return water piping system before August 12.

Note: The pipe held the thermocouple to measure the furnace cooling water outlet temperature.

- 1: Return water piping system for the EBT
- 2: Return water piping system for the elbow
- 3: Return water piping system for the roof
- 4: Return water piping system for the shell-l
- 5: Return water piping system for the shell-2

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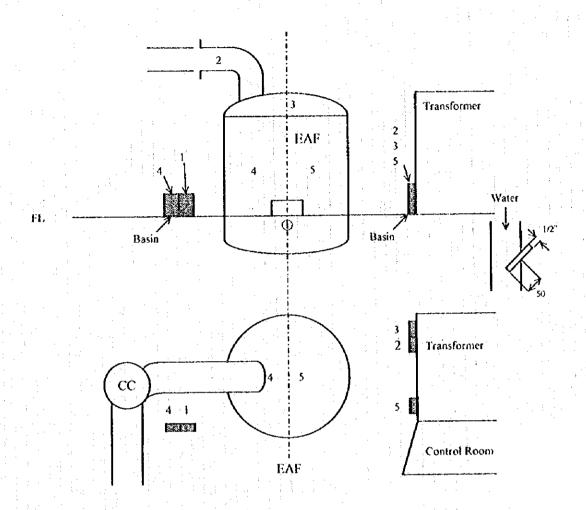


Figure 12-14 Installation of the Holder

12-9 Results of Measurement and Analysis

Measurement for the EAF heat balance was executed on following 11 heats including the trial heat.

Aug. 14 Heat No. 965729 trial

Aug. 14 Heat No. 965749 not completed in measurement of exhaust gas

Aug. 14 Heat No. 965750 not completed in measurement of exhaust gas

Aug. 14 Heat No. 965751 selected for analysis

Aug. 14	Heat No. 965752	selected for analysis
Aug. 14	Heat No. 965753	selected for analysis
Aug. 15	Heat No. 965773	
Aug. 15	Heat No. 965774	not completed in measurement of exhaust gas
Aug. 15	Heat No. 965775	not completed in measurement of exhaust gas
Aug. 15	Heat No. 965776	
Aug. 15	Heat No. 965777	

According to the procedure of JIS (the Japanese Industrial Standards), consecutive three heats of Heat No. 965751, 965752 and 965753 are selected. The operation results, measurement results, calculation of heat input and heat output and heat balance sheets are shown for three (3) selected heats in the following sections.

The study team studied and analyzed the selected three (3) heats referring to the other heats measured and observed.

12-9-1 Profile of Facilities

Table 12-13 shows profile of facilities on which measurement was executed.

Table 12-13 Profile of the Electric Arc Furnace

Name of comp	any	IDC (IZMIR DEMIR CELIK SANAYI A. S.)
Address		Foca Celik Fabrikasi 35807 Aliaga, IZMIR, TURKEY
Furnace manu	facturer	NKK Corporation (Japan)
Type	Type of furnace	AC are furnace
	Charging method	Top charge
	Tapping method	EBT system
	Other facilities (Bottom stirrer, Bottom	Water sprayed electrode, Scrap
	bubbling, Water sprayed electrode, Scrap	preheater
	preheater, etc.)	
	Nominal capacity (ton)	75
Molten steel	Bath diameter (mm)	4,475

Bath area (m²)	7.0
Bath depth (mm)	1,343
Distance between sill level and roof (mr	n) 2,192
Roof Thickness (mm)	
Radius (mm)	2,900
Diameter of electrode hole (mm)	600
Pitch circle diameter (mm)	(,300)
Proportion of water cooling area (%)	
Shell Inside diameter (mm)	5,215
Thickness (mm)	Water cooled tubular panel
Height (mm)	1,900
Height from bottom to roof (mm)	4,300
Proportion of water cooling area (%)	100
Hearth Diameter (mm)	
Thickness (mm)	725
Height of bank (mm)	332
Working door Width (mm)	
Height (mm)	
Tapping hole Diameter (mm)	159
(EBT) Depth (mm)	800
Transformer Capacity (MVA)	72
Primary voltage (kV)	34.5
Secondary voltage (V)	900
Connection	Open delta
Reactor Capacity (kVA)	12,000
Reactance (Ohm)	
Electrode Diameter (mm)	508
Oxygen Type and number	Manipulator
injection Capacity (Nm³/hr)	1 1/4 inches x 2 lancings
	(3,000 Nm³/hr, lance)
Burner Type, number and capacity	5 furnace oxy-fuel burners
	oil: 200 liter/hr, unit
	O ₂ : 500 Nm ³ /hr, unit
	1 door oxy-fuel burner
	oil: 300 liter/hr, unit

		O ₂ : 700 Nm ³ /hr, unit	-
	Kind of fuel and capacity (kg/hr)	Heavy fuel oil NO. 6	_
Carbon	Capacity and number (kg/hr)	l set	
injection	en de la companya de La companya de la co	35 kg/min	

12-9-2 Long-run Operation Results

Table 12-14 shows long-run operation results of each five (5) heats of the same grade of steel which were operated before and after sampled three (3) heats for heat balance calculation.



Table 12-14 Summary of Long-Run Operation Results of Before and After Heat Measured

			-									
	Date	Aug. 13	Aug. 14	Aug. 14	Aug. 14	Aug. 14	Aug 14	Average				
	Heat Wo	٨١	1			965750	965754	965755	965756	965757	965758	
	Charged raw material (excluding hot heel)	85.800	85.700	85.080	85.700	86.000	85.600	85.440	86.020	85.680	85.620	85.660
	1 1 1st broket (t) Scran	43.300	44,300	43.080	41.200	42.400	41.300	40.100	41.200	41,100	40.260	41.820
		27.300	26,200	26.500	28.400	32.400	31.100	29.200	27.660	27.320	27.300	28.340
	Scrape	21.000	23.000	22,200	24.000	29.000	28.000	25,500	7.000	27.300	27.300	23.430
	Piging	3 300	3.200	4.300	4.400	3.400	3.100	3.700	3,360	0.000	3.000	4.000
	1.3. 3rd bucket (t) Scrap	15.200	15.200	15.500	16.100	11.200	13,200	16.140	17.160	17.260	18.060	15.500
	2. Output (Production) (t)	77.220	77.130	76.570	77.130	77,400	77.040	76,900	77,420	77.100	77.060	77.100
	2.1. Good billet (t)	76.620	76.530	75.980	76.530	76.800	76,440	76.300	76.820	76.510	76.460	
		0.600	0.600	0.590	0.600	0.000	0.600	0.600	0.600	0.590	0.600	0.600
	3. Steel vield (%)					٠			:			
	3.1. Molten steel yield	0.06		:	٠							
	3.2. Good billet vield	89.3										
	4. Operation time (min.)		1	:			<i>3</i> .				٠	
	4.1. Tap-to-tap time	S	61	56	9	57	55	56	49	54	42	\$5
	4.2. Power on-to-power off time	47	50	- 48	43	46	45	46	44	48	45	46
	5. Consumption	:						-:				· · ·
	5.1. Electric power (kWh/t-output)	378	389	364	375	353	360	367	355	372	382	370
7	5.2. Fuel oil (kg/t-output)	4.7	4.9	4 Ci	6.1	6.5	48	0.0	4	4	4.5	4.0
	5.3. Oxygen gas (Nm³/t-output)	32.2	30.9	30.7	33.0	34.3	35.0	29.1	26.5	28.0	29.0	30.9
	Burner	17.1	16.0	14.8	18.2	19.5	18.6	15.6	15.6	15.7	16.4	16.8
	Lancing	15.1	14.9	15.9	4.8	16.8	16.4	13.5	10.9	12.3	12.6	14.3
	5.4. Burnt lime (kg/heat)	21.4	23.1	23.9	33.7	23.2	23.0	24.1	22.9	23.5	23.6	24.2
		8.5	8 4	8.8	10.0	8.6	9.7	9.1	9.7	12.6	13.5	66
	6. Composition and temperature of molten											
	steel before tapping	•		0	((6	71.0	ć	Ç		· ·	
		0.10	0.04	0.05	000		0.10	7 6	01.0		2 6	7 6
	6.2. Si (%)	0.01	0.02	0.05	0.03	0.02	0.05	0.0	0.02	0.0	0.0	70.0
	6.3. Mn (%)	0.05	0.03	0.04	0.07	0.04	0.03	2.	9 2 2 3	CO.O	CO.O	c0:0

		(000	6		.000	7000	010	6,00	1100	600	0000
0.4		2.0.2	0.00	0.010	2000	0.0.0	0.00	212	0.00	7.70.0	1.0.0	2.0
6.5. \$	(%) S	0.065	0.074	0.076	990'0	0.058	0.078	0.070	0.061	0.048	0.054	0.065
) 99	(%)	0.07	0.0	0.05	60.0	0.05	0.05	0 0 4	90.0	0.04	60.0	90.0
6.6.	Temperature (deg. C)	1620	1612	1622	1616	1630	1646	1603	1636	1601	1620	1621
7. We	Weight of hot heel (t)	10										
& &	Composition of billet (%)									-		0
8	C (%)	0.24	0.19	0.18	0.18	0.20	0.20	0.23	0.19	0.21	0.21	0.20
8.2	Si (%)	0,23	0.19	0.17	0.18	0.19	0.17	0.17	0.17	0.17	0.17	0.16
8.3	Wn (%)	0.85	0.86	0.87	98.0	0.86	0.87	0.86	0.86	98.0	0.84	0.86
84	P (%)	0.038	0.043	0.036	0.040	0.037	0.026	0.018	0.024	0.024	0.030	0.032
8.5	(%) \$	0.039	0.052	0.048	0.047	0.045	0.044	0.040	0.037	0.039	0.039	0.043
8.6		0,46	0.51	0.52	0.47	0.45	0.45	0.39	0.41	0.41	0.36	0.44
8.7		0.14	0.13	0.12	0.12	0.13	0.15	0.13	0.16	0.13	0.14	0.14
88		0.13	0.11	0.11	0.12	0.10	0.08	0.07	0.10	0.10	0.10	0.10
8.9.	Wo.	0.020	0.018	0.016	0.015	0.016	0.019	0.016	0.023	0.017	0.018	0.05
8.10.	Sn	0.019	0.021	0.024	0.019	0.019	0.021	0.017	0.016	0.017	0.015	0.02
8.11	A	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
Note:	1) Heat No. 965757 is excluded for avera	verage of pig	g iron.					1				
	2) Heat No. 965755 is excluded for avera	verage of fu	iel oil.			· · ·.	:					

12-9-3 Results of Measurement and Heat Balance

(1) Concened Personnel

Personnel are listed in Table 12-15.

Table 12-15 Personnel Involved in Measurement

Date		August 14, 199	6
Heat No.	969	5751, 965752,	965753
Company	Study team	EIE	IDC
Staff	- I. Kawakami - H. Tokano		- Suleyman Eldem - Necati Unsal
Measuring person			
Flow rate, temperature and composition of exhaust gas Outlet temperature of cooling	- T. Kawai - S. Kinoshita - N. Honda	(Bora Omurtay)(Birgul Duman)(Mehmet Sezer)	· · · · · · · · · · · · · · · · · · ·
water for furnace Electric power Oxygen gas by lancings			- Sibel Ozkan - Uguri Egeli
Carbon injection			
Burnt lime Oxygen gas and fuel oil by			- Hikmet Yuzuak
4 furnace/1 door burners Flow rate and inlet temperature of cooling			- Talip Bostanci
water for furnace Temperature of scrap bucket	- (H. Tokano)		- Duzgun Elitas
Surrounding conditions	- H. Tokano		

Note: Bracket means co-working with specialists.

(2) Summary of Results for Measured Heats

Table 12-16 Summary of Operation Results for Three (3) Measured Heats

	1				
Date	<u> </u>		Aug. 14	Aug. 14	Aug. 14
Heat No.			965751	965752	965753
I. Charge	ed raw mater	ial (excluding hot hee	1) (t) 85.400	87.400	86.000
		стар	42.100	41.1	41.2

1.2 2nd bucket (t)	28 200	26.100	27.400
Scrap	24.500	22,500	24.000
Pig iron	3.700	3.600	3.400
1.3. 3rd bucket (t) Scrap	15,100	20.200	17.400
2. Output (Production) (t)	76.860	78.660	77.400
2.1 Good billet (t)	76.260	78.050	76.800
2.2 Crop end, skull (t)	0,600	0.610	0.600
3. Steel yield (%)	14.7		
3.1. Molten steel yield		90.3	
3.2. Good billet yield	·	89.3	
4. Operation time (min.)			
4.1. Tap-to-tap time	53	53	70
4.2. Power-on-to-power-off time	43	43	57
5. Unit consumption			
5.1 Electric power (kWh/output)	363	360	379
5.2. Fuel oil (kg/t-output)	6.3	5.9	7.2
5.3. Oxygen gas (Nm³/t-output)	33.7	35.1	32.5
Burner	20.3	16.1	18.0
Lancing	13.4	19.0	14.5
5.4. Burnt lime (kg/t-output)	24.4	22.9	28.0
5.5. Carbon injection (kg/t-output)	11.4	7.6	10.4
6. Composition and temperature of molten steel			
before tapping			
6.1. C(%)	0.05	0.03	0.05
6.2. Si (%)	0.01	0.01	0.01
6.3. Mn (%)	0.04	0.04	0.05
6.4. P(%)	0.026	0.026	0.025
6.5. S(%)	0.069	0.064	0.700
6.6. Cr (%)	0.06	0.07	0.06
6.6. Temperature (°C)	1641	1673	1624
7. Weight of hot heel (t)		10 ,	
8. Composition of billet (%)			
8.1. C(%)	0.18	0.20	0.20
8.2. Si (%)	0.18	0.17	0.18
8.3. Mn (%)	0.87	0.91	0.92
8.4. P (%)	0.040	0.027	0.028
8.5. \$ (%)	0.043	0.055	0.048
8.6. Cu (%)	0.43	0.43	0.45
8.7. Ni (%)	0.13	0.12	0.15
8.8. Cr (%)	0.13	0.08	0.09
8.9. Mo (%)	0.018	0.017	0.017
8.10. Sn (%)	0.018	0.019	0.020

(3) Results of Measurments

Results of measurement for the (3) heats are shown in Table 12-17.

Table 12-17 Results of Measurement

			11
Date	Aug. 14		
Heat No.	965751		
Amount of raw materials in 1st bucket (1)		41.100	
Amount of scrap in 2nd bucket (t)		22 500	2.0
Amount of pig iron in 2nd bucket (t)	3.700	3.600	3.400
Amount of raw materials in 2nd bucket (t)	28.200	26.100	27.400
Amount of raw materials in 3rd bucket (t)	15.100	20.200	17,400
Charged raw material (excluding hot heel) (t)	85.400	87.400	86.000
Amount of scrap (t)	81.700	83.800	82.600
Amount of pig iron (t)	3,700	3.600	3.400
Output (excluding hot heel) (t)	76,860	78.660	77.400
Consumption of electric power (kWh)	27,900	28,300	29,300
Unit consumption of electric power, w ₁ (kWh/t-output)	363	360	379
Hot heel of raw material (t)	10	10	10
Unit weight of hot heel of raw material, m2 (kg/t-output)	130	127	129
Temperature of hot heel, m ₂ (°C)	1550	1550	1550
Unit consumption of scrap, (kg/t-output)	1063	1065	1067
Unit consumption of pig iron, (kg/t-output)	48	: 46	44
Unit consumption of raw materials, m3 (kg/t-output)	1111	1111	1111
Temperature of top of 1st bucket (°C)	148	215	135
Temperature of middle of 1st bucket (°C)	131	232	132
Temperature of bottom of 1st bucket (°C)	53	82	82
Mean temperature of 1st bucket (°C)	111	176	116
Temperature of top of 2nd bucket (°C)	108	121	110
Temperature of middle of 2nd bucket (°C)	118	108	128
Temperature of bottom of 2nd bucket (°C)	66	82	82
Mean temperature of 2nd bucket (°C)	97	104	107
Temperature of top of 3rd bucket (°C)	-	· · · · · · -	· · · · · · - ·
Temperature of middle of 3rd bucket (°C)	36	39	37
Temperature of bottom of 3rd bucket (°C)			
Mean temperature of 3rd bucket (°C)	1 36	39	37
Mean temperature of raw materials after SPH (°C)	93	123	97
Consumption of fuel oil at No. 1 burner (kg)	133	104	182
Consumption of fuel oil at No. 2 burner (kg)	162	107	136
Consumption of fuel oil at No. 3 burner (kg)	56	101	93
Consumption of fuel oil at No. 4 burner (kg)	112	117	124
Consumption of fuel oil at door burner (kg)	21	38	25
Consumption of fuel oil (kg)	484	467	560
Unit consumption of fuel oil, m4 (kg/t-output)	6.3	5.9	7.2
			7

Mean CO ₂ content in exhaust gas at elbow (CO ₂) (%)	14.0	17.5	12.1
Mean CO content in exhaust gas at elbow (CO)(%)	7.1	3.5	1.5
C content of molten steel before tapping (%)	0.05	0.03	0.05
Si content of molten steel before tapping (%)	0.01	0.01	0.01
Mn content of molten steel before tapping (%)	0.04	0.04	0.05
P content of molten steel before tapping (%)	0.026	0.026	0.025
Cr content of molten steel before tapping (%)	0.06	0.07	0.06
Al content of molten steel before tapping (%)	0.000	0.000	0.000
CaO content of slag (%)	23.56	23.56	23.56
Consumption of burnt lime (kg)	1860	1790	2150
Amount of slag (kg)	6868	6610	7939
Unit weight of slag, mit (kg/t-output)	89	84	103
FeO content of slag (FeO) (%)	9.01	9.01	9.01
Fe ₂ O ₃ content of slag (Fe ₂ O ₃) (%)	22.16	22.16	22.16
Consumption of carbon injection (kg)	875	595	805
Unit consumption of carbon injection (kg/t-output)	11.4	7.6	10.4
P ₂ O ₅ content of slag (P ₂ O ₅) (%)	0.48	0.48	0.48
SiO ₂ content of slag (SiO ₂) (%)	17.31	17.31	17.31
Temperature of molten steel before tapping (°C)	1641	1673	1624
Weight of hot heel of molten steel (t)	10	10	10
Unit weight of hot heel of molten steel (kg/t-output)	130	127	129
Power-on-to-power-off time (hr)	0.72	0.72	0.95
Flow rate of cooling water for roof (m³/hr)	323	323	323
Average quantity of cooling water of roof, m14b (kg/t-output)	3026	2957	3964
Mean outlet temperature of cooling water for roof (°C)	39.0	40.5	39.1
Mean outlet temperature of cooling water for roof-1 (°C)	39.1	40.6	39.2
Mean outlet temperature of cooling water for roof-2 (°C)	38.9	40.4	39.0
Inlet temperature of cooling water for roof (°C)	34.0	34.0	35.0
Flow rate of cooling water for EBT (m³/hr)	63	63	63
Average quantity of cooling water for EBT, m _{He} (kg/t-output)	590	577	773
Mean outlet temperature of cooling water for EBT (°C)	36.8	37.8	36.5
Inlet temperature of cooling water for EBT (°C)	34.0	34.0	35.0
Flow rate of cooling water for elbow (m³/hr)	117	117	117
Average quantity of cooling water for elbow, m14a (kg/t-output)	1096	1071	1436
Mean outlet temperature of cooling water for elbow (°C)	43.4	45.0	42.9
Inlet temperature of cooling water for elbow (°C)	34.0	34.0	35.0
Flow rate of cooling water for shell-1 (m³/hr)	273	273	274
Average quantity of cooling water for shell-1, m144 (kg/t-output)	2557	2499	3363
Mean outlet temperature of cooling water for shell-1 (°C)	37.0	38.3	36.8
Inlet temperature of cooling water for shell-1 (°C)	34.0	34.0	35.0
Flow rate of cooling water for shell-2 (m³/hr)	274	274	274
Average quantity of cooling water for shall-2, m ₁₄₄ (kg/t-output)	2567	2508	3363
Mean outlet temperature of cooling water for shell-2 (°C)	36.8	38.0	36.8
Inlet temperature of cooling water for shell-2 (°C)	34.0	34.0	35.0
Indoor temperature (°C)	35.0	37.0	37.0

Surface area of roof (m ²)	40	40	40
Mean surface temperature of roof (°C)	65,0	65.0	65.0
Surface area of shell (m ²)	35	35	35
Mean surface temperature of shell (°C)	41.4	42.4	41.2
Mean surface temperature of shell-1 (°C)	39,3	40.3	39.4
Mean surface temperature of shell-2 (°C)	43.5	44.5	42.9
Surface area of furnace bottom (m²)	34	34	34
Mean Surface temperature of furnace bottom (°C)	276.1	276.4	297.5
Heat in average flow of exhaust gas (1,000 kcal/min)	216	157	227
Consumption of limestone, m ₁₂ (kg/t-output)	7	7	8

Flow rate, temperature, heat content, composition of the exhaust gas at C/C inlet and outlet are shown as follows:

Figure 12-18: Heat Content and Flow Rate of Exhaust Gas at C/C-inlet, Heat No. 965751

Figure 12-19: Temperature of Exhaust Gas at C/C-inlet and outlet, Heat No. 965751

Figure 12-20: Composition of Exhaust Gas at C/C-inlet, Heat No. 965751

Figure 12-21: Composition of Exhaust Gas at C/C-outlet, Heat No. 965751

Figure 12-22: Heat Content and Flow Rate of Exhaust Gas at C/C-inlet, Heat No. 965752

Figure 12-23: Temperature of Exhaust Gas at C/C-inlet and outlet, Heat No. 965752

Figure 12-24: Composition of Exhaust Gas at C/C-inlet, Heat No. 965752

Figure 12-25: Composition of Exhaust Gas at C/C-outlet, Heat No. 965752

Figure 12-26: Heat Content and Flow Rate of Exhaust Gas at C/C-inlet, Heat No. 965753

Figure 12-27: Temperature of Exhaust Gas at C/C-inlet and outlet, Heat No. 965753

Figure 12-28: Composition of Exhaust Gas at C/C-inlet, Heat No. 965753

Figure 12-29: Composition of Exhaust Gas at C/C-outlet, Heat No. 965753

Those figures are attached to the Appendix-1.

(4) Heat Input

Heat inputs for the measured three (3) heats are calculated as shown in Table 12-18. As for calculation formulas, refer to Annex-4.

Table 12-18 Heat Input

Date								Aug 14	Aug. 14	Aug 14
Heat No.									965752	-
	ıantity of	electric p	ower,	Qı	(1000) x kc	al/t-output)	312.2	309.6	325.9
Ratio of hea	•	_						51.3	51.9	50.1

·			
Unit consumption of electric power, w ₁ (kWh/t-output)	363	360	379
(2) Potential heat of hot heel, Q2 (1000 x kcal/t-output)	41.8	40.8	41.5
Percentage of potential heat of hot heel in heat input (%)	6.9	6.8	6.4
Unit weight of hot heel of raw materials, m ₂ (kg/t-output)	130	127	129
Temperature of hot heel (°C)	1550	1550	1550
Heat content of hot heel, h ₂ (kcal/kg)	321.5	321.5	321.5
(3) Sensible heat of raw materials, Q3 (1000 x kcal/t-output	t) 11.8	15.8	12.3
Percentage of sensible heat of raw materials in heat input (%)	1.9	2.6	1.9
Unit consumption of raw materials, m3 (kg/t-output)	1111	Ш	31111
Mean temperature of raw materials after SPH (°C)	93	123	97
Heat content of raw materials after SPH hab (kcal/kg)	10.6	14.2	11.1
(4) Calorific power of fuel oil, Q4 (kcal/t-output)	63.0	59.0	72.0
Percentage of calorific power of fuel oil in heat input (%)	10.4	9.9	11.1
Unit consumption of fuel oil, m4 (kg/t-output)	6.3	5.9	7.2
Low heating value of fuel oil, q1 (kcal/kg)	10000	10000	10000
(6) Oxidation heat of electrode, Q6 (1000 x kcal/t-output)	11.2	13.1	13.7
Percentage of oxidation heat of electrode in heat input (%)	1.8	2.2	2.1
Unit consumption of electrode, m6 (kg/t-output)	1.9	1.9	1.9
C content of electrode (%)	99.7	. 99.7	99.7
Amount of C oxidation of electrode (kg/t-output)	1.9	1.9	1.9
Oxidation heat of electrode at CO ₂ formation, q ₆ CO ₂ (kcal/kg)	7829	7829	7829
Oxidation heat of electrode at CO formation, q ₆ co (kcal/kg)	2200	2200	2200
Mean CO ₂ content in exhaust gas (CO ₂) (%)	14.0	17.5	12.1
Mean CO content in exhaust gas (CO) (%)	7.1	3.5	1.5
(7) Oxidation heat of charge, Q, (1000 x kcal/t-output)	103.8	107.4	113.7
Percentage of oxidation heat of charge in heat input (%)	17.1	18.0	17.5
(7a) Oxidation heat of charged C, Q1 (1000 x kcal/t-output)	30.7	36.4	36.1
Unit consumption of scrap (kg/t-output)	1063	1065	1067
Unit consumption of pig iron (kg/t-output)	48	46	44
C content of scrap (%)	0.35	= 0.35 [0.35
C content of pig iron (%)	3.65	3.65	3.65
C content of molten steel before tapping (%)	0.05	0.03	0.05
Oxidation amount of charged C, m _{7a} (kg/t-output)	5.0	5.1	4.8
Heat of C oxidation at CO ₂ formation, q _{7cco2} (kcal/kg)	8075	8075	8075
Heat of C oxidation at CO formation, q _{7cco} (kcal/kg)	2448	2448	2448
(7b) Oxidation heat of charged Si, Q7b (1000 x kcal/t-output	5.7	21.1	21.0
Si content of scrap (%)	0.23	0.23	0.23
Si content of pig iron (%)	1.05	1.05	1.05
Si content of molten steel before tapping (%)	0.01	0,01	0.01
Oxidation amount of charged Si, m _{7b} (kg/t-output)	2.85	2.83	2.82
Heat of Si, q _{76Si} (kcal/kg)	7459	7459	7459
(7c) Oxidation heat of charged Mn, Qrc (1000 x kcal/t-outpo		13.2	13.1
Mn content of scrap (%)	0.75	0.75	0.75
Mn content of pig iron (%)	0.70	0.70	0.70
Mn content of molten steel before tapping (%)	0.04	0.04	0.05
Oxidation amount of Mn m _{7c} (kg/t-output)	7.91	7.91	7.81

	1000		100
Heat of Mn, q _{2cMn} (kcal/kg)	1674	1674	1674
(7d) Oxidation heat of charged P, Q7d (1000 x keal/t-output)	3.2	3.2	3.3
P content of scrap (%)	0.075	0.075	0.075
P content of pig iron (%)	0.040	0.040	0.040
P content of molten steel before tapping (%)	0.026	0.026	0.025
Oxidation amount of P, m74 (kg/t-output)	0.56	0.56	0.57
Heat of P, q _{7dP} (kcal/kg)	5811	5811	5811
(7e) Oxidation heat of charged Cr, Qze (1000 x kcal/t-output)	2.6	2.4	2.6
Cr content of scrap (%)	0.150	0.150	0.150
Cr content of pig iron (%)	0.000	0.000	0.000
Cr content of molten steel before tapping (%)	0.060	0.070	0.060
Oxidation amount of Cr, m _{7c} (kg/t-output)	0.99	0.90	1.00
Heat of Cr, q _{hCr} (kcal/kg)	2620	2620	2620
(7f) Oxidation heat of charged Al, Q71 (1000 x kcal/t-output)	1.3	1.3	1.3
Al content of scrap (%)	0.017	0.017	0.017
Al content of pig iron (%)	0.000	0.000	0.000
Al content of molten steel before tapping (%)	- '		•
Oxidation amount of Al, m _H (kg/t-output)	0.18	0.18	0.18
Heal of Al, q _{7fAl} (kcal/kg)	7419	7419	7419
(7g) Oxidation heat of charged Fe, Q7g (1000 x kcal/t-output)	31.4	29.6	36.3
Unit weight of slag, m _{7g} (kg/t-output)	89	84	103
Heat of Fe oxidation at FeO formation, 978FeO (kcal/kg)	1151	1151	1151
Heat of Fe oxidation at Fe ₂ O ₃ formation, q _{7gFe2O3} (kcal/kg)	1756	1756	1756
FeO content in slag, (FeO) (%)	9.01	9.01	9.01
Fe ₂ O ₃ content in slag (Fe ₂ O ₃) (%)	22.16	22.16	22.16
(8) Oxidation heat of carbon injection, Q8 (1000 x kcal/t-	56.4	43.4	62.0
output)			
Percentage of oxidation heat of carbon injection (%)	9.3	7.3	9.5
Unit consumption of carbon injection (kg/t-output)	11.4	7.6	10.4
C content in carbon injection (%)	80.0	80.0	80.0
Oxidation amount of carbon injection, m _{8a} (kg/t-output)	91	6.1	8.3
(9) Heat of slag formation, Q9 (1000 x kcal/t-output)	8.2	7.7	9,5
Percentage of heat of slag formation in heat input (%)	1.3	1.3	1.5
Heat of SiO ₂ reaction at Ca ₂ SiO ₄ formation, q _{98iO2} (kcal/kg)	502	502	502
Heat of P2O ₅ reaction at Ca ₃ P ₂ O ₃ formation, q _{9P2O5} (kcal/kg)	1070	1070	1070
SiO ₂ content in slag (%)	17.31	17.31	17.31
P2Os content in slag (%)	0.48	0.48	0.48
Heat Input,	608.3	596.7	650.6
Qingut (1000 x kcal/t-output)		<u> </u>	

(5) Heat Output

The Heat outputs for the three (3) measured heats are calculated as shown in Table 12-19. As for calculation formula, refer to Annex-4.

Date		Aug. 14	
Heat No.	965751	965752	
(10) Potential heat of molten steel, Q10 (1000 x keal/t-output)	384.9	391.4	380.5
Ratio of potential heat of molten steel (%)	63.3	65.7	58.5
(10a) Potential heat of output, q10a (1000 x kcal/t-output)	340,6	347.3	337.0
Temperature of molten steel before tapping (°C)	1641	1673	1624
Heat content of molten steel before tapping, h _{10a} (kcal/kg)	340.6	347.3	337.0
(10b) Potential heat of hot heel, q10b (1000 x kcal/t-output)	44.3	44.1	43.5
Unit weight of hot heel of molten steel, m2 (kg/t-output)	130	127	129
(11) Potential heat of slag, Q11 (1000 x kcal/t-output)	43.1	42.3	48.8
Ratio of potential heat of slag in output (%)	7.1	7.1	7.5
Unit weight of slag, m ₁₁ (kg/t-output)	89	84	103
Temperature of slag (°C)	. 1641	1673	1624
Heat content of slag, h _H (kcal/kg)	484.2	503.4	474.1
(12) Heat of limestone decomposition, Q12 (1000 kcal/t-output)	2.9	2.9	3.3
Ratio of heat of limestone decomposition (%)	0.5	0.5	0.5
Unit consumption of limestone, m ₁₂ (kg/t-output)	7	7	8
Heat of decomposition of limestone, q _{12CaCO3} (kcal/kg)	757	757	757
CaO content in limestone (%)		55	55
(14) Heat in cooling water, Q ₁₄ (1000 x kcal/t-output)	41.9	54.0	40.9
Ratio of heat in cooling water (%)	6.9	9.1	6.3
(14a) Heat in cooling water for clbow, Q14a (1000 x kcal/t-	10,3	11.8	11,3
output)			
Average quantity of cooling water for elbow, m143 (kg/t-output)	1096	1071	1436
Mean outlet temperature of cooling water for elbow (°C)	43.4	45.0	42.9
Inlet temperature of cooling water for elbow (°C)	34.0	34.0	35.0
(14b) Heat in cooling for roof, Q14b (1000 x kcal/t-output)	15,1	19.2	16.3
Average quantity of cooling water for roof, m _{14b} (kg/t-output)	3026	2957	3964
Mean outlet temperature of cooling water for roof (°C)	39.0	40.5	39.1
Inlet cooling water for roof (°C)	34.0	34.0	35.0
(14c) Heat in cooling water for EBT, Q _{14c} (1000 x kcal/t-	1.7	2.2	1.2
output)			1
Average quantity of cooling water for EBT, m _{14c} (kg/t-output)	590		
Mean outlet temperature of cooling water for EBT (°C)	36.8		
Inlet temperature of cooling water for EBT (°C)	34.0	· · · · · · · · · · · · · · · · · · ·	
(14d) Heat in cooling water for shell-1, Q14d (1000 x kcal/t-	7.7	10.7	6.1
output)			222
Average quantity of cooling water for shell-1, m ₁₄₁ (kg/t-output)	2557		
Mean outlet temperature of cooling water for shell-1 (°C)	37.0		and the second second
Inlet temperature of cooling water for shell-1 (°C)	34.0		
(14e) Heat in cooling water for shell-2, Q _{14e} (1000 x kcal/t-	7.2	10.0	6.1
output)	02/0	0500	2262
Average quantity of cooling water for shell-2, m _{t40} (kg/t-output)	2567		
Mean outlet temperature of cooling water for shell-2 (°C)	36.8	38.0	36.8

(6) Heat Balance

Heat balance sheets are shown in Figures 12-30, 12-31 and 12-32.

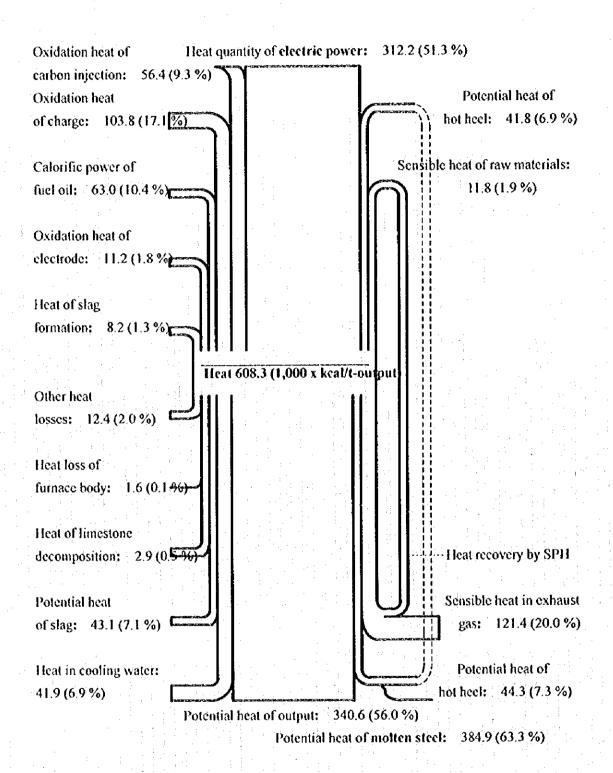


Figure 12-30 Heat Balance of Heat No. 965751

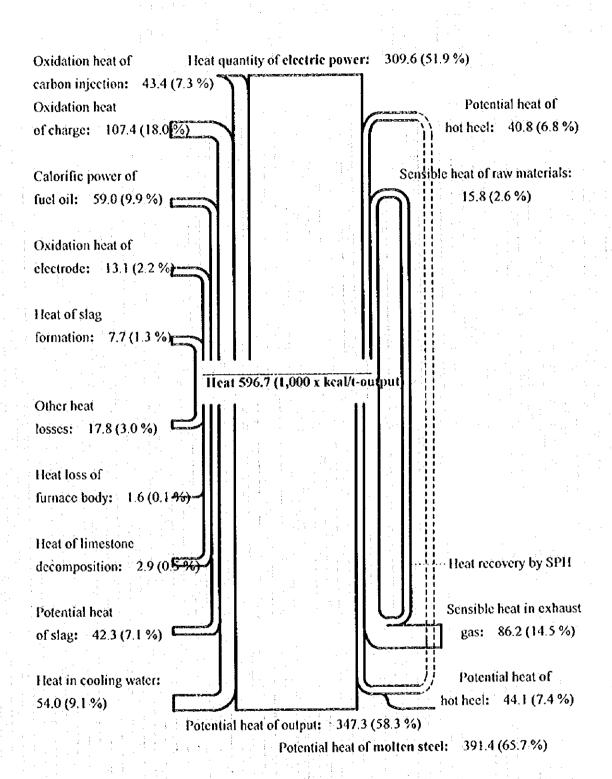


Figure 12-31 Heat Balance of Heat No. 965752

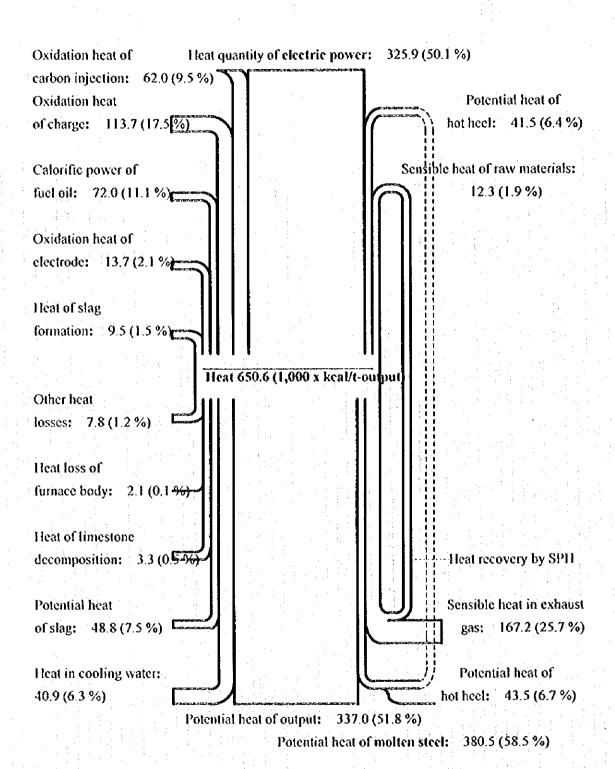


Figure 12-32 Heat Balance of Heat No. 965753

12-9-4 Analysis

(1) General Analysis

1) Observed Characteristics of Heat Balance

Each heat balance sheef shows the following characteristics.

- 1. "Other heat losses" shows almost the same values for all heats, 2.0 % for Heat No. 965751, 3.0 % for Heat No. 965752 and 1.2 % for No. 965753.
- 2. Production of one ton required 650,600 kcal for Heat No. 965753, which is approximately 7 to 8 percent higher than 608,300 kcal for Heat No. 965751 and 596,200 kcal for Heat No. 965752.

In this measurement the electrical heat loss was not measured. The electrical heat loss is usually approximately 2 percent. Therefore, for Item 1 above, "other heat losses" of 1.2 to 3.0 percent is very close to the practical minimum value and also corresponds to the accepted level of electrical heat loss. This may be interpreted to mean that the selected three heats were successfully measured. Incidentally, for other heats, Heat Nos. 965729, 965773, 965776 and 965777 for example, "other heat losses" shows 6.6, 4.9, 11.4, and minus 0.2 percent, respectively. For Item 2 above, Heat No. 965753 required a longer tap-to-tap time than other heats; naturally, heat consumption increased.

2) Comparison with Other Furnaces

Table 12-19 compares the heat balances of IDC with certain Japanese cases. The following may be noted from this table.

- 1. IDC's heat quantity of electric power. Not different for the Japanese cases
- 2. IDC's potential heat of hot heel: No comment
- 3. IDC's sensible heat of raw materials: Lower than the Japanese cases. This could be attributable to operation of the SPH of which improvement is required in IDC.
- 4. IDC's calorific power of fuel oil: Higher. This shows a better performance of IDC from the standpoint of heat balance.
- 5. IDC's oxidation heat of electrode: Lower. This shows inferior performance from the standpoint of heat balance, although the operation is better.
- 6. IDC's oxidation heat of charge: Lower. Oxidation heat of charge depends on the types of the raw materials.
- 7. IDC's oxidation heat of additives: Lower. Furnaces in Japan usually use lump

cokes in the first bucket which IDC does not use.

- 8. IDC's heat of slag formation: No comment.
- 9. IDC's potential heat of molten steel: Higher. This is indicative of better operation.
- 10. HDC's potential heat of hot heel: No comment.
- 11. IDC's potential heat of slag: Higher. Higher potential heat of slag is undesirable from the standpoint of heat balance.
- 12. IDC's heat of limestone decomposition: No comment.
- 13. IDC's heat in cooling water: Lower, Lower ration means that the heat balance is better.
- 14. IDC's sensible heat in exhaust gas: Higher. Higher sensible heat is worse from the standpoint of heat balance.
- 15. IDC's heat loss at furnace body: No comment.
- 16. IDC's other heat loss: No comment.

Table 12-19 Comparison of Heat Input and Output

	Trace VI	Leat Mo	Light V.			
Fumace	965751	965752	965753	125 t Fce	80 tF'ce	50 t F'ce
items ***	Ratio (%)	Ratio (%)	Ratio (%)	Ratio (%)	Ratio (%)	Ratio (%)
1. Heat Input						:
(1) Heat quantity of electric power	51.3	51.9	50.1	50.4	47	54.9
(2) Potential heat of hot heel	6.9	6.8	6.4	8.0		*
(3) Sensible heat of raw materials	5.1	2.6	1.9	4.9		2.9
(4) Calorific power of fuel oil	10.4	6.6	11.1	0.5	**************************************	4.7
(5) Oxidation heat of electrode	1.8	2.2	2.1	2.8	પ	4.3
(6) Oxidation heat of charge	17.1	18.0	17.5	21.0	20	9.5
(7) Oxidation heat of additives	9.3	7.3	5.6	12.4	20	23.7
(8) Heat of stag formation	61 61 61 61	1.3	1.5	0.4		
(9) Heat input, total	100.0	100.0	100.0	100.0	100	100.0
2. Heat Output						
(1) Potential heat of molten steel	26.0	58.3	51.8	51.2	51	51.5
(2) Potential heat of hot heel	7.3	7.4	6.7	8.0		
(5) Potential heat of slag	7.1	7.1	7.5	4.9	v	4.4
(4) Heat of limestone decomposition	0.5	0.5	0.5	•	•	

(5) Heat in cooling water	6.9	1.6	6.3	7.6	15.0	20.1
(6) Sensible heat in exhaust gas	20.0	14.5	25.7	16,4	15.0	•
(7) Heat loss at furnace body	0.1	0.1	0.1	•	t .	•
(8) Other heat loss	2.0	3.0	1.2	11.9	14.0	24.0
(9) Heat output, total	100.0	100.0	100.0	100.0	100.0	100.0

Note: 1) Oxidation heat of additives means that of carbon charged and injected.

2) In case of IDC's heat, electrical heat loss is not measured.

3) In case of 125 t Fice, heat loss at fumace body and electrical heat loss are not measured.

4) In case of 80 t Fice, heat of slag formation is not measured.

5) In case of 80 t Fice, oxidation heat of charge includes the calorific power of oil adhered to scrap of 8%.

6) In case of 80 t Fee, other heat loss is of heat loss at furnace body and electrical heat loss.

7) In case of 50 tFce, heat of slag formation is not measured.

8) In case of 50 tFice, sensible heat in exhaust gas is included in other heat loss.

(2) Heat Quantity of Electric Power

It may be noted from Table 12-19 that the heat quantity of electric power of about 51 percent is not different from those of the selected furnaces of Japan.

(3) Potential Heat of Hot Heel

3

The Potential heat of hot heel is about 7 percent in the heat input on the assumption that the temperature of hot heel is 1,550 °C. From the standpoint of heat balance, if its temperature is higher by 10 °C, energy saving of 2,100 kcal/t (2.4 kWh/t) will be realized. Therefore, power-on of the subsequent heat should start immediately to prevent the temperature drop of hot heel.

(4) Sensible Heat of Raw Materials - Effectiveness of SPII

In this measurement, temperature of raw materials is represented by bucket temperature. Raw materials are charged to the EAF by three buckets, of which the first and second buckets are preheated in SPH. Figure 12-33 shows the relationship between bucket temperature and the residence time in SPH. With the exception of one measurement, longer residence times in SPH indicate a possibility to raise the bucket temperature higher. The residence time in SPH should be as longer as possible within the tap-to-tap time limit of EAF.

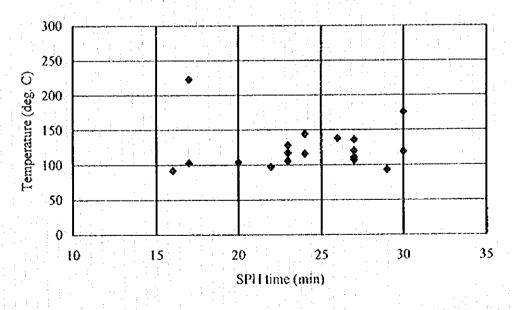


Figure 12-33 Relationship between Bucket Temperature and SPH Time

The heat balance indicates that if the third bucket could also be preheated by modification of the existing SPH, electric power of 6.8 kWh/t and production cost of 17,900 USD/M will be saved.

At present:

1st bucket 42 t x 150 °C after 30 min. preheating

2nd bucket 28 t x 100 °C after 15 min. preheating

3rd bucket 16 t x 35 °C at indoor temperature

Average

112.3 °C

Heat input = 12.88 kcal (heat of scrap at 112.3 °C for 1 kg)

x 1111 kg (unit consumption of raw materials)

/860 kcal (per kWh) = 16.6 kWh/t

Improvement: 1st bucket 42 t x 150 °C after 30 min. preheating

2nd bucket 28 t x 180 °C after 40 min. preheating

3rd bucket 16 t x 130 °C after 25 min. preheating

Average

156.0 °C

Heat input = 18.14 kcal (heat of scrap at 156 °C for 1 kg)

x 1111 kg (unit consumption of raw materials)

/860 kcal (per kWh) = 23.4 kWh/t

Saving of electric power: 23.4 - 16.6 = 6.8 kWh/t

Saving of production cost: 6.8 kWh/t x 0.044 USD/kWh = 0.299 USD/t

= 17,900 USD/M at production of 60,000 t/M.

Schematic diagrams of SPH operation and bucket charge for existing and after modification are shown in Figure 12-34 and Figure 12-35, respectively.

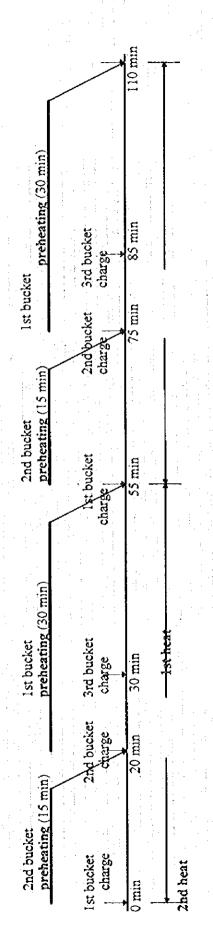


Figure 12-34 Schematic Diagram of SPH Operation and Bucket Charge - Existing -

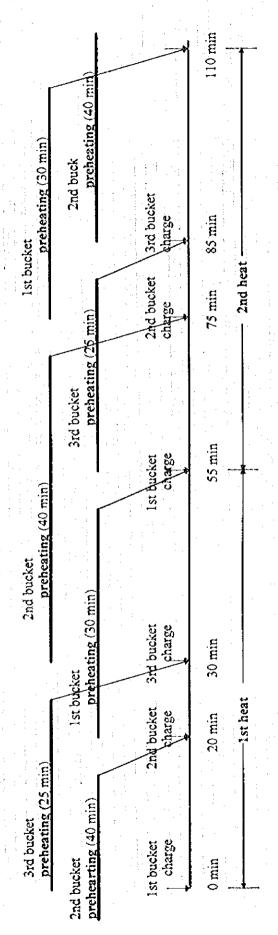


Figure 12-35 Schematic Diagram of SPH Operation and Bucket Charge - After Modification -

(5) Calorific Power of Fuel - Effectiveness of Oxy-Fuel Burners

= 5.2 kWh/t-output

The ratio of calorific power of fuel oil to the total heat input is approximately 10 percent. The charged material, scrap and pig iron, shall be melted of as fast as possible by energy of electric power with the help of the oxy-fuel burners and oxy-lancing method.

The ratio of O₂/Oil (Nm³/kg) of oxy-fuel burners varies from one heat to another and with the burner as shown in Table 12-20. The ratio of O₂/Oil occasionally varies accidentally due to clogging of the burner tip. Clogging must be prevented by daily maintenance and the burner designs must be improved to prevent clogging.

Table 12-20 shows that the oxy-fuel burners were used for 24 minutes and the flow rate of fuel oil was 293 kg/hr. If the flow rate of fuel oil is increased to 380 kg/hr and a constant ratio of O₂/Oil is maintained by improvement of this system, electric power consumption of 5.2 kWh/t-output, which corresponds to the production cost of 9,700 USD/M, will be saved.

Saving of electric power: (380 - 293) kg/hr x 24 min./60 min. = 35 kg/heat = 0.45 kg/t-output (x 10,000 kcal/kg) = 4,500 kcal/t-output (/860 kcal/kWh)

Saving of production cost: 5.2 kWh/t x 0.044 USD/kWh - 0.45 kg/t x 0.15 USD/kg = 0.161 USD/t = 9,700 USD/M at production of 60,000 t/M

Table 12-20 Flow Rate of Oxygen Gas and Fuel Oil for Oxy-Fuel Burner

Heat No		965749	965750	157596	965752	965753	965773	965774	965775	96 977 896	965777	Average
No. 1	Duration (min)	24	25	20	26	23	24	25	21	24		23.6
Burner	O ₂ (Nm ³)	280	290	232	365	249	252	273	236	268		272
-	Oil (kg)	121	105	133	104	182	94	121	95	107		118
	O ₂ (Nm ³ /nr)	700	969	969	842	650	630	655	674	029		690
	Oil (kg/hr)	303	252	399	240	475	235	290	271	268		304
	O ₂ /Oil (Nm³/kg)	2.31	2.76	174	3.51	1.37	2.68	2.26	2.48	2.50		2.40
No.2	Duration (min)	24	25	25	26	23	24	25	21	24		24.1
Burner	O. (Na.)	248	283	285	301	255	278	293	257	278	t.	275
	Oil (kg)	132	158	162	107	136	101	151	169	164		142
	O. (Nm3/hr)	620	619	684	695	665	695	703	734	569		989
	Oil (kg/hr)	330	379	389	247	355	253	362	483	410		356
	O ₂ /Oil (Nm³/kg)	1.88	1.79	1.76	2.81	1.88	2.75	1.94	1.52	1.70		2.00
No.3	Duration (min)	24	25	112	26	23	24	25	21	24		22.7
Burner	O. (Nm³)	265	286	152	307	249	264	282	250	268	1 1+	258
:	Oii (kg)	8	100	56	101	93	82	95	85	87		88
	O ₂ (Nm³/hr)	663	989	760	708	650	099	677	714	670		688
: :	Oil (kg/nr)	240	240	280	233	243	205	228	275	218	i i	240
	O ₂ /Oil (Nm³/kg)	2.76	2.86	2.71	3.04	2.68	3.22	2.97	2.94	3.08		2.92
No.4	Duration (min)	24	25	25	26	23	24	25	21	24	•	24.1
Burner (O. (Nm³)	288	307	297	308	272	332	330	275	297		301
	Oil (kg)	106	113	112	117	124	102	116	83	105		110
	O ₂ (Nm ² /hr)	720	737	713	711	710	830	792	786	743		749
	Oil (kg/hr)	265	27.1	269	270	323	255	278	266	592		273
	O ₂ /Oil (Nm³/kg)	2.72	2.72	2.65	2.63	2.19	3.25	2.84	2.96	2.83		2.76

(6) Oxidation Heat of Electrodes

The oxidation heat of electrodes is approximately 2 percent on the heat input. It is lower than that of other furnaces in Japan. Lower oxidation heat of electrodes is worse from the standpoint of heat balance but is better in operation results.

(7) Oxidation Heat of Charge

The oxidation heat of charge is approximately 18 percent, second only to the 51 percent of electrical energy in heat input. This figure varies depending on the kind of raw materials. Table 12-21 compares oxidation heat of pig iron and scrap. Calculation was done assuming the conditions to be the same as those of the measurement. Oxidation heat of pig iron is higher by 298,200 kcal/t than that of scrap. This means that pig iron has an additional value of approximately 14 USD/t compared with scrap in the heat balance. This will be a factor for judging which raw material should be purchased.

(373,400 - 98,100) kcal/860 kcal (per kWh) x 0.044 USD (per kWh) = 14.09 USD/t

Table 12-21 Comparison of Oxidation Heat of Pig iron and Scrap

interior de la companya de la compa	and the second second	
Kind of raw materials	pig iron	scrap
Mean CO ₂ content in exhaust gas (CO ₂) (%)	14.7	14.7
Mean CO content in exhaust gas (CO) (%)	4.7	4.7
(7) Oxidation heat of charge, Q ₁ (1000 x kcal/t-output)	373.4	98.1
(7a) Oxidation heat of charged C, Q _{7i} (1000 x kcal/t-output)	245.8	23.6
Unit consumption of scrap (kg/t-output)	0	1000
Unit consumption of pig iron (kg/t-output)	1000	0
C content(%)	3.65	0.35
C content of molten steel before tapping (%)	0.00	0.00
Oxidation amount of charged C, m _{7a} (kg/t-output)	36.5	3.5
Heat of C oxidation at CO ₂ formation, q _{2CCO2} (keal/kg)	8075	8075
Heat of C oxidation at CO formation, q _{7CCO} (kcal/kg)	2448	2448
(7b) Oxidation heat of charged Si, Q _{7b} (1000 x keal/t-output)	78.3	17.2
Si content	1.05	0.23
Si content of molten steel before tapping (%)	0.00	0.00
Oxidation amount of charged Si, m ₇₆ (kg/t-output)	10.50	2.30
Heat of Si, q _{nss} (kcal/kg)	7459	7459
(7c) Oxidation heat of charged Mn, Q _{rc} (1000 x kcal/t-output)	11.7	12.6
Mn content(%)	0.70	0.75
Mn content of molten steel before tapping (%)	0.00	0.00
Oxidation amount of Mn, m _{7c} (kg/t-output)	7.00	7.50
Heat of Mn, q _{1cMn} (kcal/kg)	1674	1674
(7d) Oxidation heat of charged P, Q2d (1000 x kcal/t-output)	2.3	4.4

		:
P content (%)	0.040	0.075
P content of molten steel before tapping (%)	0.000	0.000
Oxidation amount of P, m ₇₄ (kg/t-output)	0.40	0.75
Heat of P, q_{7dP} (kcal/kg)	5811	5811
(7e) Oxidation heat of charged Cr, Q _{1e} (1000 x kcal/t-output)	0.0	3.9
Cr content of scrap (%)	0.000	0.150
Cr content of molten steel before tapping (%)	0.000	0,000
Oxidation amount of Cr, m _{7e} (kg/t-output)	0.00	1.50
Heat of Cr, q _{xcr} (kcal/kg)	2620	2620
(7f) Oxidation heat of charged Al, Qn (1000 x kcal/t-output)	0.0	1.3
Al content (%)	0.000	0.017
Al content of molten steel before tapping (%)	0.0	0.0
Oxidation amount of Al, m ₇₁ (kg/t-output)	0.00	0.17
Heat of Al, q _{7/Al} (kcal/kg)	7419	7419
(7g) Oxidation heat of charged Fe, Q _{Ig} (1000 x kcal/t-output)	35,3	35.3
Unit weight of slag, m _{7g} (kg/t-output)	100	100
Heat of Fe oxidation at FeO formation, q _{1gFeO} (kcal/kg)	1151	1151
Heat of Fe exidation at Fe ₂ O ₃ formation, q _{1gFe2O3} (kcal/kg)	1756	1756
FeO content in slag, (FeO) (%)	9.01	9.01
Fe ₂ O ₃ content in slag (Fe ₂ O ₃) (%)	22.16	22.16

(8) Oxidation Heat of Carbon Injection

Carbon injection is originally one of the operation methods to generate foamy slag in order to save electrical energy surrounding the arc by foamy slag. But in heat balance, it is considered one of the energy sources. In the case of IDC the oxidation heat is approximately 8 percent of the heat input and is lower than that of furnaces in Japan, because furnaces in Japan usually use lump cokes in the first bucket in addition to carbon injection.

(9) Iteat of Slag Formation

Table 12-22 gives consumption of burnt lime, slag generation and their heat balance recorded in Heat No. 965752 and 965753 as an example of heat of slag formation.

Table 12-22 Consumption of Burnt Lime, Slag Generation and Heat Balance

Heat No.	965752	965753
Consumption of burnt lime (kg/heat)	1790	2150
Unit consumption of burnt lime (kg/t-output)	22.8	27.8
Slag generation (kg/heat)	66t0	7939
Unit weight of slag (kg/t-output)	84	103
Heat of slag formation (1,000 kcal/t-output), Heat Input	7.7	9,5

The potential heat of slag of Heat No. 965753 is higher by 6,500 kcal/t-output than that of Heat No. 965752 while the heat of slag formation of the former is higher only by 1,800 kcal/t-output than the latter. This suggests excessive charging of burnt lime; therefore, the consumption of burnt lime should be restudied taking the heat balance and also content of phosphorus and sulfur in the molten steel into consideration.

If consumption of burnt lime would be decreased to 25.6 kg/t-output (2,000 kg/heat) from 38.4 kg/t-output (3,000 kg/heat), the data collected during the first field survey, Table 12-23 shows the expected improvements in lime consumption, slag generation and heat balance resulting from the decrease in the use of burnt lime.

Table 12-23 Expected Improvements from Reduction of Burnt Lime Use

	Jan Oct.	Proposed	Improve-
	1995	target	ment
Consumption of burnt lime (kg/heat)	3,000	2,000	+ 1,000
Unit consumption of burnt lime (kg/t-output)	38.4	25.6	+ 12.8
Slag generation (t/heat)	11.078	7.385	+ 3.693
Unit weight of slag (kg/t-output)	142	95	+ 47
Heat of slag formation (1,000 kcal/t-output), Heat Input	13.1	8.7	-4.4
Potential heat of slag (1,000 kcal/t-output), Heat Output	67.0	44.8	1 22.2

Accordingly, 17,800 kcal/t-output (20.7 kWh/t-output) will be reduced in heat balance and unit consumption of burnt lime will also be reduced by 12.8 kg/t-output.

Production cost saving: Burnt time 12.8 kg/t-MS x 0.043 USD/kg = 0.55 USD/t-MS 20.7 kWh/t/MS x 0.044 USD/kg = 0.91 USD/t-MS

Total: 1.46 USD/t-MS

= 87,000 USD/M at production of 60,000 VM.

(10) Potential Heat of Molten Steel

The potential heat of molten steel, excluding that of hot heel, is 56.0, 58.3 and 51.8 percent of the heat output. These percentages are higher than the approximate Japanese value of 51 percent and may be regarded as an indication of IDC's good performance.

(11) Potential Heat of Slag

See sub-section 12-9-4 (9) above.

(12) Heat of Limestone Decomposition

The quality of burnt lime used by IDC is not always good. It contains approximately 30 percent of uncalcined limestone. If burnt lime of good quality could be purchased at the same price as the present burnt lime, 3.4 kWh/t will be saved from the standpoint of heat balance.

2.900 kcal/t (/860 kcal/kWh) = 3.4 kWh/t at the unit consumption of burnt lime of 23.3 kg/t

(13) Heat in Cooling Water

Heat carried out by cooling water was 6.9, 9.1 and 6.3 percent on the heat output, rather large heat losses. As shown in Table 12-24, average temperature difference between the inlet and outlet was 9.7 °C for the elbow, 5.0 °C for the roof, 2.6 °C for EBT, 2.3 °C shell-1 and 2.5 °C shell-2. These temperature differences were very small. Mean maximum outlet temperatures were 49.8 °C for elbow, 42.1 °C for roof, 38.7 °C for EBT, 38.3 °C for shell-1 and 38.1 °C for shell-2. If the mean maximum outlet temperature could be raised to 58 °C, the design temperature rise of 23 °C higher than the inlet temperature of 35 °C in tubular cooling system, the quantity of cooling water would be decreased by about 7.7 t/t-output (800 m³/hr) and 6,500 USD/M would be saved from the standpoint of heat balance.

(10,159 - 2,442) kg/t-output x 0.014 USD/t = 0.108 USD/t-output = 6,500 USD/M at 60,000 t/M of production

 $(1.053 - 254) \text{ m}^3/\text{hr} = 799 \text{ m}^3/\text{hr}$

The present practice of muffling the arc by slag, thereby preventing water cooling panels from being exposed directly to the arc, is a very good one. In case IDC decreases the amount of cooling water, it should be done step by step each time by a small amount and IDC should see the results before further decreasing cooling water.

Table 12-24 Improvement of Cooling Water for Furnace

					:							
Date	Aug.	Aug.	Aug.	Aug.	Aug.	Aug.	Aug.	Aug.	Aug.	Aug. Aver-		Im- prove-
Heat No.	5749	5750	5751	5752	5753	5773	5774	5775	5776	5777		ment
(14) Heat in cooling water, Q14 (1000 x kcal/t-output)	42.2	44.5	41.9	54.0	40.9	44.3	39.4	40.8	26.7	25.7	40.6	9.04
Quantity of cooling water (kg/t-output)	5066	10576	9836	9612	12898	10113	9848	8439	8993	11373	10159	2442
Output (t/heat)	77.130 77.400		76.860	78.660	77.400	77.238	76.914	76.824	75.960	76.626	77.101	77.101
Power on-to-power off time (hr/heat)	0.72	0.77	0.72	0.72	0.95	0.73	0.72	0.63	0.65	0.83	0.74	0.74
Flow rate of cooling water (m³/far)	1061	1063	1050	1050	1051	1070	1052	1029	1051	1050	1053	254
(14a) Heat in cooling water for elbow, Q;4a (1000 x kcal/t-output)	11.8	11.3	10.3	11.8	11.3	11.3	12.5	10.5	8.8	8.0	10.9	10.9
Quantity of cooling water for elbow, m _{14a} (kg/t-output)	1092	1164	9601	1071	1436	1106	1095	956	1001	1267	1129	637
Maximum outlet temperature of cooling water for elbow $(^{\circ}C)$	50.3	48.1	6.65	51.3	\$2.1	48.2	51.1	50.1	49.3	47.5	49.8	28.0
Mean outlet temperature of cooling water for elbow (°C)	44.8	43.7	43.4	45.0	45.9	42.2	45.4	45.0	44.8	41.3	43.9	52.1
Inlet temperature of cooling water for elbow $({}^{\circ}C)$	34.0	34.0	34.0	34.0	35.0	32.0	34.0	34.0	36.0	35.0	34.2	35.0
(14b) Heat in cooling for roof, Q14b (1000 x kcal/t-output)	15.2	16.7	15.1	19.2	16.2	17.9	15.1	14.8	11.9	11.9	15.6	15.6
Quantity of cooling water for roof, m ₁₄₈ (kg/t-output)	3109	3343	3026	2957	3963	3138	3024	2469	2764	3499	3129	776
Maximum outlet temperature of cooling water for roof $({}^{\circ}C)$	41.3	41.6	42.3	43.3	41.9	40.7	42.1	42.3	43.6	42.2	42.1	58.0
Mean outlet temperature of cooling water for roof (°C)	38.9	39.0	39.0	40.5	39.1	37.7	39.0	0.0	40.3	38.4	39.2	55.1

		: - (0	000	076	, C	3,40	35.0	6	35.0
Inlet cooling water for fool $({}^{\circ}C)$	0.46	0.40	5 5) 1	0.00	0.70	74.0	7.0	20.0	3.00	i r	S
(14c) Heat in cooling water for EBT, Qiac (1000 x kcal/t-output)	1.6	8	1.7	5. Ci	1.2	1.7	8.	8 .	8.0	8.0	1.5	1.5
Average quantity of cooling water for EBT m _{14c} (kg/t-output)	588	627	290	577	773	595	290	517	239	682	809	71
Maximum outlet temperature of cooling water for EBT (°C)	38.2	40.1	39.2	39.3	37.5	36.4	40.1	38.6	39.4	38.1	38.7	58.0
Mean outlet temperature of cooling water for FBT (°C)	36.7	36.8	36.8	37.8	36.5	34.8	37.0	37.4	37.5	36.2	36.8	56.1
Inlet temperature of cooling water for EBT (°C)	34.0	34.0	34.0	34.0	35.0	32.0	34.0	34.0	36.0	35.0	34.2	35.0
(14d) Heat in cooling water for shell-1, Old (1000 x kcal/t-output)	6.4	8.9	7.7	10.7	6.1	6.4	5.1	6.3	1 6	1.2	9	0.9
Quantity of cooling water for shell-1, mise (kg/t-output)	2558	2716	2557	2499	3363	2665	2565	2247	2336	2957	2646	652
Maximum outlet temperature of cooling water for Shell-1. (°C)	37.8	37.7	41.7	40.6	38.7	35.7	38.2	38.0	37.8	36.8	38.3	58.0
Mean outlet temperature of cooling water for shell-1 (°C)	36.5	36.5	37.0	38.3	36.8	34.4	36.0	36.8	36.8	35.4	36.5	44.2
Inlet temperature of cooling water for shell-1 $(^{\circ}C)$	34.0	34.0	34.0	34.0	35.0	32.0	34.0	340	36.0	35.0	34.2	35.0
(14e) Heat in cooling water for shell-2, Q _{14e} (1000 x kcal/t-output)	7.2	6.2	7.2	10.0	6.1	7.0	6.9	7.4	3.3	3.9	9.9	9.9
Quantity of cooling water for shell-2, m _{3.44} (kg/t-output).	2558	2726	2567	2508	3363	2609	2574	2247	2353	2968	2647	306
Maximum outlet temperature of cooling water for shell-2 $(^{\circ}C)$	38.5	37.9	38.4	39.1	37.9	36.2	37.7	38.6	38.5	37.8	38.1	58.0
Mean outlet temperature of cooling water for shell-2 (°C)	36.8	36.9	36.8	38.0	36.8	34.7	35.9	37.3	37.4	36.3	36.7	56.6
Inlet temperature of cooling water for shell-2 $({}^{\circ}C)$	34.0	34.0	34.0	34.0	35.0	32.0	34.0	34.0	36.0	35.0	34.2	35.0

(14) Sensible Heat in Exhaust Gas

The sensible heat of the exhaust gas was 20.0 %, 14.5 % and 25.7 % of the heat output. These heat losses are slightly higher than those of the furnaces in Japan.

According to Table 12-17, CO gas remains in the exhaust gas at 1.5 to 7.1 percent. The CO gas is now burned out in the combustion chamber in which the energy of CO gas is transferred to the cooling water. If the residual CO gas could be burned in the EAF by supplying more oxygen gas and its heat of combustion could be transferred to the molten steel, better energy efficiency would be achieved. This application is called post combustion technology.

The obtainable heat of combustion of CO gas is calculated from the following equations.

$$C + O_2 = CO_2 + 33,810 \text{ kj/kg-C}$$

$$C + 1/2 O_2 = CO + 10,250 \text{ kg/kg-C}$$

Therefore,

$$CO + 1/2 O_2 = CO_2 + 23,560 \text{ kj/kg-C}$$

The heat of 23,560 kj/kg-C is arithmetically converted into 3,015 kcal/Nm3-CO.

Based on the measured data in IDC, the heat contents of residual CO gas for each heat number are calculated.

Heat No.		F _{IN} (Nm³/min)	Time	- 1 T 1		F-CbO ₂	T-CbO2 (Nm3/heat)
	(%)	(isin rinni).	(min)				-
965751	7.1	555	43	111,600	4,800,000	18.5	796
965752	3.5	401	43	40,400	1,730,000	6.7	288
965753	1.5	657	57	29,900	1,700,000	5.0	285
Average	4.0				2,743,000		456

Note:

Cb-Ht_{in}: Calculated combustion heat flow rate (kcal/min)

Cb-Ht_{IN} = $(CO_{IN}/100) \times F_{IN} \times 3,015$

T-Ch-Ht_{IN}: Total calculated combustion heat (kcal/heat)

 $T-Cb-Ht_{IN} = Cb-Ht_{IN} \times Time$

Time: Power-on to power-off time (min)

F-CbO₂: Calculated combustion O₂ gas flow rate to burn CO gas flow rate (Nm²/min)

 $F-CbO_2 = F_{EN} x (CO_{EN}/100) x 1/2$ $T-CbO_2$: Total calculated combustion O_2 gas quantity to burn CO gas flow (Nm³/heat) $T-CbO_2 = F-CbO_2 x$ Time

The calculation results show that the heat of 2,743,000 kcal/heat (41.1 kWh/t) will be generated by burning the CO gas of 4.0 % with consumption of the oxygen gas of 456 Nm³/heat (5.9 Nm³/t). Namely, the 26.6 kWh/t seems to be added to the molten steel considering that the average ratio of potential heat of molten steel is 62.5 percent of heat input (See ratio of potential heat of molten steel in Table 12-19).

All of the 26.6 kWh/t will not be saved for the following reasons.

- 1. There is much difference between heat input mainly consisting of electric power, oxidation heat of charge and hot heel and heat generated by burning of CO gas in the exhaust gas. It is considered that the former is directly added to the molten steel and the latter is indirectly added. In other words, efficiency of heat radiation of the latter is small. Therefore the ratio of potential heat of molten steel will be far smaller than 62.5%.
- 2. Heat generated above the molten steel in the EAF is absorbed not only in the molten steel but also in the shell and the roof. Considering that the heat absolution area of shell and roof is larger than the surface of molten steel, heat to be absorbed in the molten steel will be smaller than that of shell and roof.

The results shown in the above table are calculated on the assumption that all the residual CO gas of 4 percent is burned by the oxygen consumption of 5.9 Nm³/t. However, in actual operation, generation of CO gas varies during heat melt time shown in Figure 12-20. The flow rate of oxygen gas could not be controlled following the generation of CO gas. Therefore, several times as much oxygen as 5.9 Nm³/t will be necessary to burn all CO gas generated.

From the above considerations, the effect of post combustion technology is doubtful. In addition, this technology has not yet been introduced in Japan.

(15) Heat Loss at Furnace Body

The ratio of heat loss at furnace body to the heat output is very small.

(16) Other Heat Loss See sub-section 12-9-4 (1) 1) above.

12-10 Energy Flowchart of Factory and Major Energy Consuming Facilities

The energy flow chart is shown in Figure 12-36.

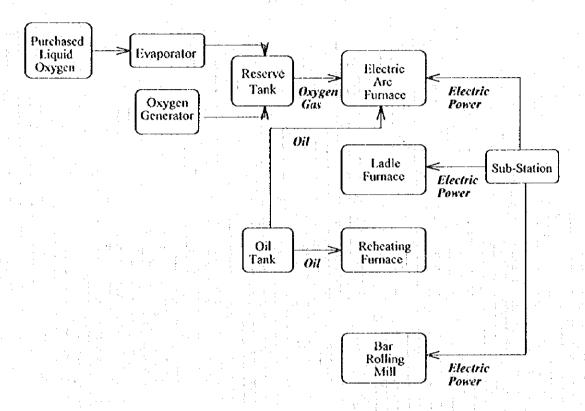


Figure 12-36 Energy Flow in IDC

Major energy consuming facilities are described in sub-section 12-6-1.

12-11 Formulation and Recommendation of Countermeasures for Energy Conservation

12-11-1 Modifications of Facilities and Operation

Based on the measurement and analysis of the heat balances of the electric arc furnace (EAF), the following measures for streamlining of energy use are recommended. Hot temperature charge of billets to the reheating furnace is also recommended based on NKK's experience. Recommendations 1 to 5 are analyzed above in monetary terms and 6 to 8 are analyzed in terms of operation improvement.



Modify the scrap preheaters (SPH) to accommodate a 3rd bucket for preheating.
 The following facilities should be installed to operate SPH's simultaneously and independently:

A new 2nd hood and a new 2nd inlet duct to the existing 2nd chamber

A new 2nd outlet duct from the existing 2nd chamber

A new 2nd damper for a new 2nd outlet duct

Two blowers for the existing 1st and a new 2nd outlet ducts.

- 2. Improve maintenance of oxy-fuel burners of EAF to keep the O2/Oil ratio constant.
- 3. Standardize burnt lime addition into EAF.
- 4. Decrease flow rate of cooling water for EAF.
- 5. Introduce billet cooling system at higher temperature in front of the reheating furnace to prevent crack generation in cast billet.

Pumping system, water piping, spray nozzles and control system should be installed. (Detail concept is shown in the end of this chapter in Annex-3 HOT CHARGE OF BILLET).

- 6. Turn power on as immediately as possible to prevent the drop of hot heel temperature
- 7. Preheat scrap as longer as possible to raise it temperature.
- 8. Purchase as well calcined burnt lime as possible to prevent heat loss in decomposition of limestone.

12-11-2 Enhancement of Morale

To streamline energy use, concerned people at all levels of management, engineers and workers should have their respective roles to play.

(1) Energy Conservation Activity at Management Level

The management should set the ground rules for the activities. The following three items are essential in the promotion of energy conservation throughout the steel works.

- 1. Establishment of policy (top management's stance)
 - (1) Announcement of firm determination for energy saving
 - (2) Setting quantitative target
 - (3) Conditions and criteria (investment limit, ROI/PBP)
 - (4) Others
- 2. Establishment of organization for promotion of energy use streamlining (committee of engineers)

- (1) Investigation of energy consumption and costs
- (2) Planning and following up of the results of activities
- (3) Collection of proposals from staff and employees
- (4) Budgeting
- (5) Education, PR
- (6) Others
- Promotion of small team's activities (encouragement of worker team's motivation and adoption of their ideas)
 - (1) Formation of teams
 - (2) Introduction of study support system
 - (3) Result of study presentation and following up of performance
 - (4) Appraisal and prize system
 - (5) Others

These are important ground rules that the measurement can set and are indispensable to creating an atmosphere for energy saving in the entire steel works.

(2) Energy Conservation of Activity at Engineer Level

The level of activity is no less important and should be the core of energy saving, because only at this level can technical studies be done.

- 1. Decrease heat input.
 - (1) Change of heat sources.

Auxiliary combustion (Oxy-fuel burner)

Oxygen injection

Scrap preheating by fuel

(2) Utilization of waste heat

Scrap preheating

Utilization of hot return scrap

Utilization of recuperator

Hot heal operation

Hot charge of billet

(3) Lowering of molten steel temperature

Increase of ladle temperature

Ladle hood on molten steel

2. Reduction of heat loss

(1) Improvement of facilities

Furnace capacity up

High electric power system

Enhancement of performance of automatic electrode controller

Minimization of secondary conductors and electrodes

Reduction of openings on furnace

(2) Improvement of operation

Prompt analysis of molten steel

Improvement of scrap mixing

Shorter time for scrap charge

Optimization of electric power use

Shorter time of refining and repair

Effective operation of dust removal system

Prevention of electrode breakage

Elimination of waiting time of continuous casting

Prevention of uneven melting of steel

Longer carbon injection time for slag foaming

Others

- 3. Improvement of good quality billet yield
 - (1) Reduction of errors in casting
 - (2) Decrease of residual molten steel
- 4. Utilization of waste heat
 - (1) Utilization of hot waste water (for air conditioning, boiler water preheating)
 - (2) Utilization of exhaust gas (steam, electricity)
 - (3) Utilization of slag sensible heat (steam, electricity)
 - (4) Utilization of molten steel sensible heat (steam, electricity)

(3) Energy Conservation Activity at Workers Level

This level of activity is very important for enhancing the morale of field workers. At the same time this can put into practice their ideas based on practical knowledge and operational experience that concern not only energy saving but safety and environmental conservation.

- 1. Heat energy saving
 - (1) Proposal for small improvements
 Insulation cover on slag door
 Sand gasket of flange of EAF

Others

(2) Standardizing operation
Opening and closing of burners
Reheating of scrap
Opening of slag door
Removing of water from scrap

- 2. Electric energy saving (Rotating machines: pumps, compressor, etc)
 - (1) Change of rotation speed

Change of pulleys

Change of motor poles

Installation of VVVF (Variable voltage variable frequency)

Others

(2) Remodeling of equipment Impeller diameter reduction Replacement of pumps Optimization of pump line-up

Others

(3) Energy recovery, etc.

Suspension of pump operation (Transfer by pressure)

Suspension of blower operation (Use of stack effect)

Hydraulic turbine/expansion turbine

Pressure reduction turbine

Thermo compressor/heat pipe

Others

To perform these activities successfully, the staff, who are in the position to support the workers technically, should prepare the instruction materials and guide them in detail.

An example of these activities is shown in Annex-5.

12-12 Cost Estimation of Countermeasures

12-12-1 Modification of scrap preheater (SPH)

1 2nd hood

- 2. Inlet duct and damper to 2nd chamber
- 3. Outlet duct and damper from 2nd chamber
- 4. Blowers for 1st and 2nd chambers
- 5. Electrical equipment
- 6. Erection

Total budget cost: 200,000 US Dollars

Schematic layout is shown in Figure 12-37.



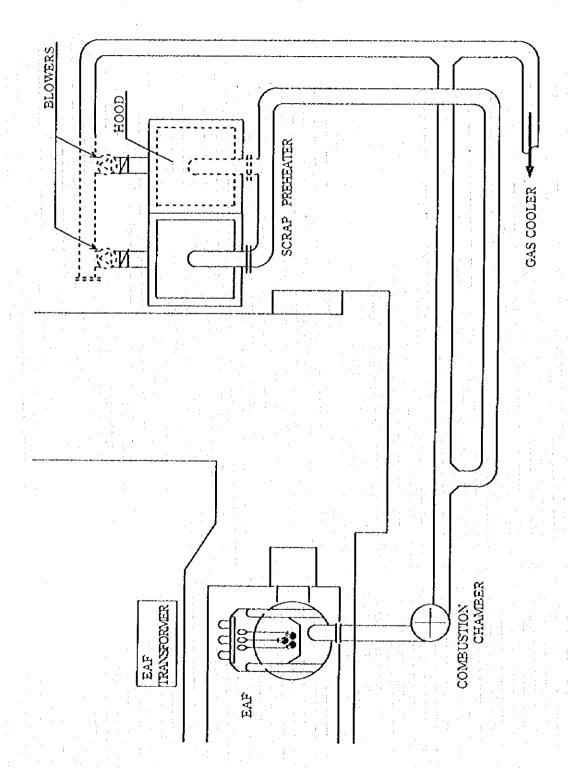


Figure 12-37 Schematic Layout of SPH Modification



12-12-2 Introduction of Billets Cooling System

- 1. Pumps: 2 units for 300 m³/hr of water spray
- 2. Piping, valves and fittings
- 3. Spray nozzles: 144 pieces
- 4. Electrical equipment
- 5. Erection

Total budget cost: 21,300 US Dollars

12-13 Overall Evaluation of Countermeasures for Energy Conservation

Payout periods for the recommended modifications are calculated below.

1. Modification of scrap preheater (SPH) to accommodate preheating the 3rd bucket.

See sub-section 12-9-4 (4).

Saving of electric power: 6.8 kWh/t-MS (Molten Steel).

Saving of production cost: 0.299 USD/t-MS.

= 17,900 USD/ M at production of 60,000 t/M.

Modification budget cost: 200,000 USD.

Conclusion: Full recovery of investment is expected within 12 months.

2. Maintenance of oxy-fuel burners of EAF to keep the constant O2/Oil ratio.

See sub-section 12-9-4 (5).

Saving of electric power: 5.2 kWh/t-output.

Saving of production cost: 0.161 USD/t-MS.

= 9,700 USD/M at production of 60,000 t/M.

3. Standardization of burnt lime addition into EAF.

See sub-section 12-9-4 (9).

Saving of burnt lime: 12.8 kg/t-MS comparing 1st half year of 1995.

Saving of electric power: 20.7 kWh/t-MS.

Saving of production cost: 87,000 USD/M at production of 60,000 t/M.

4. Decreasing flow rate of cooling water for EAF.

See sub-section 12-9-4 (13).

Saving of flow rate of cooling water: 800 m³/hr.
Saving production cost: 6,500 USD/M at production of 60,000 t/M.

- 5. Introduction of billets cooling system in front of reheating furnace (Detail is shown in the end of this chapter titled as HOT CHARGE OF BILLET).
- (1) At present: 60 70 % (65 %) of billet is hot charged at 300 -600 °C (average 450 °C) and heated to 1,100 1,150 °C (1,125 °C).
- (2) Improvement: 60 70 % (65 %) of billet is hot charged at 690 °C and heated to 1,100 1,150 °C (1,125 °C).

Saving of fuel oil: Heat content (100.8 kcal/kg at 690 °C - 58.1 kcal/kg at 450 °C) x Billet weight 1,000 kg/Low heating value of fuel oil 10,000 kcal/kg = oil 4.3 kg/t-BT (billet)/Combustion efficiency 50 % = oil 8.5 kg/t-BT x 65 % = oil 5.6 kg/t-BT

Saving production cost: 5.6 kg/t-BT x 0.15 USD/kg-oil = 0.84 USD/t-BT = 50,400 USD/M at production of 60,000 t/M.

Construction cost: 21,300 USD

Conclusion: Full recovery of investment is expected within one month.

Following recommendations are guidelines for EAF's operation.

- 6. Turn power on immediately to prevent the drop of hot heel temperature.

 If hot heel temperature is higher by 10 °C, electricity of 2.4 kWh/t-MS will be saved.
- 7. Preheat scrap as long as possible to obtain higher temperature.
- 8. Purchase as well calcined burnt lime as possible to prevent heat loss for decomposition of limestone.

12-14 Technical Guidelines for Energy Conservation

Measurement procedure is described for determination of heat balance of the electric arc furnace.

12-14-1 Heat balance

Heat balance consists of heat input and heat output.

(1) Items of Heat Input

Heat input includes the following items.

- 1. Heat quantity of electric power
- 2. Potential heat of hot metal, hot heel of raw materials and residual slag
- 3. Sensible heat of raw materials
- 4. Calorific power of fuel oil
- 5 Sensible heat of fuel oil
- 6. Oxidation heat of electrode
- 7. Oxidation heat of charge
- 8. Oxidation heat of additives
- 9. Heat of slag formation

(2) Items of Heat Output

Heat output includes the following items.

- 10. Potential heat of output and hot heel of molten steel
- 11. Potential heat of slag
- 12. Heat of decomposition for limestone and iron ore
- 13. Electrical heat loss
- 14. Heat in cooling water
- 15. Sensible heat of exhaust gas
- 16. Heat loss at furnace body
- 17. Other heat loss difference between heat input (items from 1 to 9) and heat output (items from 10 to 16)

12-14-2 Measuring Hems and Equipment/Instruments

資

(1) Measuring Items

Table 12-25 shows measuring items for calculation of heat input and output.

(2) Equipment/Instruments

Equipment/instruments are also shown in Table 12-25 and all equipment/instruments prepared by JICA are shown in Table 12-26.

Table 12-25 Measuring Items and Method and Equipment/Instruments

Measuring Items	Measurement/	Measurement/ Equipment/Instruments	its Preparation of	Measuring/Estimation	Remarks
	Estimation		Equip./Instruments	Interval	
1 Heat quantity of 1) Consumption of electric) M	1) kWh meter	1) Existing, IDC	1) Each heat	
electric power power					
2) Output	2) E	2)	2) .	2) Each heat	2) Calculation. See Note 1).
2 Potential heat of			· .		
hot metal, hot heel of					
raw material and slag					
2-1 Potential heat of					
hot metal					Hot metal was not used in this
					study.
2-2 Potential heat 1) Weight of hot heel	1) E	1) -	1)	1) Constant value	1) Amount of hot heel was
of hot heel					estimated 10 t/heat in this study.
2) Temperature of hot heel	2) E	2) -	3	2) Constant value	2) Hot heel temperature was
3) Output					estimated 1,550 °C in this study.
	з) Е	. (6	3) -	3) Each heat	3) Calculation. See Note 1).
2-3 Potential heat					Small amount of residual slag
of residual slag					was not considered in this study.
3 Sensible heat of 1) Consumption of raw	1) M	1) Weigher	1) Existing, IDC	1) Each bucket	
raw materials materials					
2) Temperature of raw	2) M	2) Surface thermo-	2) JICA	2) At top, middle	2) Surface temperature of bucket
materials		meter		and bottom of each	represented scrap temperature in
				bucket before charge—this study.	this study.
				to EAF	

						- :			
4	Calonific powe	Calonific power 1) Consumption of fuel oil) X	1) Integrator	(1	Existing, IDC	1) Each heat		
		2 3			ć			1) In the second of the second	
ö . 10		2). Low heating value of fuel 2)	n (7.	- (7	₹		 Constant value 	2) IDC prepared standard	
fuel oil	Ę	oil	÷					value.	
		3) Output	3) E	3) -	3		3) Each heat	3) Calculation. See Note 1).	:
N N	5 Sensible heat of							The small sensible heat of fuel	
fuel oil	Ę							oil was not considered in this	1
		The second secon						study.	
0 9	xidation heat of	Oxidation heat of 1). Unit consumption of	1) E	1) • • • • • •	1	•	1) Constant value	1) Unit consumption of	
electrode	apo.	electrode			: .			electrode was statistically	
: · .								estimated in this study.	
	· .	2) C content of electrode	2) E	2) -	5		2) Constant value	2) IDC prepared standard	
								value.	
		3) CO2 and CO contents of	3) M	3) CO/CO2 analyzer	<u> </u>	JICA	3) Continuously		
		exhaust gas		and the second of the second o					
7	Oxidation heat of				*.				
charge	9								
7-1	Oxidation heat	7-1 Oxidation heat 1) Consumption of scrap, pig 1) M	M (1	I) wheigher	$\widehat{}$	Existing, IDC	1) Each heat	1) Charged carbon powder was	
of ch	of charged carbon	iron and charged carbon	G					not used in this study.	
	-	powder	2) E	2) -	2)		2) Constant value	2) IDC prepared standard	
		2) C content of scrap, pig iron	ď					value.	
	-	and charged carbon powder			•				
		3) C content of molten steel	3) M	3) Analyzer	୍ଟି	Existing, IDC	3) Each heat		
:		before tapping			: :				
		4) CO2 and CO content of	4) M	4) CO/CO ₂ analyzer	<u>4</u>	JICA	4) Continuously		
		exhaust gas							
		5) Output	5) E	5)	\$		5) Each heat	5) Calculation. See Note 1).	

7-2 Oxidation heat 1) Consumption of scrap and 1) M 1) Weigher	1)	Existing, IDC	1) Each heat	
of charged silicon pig iron				
2) Si content of scrap and pig 2) E 2) -	7		2) Constant value	Constant value 2) IDC prepared standard
iron				value.
3) Si content of molten steel 3) M. 3) Analyzer	3)	3) Existing, IDC	3) Each heat	
before tapping				
4) Output	4)		4) Each heat	4) Calculation. See Note 1).
7-3 Oxidation heat of charged manganese				Same as 7-2.
7-4 Oxidation heat of charged phosphorus				Same as 7-2.
7-5 Oxidation heat of charged chromium		4 7		Same as 7-2.
7-6 Oxidation heat of charged aluminum		i i		Same as 7-2.
7-7 Oxidation heat 1) Slag weight 1) E 1)	1)	: 	1) Each heat	1) Calculation. See Note 2).
of charged iron 2) FeO, FeO, and CaO 2) M 2) Slag analyzer	/zcr 2)	Existing, IDC	2) Each heat	2) Mean value of 7 heats was
				used in this study.
3) Consumption of burnt lime 3) M 3) Weigher	3)	Existing, IDC	3) Each heat	
4) CaO content of burnt lime 4) E 4) -	(4	•	4) Constant value	4) IDC prepared standard
			S) Hook heat	value.
	•			
8-1 Oxidation heat 1) Consumption of carbon 1) M 1) Weigher	<u>(1</u>	Existing, IDC	1) Each heat	1) Fe-Si, Fe-Mn and Si-Mn
of carbon of injection, Si-Mn, Fe-Si and Fe-				were not used in this study.
additives				
2) C content of carbon 2) E 2)	2)-		2) Constant value	2) IDC prepared standard
injection, Si-Mn, Fe-Si and Fe-				value.
			-	

Standard gas	2) Continuously 4) Each heat 7) Calculation. See Note 1). 7-Si, Fe-Mn and Si-Mn were not used in this study. 7-Fe-Si, Fe-Mn and Si-Mn were not used in this study. 7) Each heat 7) Calculation. See Note 2). 7) Each heat 7) Mean value of 7 heats was used in this study. 7) Each heat 8) See Note 2).
exhaust gas 4) E 4)	4) Each heat 1) Each heat 2) Each heat 3) Each heat
8-2 Oxidation heat of silicon of additives 8-3 Oxidation heat of manganese of additives 9 Heat of slag 1) Slag weight content of slag 3) Consumption of burnt lime 3) M 3) Weigher 2) 10 Potential heat of Molten steel 10-1 Potential heat 1) Temperature of molten 10-2 Potential heat 1) Weight of hot heel 10 Potential heat 1) Temperature of molten 10 Potential heat 1 Temperature of molten 1 Temperature 1 Temperature of molten 1 Temperature 1	4) Each heat 1) Each heat 2) Each heat 3) Fach heat
tion heat to heat to heat to heat se of. 2) CaO, SiO ₂ and P ₂ O ₃ 2) M 2) Siag analyzer 2) CaO, SiO ₂ and P ₂ O ₃ 3) Consumption of burnt lime 3) M 3) Weigher 4) CaO content of burnt lime 4) E 5) Output 5) E 5) Output 5) Informocouple 1) M 1) Thermocouple 1) steel before tapping. 1) Right of hot heel 1) E 1) - 1	1) Each heat 2) Each heat 3) Each heat
tion heat se of slag 1) Slag weight 1) E 1) - content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 5) Output 5) Dutput 5) E 5) - 5) Output 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping steel before tapping 1) E 1) - 11)	1) Each heat 2) Each heat 3) Fach heat
stag 1) Slag weight 1) E 1) - slag 1) Slag weight 1) E 1) - content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 5) Output 5) E 5) - 5) Output 5) E 5) - 5) Inflicat of an anial heat 1) Thermocouple 1) mital heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping	1) Each heat 2) Each heat 3) Each heat
ste of slag 1) Slag weight 1) E 1) - content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 5) Output 5) E 5) - 5) Output 5) E 5) - 5) Infermocouple 1) M 1) Thermocouple 1) all heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping	1) Each heat 2) Each heat 3) Fach heat
slag 1) Slag weight 1) E 1) - 1)- slag 1) Slag weight 1) E 1) - 1)- 2) CaO, SiO ₂ and P ₂ O ₃ 2) M 2) Slag analyzer 2) content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 4) 5) Output 5) E 5) - 5) ial heat of nital heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping nital heat 1) Weight of hot heel 1) E 1) - 1)	1) Each heat 1) Calculation. 2) Each heat 2) Mean value of used in this study. 3) Fach heat
slag 1) Slag weight 1) E 1) - 1)- 2) CaO, SiO ₂ and P ₂ O ₃ 2) M 2) Siag analyzer 2) content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 4) 5) Output 5) E 5) - 5) ial heat of nital heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping nital heat 1) Weight of hot heel 1) E 1) - 1)	1) Each heat 1) Calculation. 2) Each heat 2) Mean value of used in this study.
2) CaO, SiO ₂ and P ₂ O ₃ 2) M 2) Siag analyzer 2) content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 4) ial heat of 5) Output 5) E 5) - 5) ial heat of nital heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping nital heat 1) Weight of hot heel 1) E 1) - 1)	2) Each heat 2) used as Each heat
content of slag 3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 4) 5) Output 5) E 5) - 5) ial heat of intial heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping intial heat 1) Weight of hot heel 1) E 1) - 1)	3) Hach heat
3) Consumption of burnt lime 3) M 3) Weigher 3) 4) CaO content of burnt lime 4) E 4) - 4) ial heat of nitial heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping nitial heat 1) Weight of hot heel 1) E 1) - 1)	6
4) CaO content of burnt lime 4) E 4) - 4) ial heat of intial heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping intial heat 1) Weight of hot heel 1) E 1) - 1)	ì
5) Output 5) E 5) - 5) ial heat of 1) Temperature of molten 1) M 1) Thermocouple 1) ntial heat 1) Temperature of molten 1) M 1) Thermocouple 1) ntial heat 1) Weight of hot heel 1) E 1) - 1)	4) Constant value 4) IDC prepared standard
5) Output ial heat of full heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping futal heat 1) Weight of hot heel 1) E 1) - 1)	value.
ial heat of Integration of molten 1) M 1) Thermocouple 1) Integrated before tapping Integrated 1) Weight of hot heel 1) E 1) - 1)	5) Each heat 5) Calculation, See Note 1).
ntial heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping ntial heat 1) Weight of hot heel 1) \(\mathbb{E} \)	
ntial heat 1) Temperature of molten 1) M 1) Thermocouple 1) steel before tapping ntial heat 1) Weight of hot heel 1) E 1) - 1)	
steel before tapping ntal heat 1) Weight of hot heel 1) E 1) -	, IDC 1) Each heat
ntial heat 1) Weight of hot heel 1) E 1) -	
	1) Constant value 1) Amount of hot heel was
of hot beel of	estimated 10 theat in this study
molten steel 2) Temperature of hot heel 2) M 2) Thermocouple 2) Existing, IDC	, IDC 2) Each heat 2) Temperature of hot heel is
	same as that of molten steel
	before tapping.
3) Output	3) Each heat 3) Calculation. See Note 1).

11 Potential heat of 1) Slag weight	Slag weight 1) 臣	1).	1)	1) Each heat	1) Calculation See Note 2).
slag 2)	CaO content of slag 2	×	2) Slag analyzer	2) Existing, IDC	2) Each heat	2) Mean value of 7 heats was
						used in this study.
3)	3) Consumption of burnt lime 3)	X	3) Weigher	3) Existing, IDC	3) Each heat	
(†	4) CaO content of burnt lime 4)	ш	4).	4)	4) Constant value	4) IDC prepared standard
:		:				value.
(\$	5) Output 5)) E	5) -	5) - (5	5) Each heat	5) Calculation. See note 1).
12 Heat of						
decomposition						
1	Consumption of limestone 1)	<u>ы</u>	1) -	1) -	1) Each heat	1) Limestone of 30 % was
Limestone						included in burnt lime in this
Decomposition						study.
2	CaO content of limestone 2)	щ	2) -	2) -	2) Constant value	2) IDC prepared standard
		: 		* * * * * * * * * * * * * * * * * * *		value
3)	Output 3	ш	3) -	3) -	3) Each heat	3) Calculation, See Note 1).
12-2 Heat of iron						Iron ore was not used in this
ore decomposition						study.
13 Electrical heat						Electrical heat loss was not
toss						measured.
13-1 Heat loss in	· · · · · · · · · · · · · · · · · · ·					
secondary conductors						
15.2. Heat loss of						
transformer						
		٠				
٠						

12	14 Heat in cooling	81	÷		•		
- 1	water						
04	14-1 Heat in	1) Flow rate of cooling water 1) M	1) Magnetic flov	Magnetic flow meter 1) Existing, IDC	1) At start of heat		
	cooling water for	2) Outlet temperature of 2) M	2) Thermocouple	ple 2) JICA	2) Continuously		
	elbow	cooling water					
		3) Inlet temperature of 3) M	3) Thermocouple	ple 3) Existing, IDC	3) At start of heat		
		cooling water					٠.
		4) Power-on to power-off 4) M	4) Clock	4) Existing, IDC	4) Each heat		
		time					
	5) Output	5) Output 5) E	- (5	- (5	5) Each heat	5) Calculation.	See Note 1).
	14-2 Heat in coo	14-2 Heat in cooling water for roof				Same as 14-1.	
	14-3 Heat in coo.	14-3 Heat in cooling water for EBT				Same as 14-1.	
	14-4 Heat in coo.	14-4 Heat in cooling water for shell				Same as 14-1.	
	15 Sensible heat	15 Sensible heat of 1) CO ₂ /CO content of exhaust 1) M	1) CO/CO ₂ and	1) CO/CO ₂ analyzer 1) JICA	1) Continuously	:	
	exhaust gas				1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	:	
		2) O ₂ content of exhaust gas 2) M	2) O ₂ analyzer	r 2) JICA	2) Continuousiy		
		3) Temperature of exhaust 3) M	3) Thermocouple	ple 3) JICA	3) Continuously		
		gas 4) M	4) Digital manometer 4)	nometer 4) JICA	4) Continuously		•
		4) Dynamic and static			: 1		
		pressure of exhaust gas 5) M	5) Clock	5) Existing, IDC	5) Each heat		
		5) Power-on to power-off					
		time 6)E	- (9	- (9	6) Each heat	6) Calculation.	See Note 1).
		6) Output					
	•						

Heat loss at 1) Power-on to power-off 1) M 1) Clock 1) Existing, IDC 1) 2) Surface area of roof 2) E 2) - 2) 3) Output 3) E 3) - 31 - 33 4) Surface temperature 4) M 4) Pyrometer 4) IICA 4) heat loss at 1) Power-on to power-off 1) M 1) Clock 1) Existing, IDC 1) time 2) Surface area of shell 2) E 2) - 2) - 2) 3) Output 3) E 3) - 31 - 33 4) Surface temperature 5) M 5) Pyrometer 6) IICA 6) 5) Indoor temperature 7) M 4) Pyrometer 7) IICA 4) 5) Indoor temperature 5) M 5) Pyrometer 7) IICA 5) Heat loss at 30 Output 5) M 5) Pyrometer 7) IICA 6)	16 Heat loss at						
2) Surface area of roof 2) E 2) - 2) 3) Output 3) E 3) - 3) 4) Surface temperature 4) M 4) Pyrometer 4) IICA 4) heat 5) Indoor temperature 5) M 5) Pyrometer 5) IICA 5) Heat loss at 1) Power-on to power-off 1) M 1) Clock 1) Existing, IDC 1) 2) Surface area of shell 2) E 2) - 2) 2) Surface temperature 4) M 4) Pyrometer 4) IICA 4) 4) Surface temperature 5) M 5) Pyrometer 5) IICA 5) 5) Indoor temperature 5) M 5) Pyrometer 5) IICA 5)	6-1 Heat loss at	1) Power-on to power-off	1) M	1) Clock			
3) Output 4) Surface temperature 4) M 4) Pyrometer 4) IICA 4) hea 5) Indoor temperature 5) M 5) Pyrometer 5) IICA 5) Indoor temperature 5) M 5) Pyrometer 5) IICA 5) Indoor temperature 6) M 6) Pyrometer 7) IICA 6) IICA 7) Indoor temperature 7) B 7) Clock 7) Indoor temperature 7) B 7) Indoor temperature 7) M 7) Pyrometer 7) IICA 7) Indoor temperature 7) M 7) Pyrometer 7) IICA 8) I		2) Surface area of roof	2) E	2)	?	2) Constant value	2) IDC prepared standard value.
4) Surface temperature 4) M 4) Pyrometer 4) JICA hea 5) Indoor temperature 5) M 5) Pyrometer 5) ICA 5) time 2) Surface area of shell 2) E 2) - 2) 2) Surface temperature 4) M 4) Pyrometer 4) JICA 4) 4) Surface temperature 5) M 5) Pyrometer 4) JICA 4) Heat loss at Heat loss at		3) Output	3) E	3)	3) -		3) Calculation. See Note 1).
heat loss at 1) Power-on to power-off 1) M 1) Clock 1) Existing, IDC 1) time 2) E 2) Clock 2) Surface area of shell 2) E 2) Clock 3) Clock 3) Clock		4) Surface temperature	4) M	4) Pyrometer	4) JICA	4) A few times a	4) Mean value of ten tunes
Solution						heat	measurement was used for each
Solution temperature Solution						:	heat in this study.
Heat loss at 1) Power-on to power-off 1) M 1) Clock 1) Existing, IDC 1) 2) Surface area of shell 2) E 2) - 2) - 2) 3) Output 3) E 3) - 3) - 3) 4) Surface temperature 4) M 4) Pyrometer 4) IICA 4) 5) Indoor temperature 5) M 5) Pyrometer 5) IICA 5) Heat loss at 5) Accounter 5) IICA 5)		5) Indoor temperature	5) M		•	5) At start of heat	
time 2) Surface area of shell 2) E 2) - 2)- 2) 3) 3) Output 3) E 3) - 4) Surface temperature 4) M 4) Pyrometer 4) IICA 5) Indoor temperature 5) M 5) Pyrometer 5) IICA 5) Heat loss at	6-2 Heat loss at	1) Power-on to power-off	1) M	1) Clock	1) Existing, IDC	1) Each heat	
3) - 3) - 3) - 3) - 3) - 3) - 3) - 3) -	hell	time 2) Surface area of shell	2) E	2) •	2)-	2) Constant value	2) IDC prepared standard
3) Output 3) E 3) - 3) - 3) 6 4) Surface temperature 4) M 4) Pyrometer 4) JICA 4) Heat loss at							value
4) Surface temperature 4) M 4) Pyrometer 4) JICA 4) 5) Indoor temperature 5) M 5) Pyrometer 5) JICA 5) Heat loss at		3) Output	3) E	3) - (8	3)		3) Calculation. See Note 1).
5) Indoor temperature 5) M 5) Pyrometer 5) JICA 5) Heat loss at		4) Surface temperature	4 M	4) Pyrometer		4) Continuously	
16-3 Heat loss at		5) Indoor temperature	s) M	5) Pyrometer	s) JICA	5) At start of heat	***************************************
	16-3 Heat loss at						Same as 16-2
ροποιο	oottom					The second secon	

Note 1) Output means molten steel tapped into the ladle. It consists of billets, crop and skulls in the ladle and tundish. These weights could not be actually weighed. Output is calculated as follows:

Output = Charged raw materials x steel yield

Steel yield is statistically determined.

Note 2) As generated slag could not be actually weighed, slag weight is calculated by mass balance of CaO as follows: Amount of slag = Consumption of burnt lime x CaO content of burnt lime/CaO content of slag.



	ITEM	DESCRIPTION
a-1)	CO/CO ₂ analyzer 1 set	To measure CO and CO ₂ contents of the exhaust gas at C/C inlet
a-2)	CO/CO ₂ analyzer 1 set	To measure CO and CO ₂ contents of the exhaust gas at C/C outlet
b-1) ⁽	O ₂ analyzer 1 set	To measure O ₂ content of the exhaust gas at C/C inlet
b-2)	O ₂ analyzer 1 set	To measure O ₂ content of the exhaust gas at C/C outlet
c)	Pretreatment unit (filter, drain pot, cooler) 1 set	To remove dusts and moisture in the exhaust gas
d-1)	Data logger 2 sets	To input into the personal computer the output data (analog signals) from the measuring devices after they are converted into digital signals
d-2)	Data logger 1 set	Stand-by
e-1)	Personal computer 2 sets	To record and exhibit on the monitoring screen the output data after being converted to digital signals by the data logger
e-2)	Personal computer 1 set	Stand-by. To be used in case of trouble of one of e-1). Actually this computer was used
f-1)	Thermocouple PR type (JIS type R) 4 sets	To measure the temperature of the exhaust gas at C/C inlet
f-2)	Thermocouple CA type (JIS type K) 24 sets: 1 set for outlet, 16 sets for furnace, 7 sets for spare	To measure the temperature of the exhaust gas at C/C outlet and temperature of the furnace shell and bottom
f-3)	Thermocouple CC type (JIS type T) 10 sets (including 4 sets of spare)	To measure the temperature of the cooling water for the furnace
g-1)	Cable of 100 m for thermocouple for CA type	To connect the thermocouple to the data logger
g-2)	Cable of 1,000 m for thermocouple for CC type	To connect the thermocouple to the data logger
g-3)	Cable of 300 m for thermocouple for CA	To connect the thermocouple to the data logger

	type	
h)	Ultrasonic flow meter 1 set	To measure the flow rate of cooling water for the furnace. Not used. IDC's instrument
		was used.
i)	Gas sampling unit 4 pieces	Sampler for the exhaust gas at C/C inlet.
		Water cooled. Stainless-steel-made for C/C outlet.
j-1)	Digital manometer 2 sets	To measure the dynamic and static pressure of
:		the exhaust gas for measurement of the flow rate at the C/C outlet.
j-2)	Digital manometer 2 sets	Stand-by
k)	Pitot' tube 4 sets	Used for measurement of the flow rate of the exhaust gas at the C/C outlet.
l)	Surface thermometer 2 sets	To measure the surface temperature of the
<u> </u>		scrap bucket for scrap temperature
m)	Printer: 1 set	To print out the of measurement results.
n-1)	Transformer 3 sets	Step down transformer (200 V to 100V) and stabilizer for instruments
n-2)	Transformer 1 set	Stand-by
0)	Pyrometer (Thermometer) 2 sets	To measure the roof surface temperature and
		for back-up use for measurement of surface
		temperature of the furnace shell and bottom.
		One is for temperatures higher than 800°C and
- : : : : :		the other is for temperatures lower than 500 °C
p) l	Equipment for moisture 1 set	To measure moisture in the exhaust gas.

12-14-3 Calculation Formulas and Reference Figures for Heat Balance, and Calculation of Heat Content of Exhaust Gas

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The heat balance calculating method in arc furnace complies with Heat Balance System in Arc Furnace (JIS G 0703).

(1) Calculation Formulas and Reference Figures for Heat Balance

Calculation formulas and reference figures for the heat balance are shown below:

Table 12-27 Calculation Formulas for Heat Input

Table 12-28 Calculation Formulas for Heat Output

Table 12-29 Heat Content (Iron, Steel and Slag)

Table 12-30 Reaction Heat

Table 12-27 Calculation Formulas for Heat Input

(1) Heat Quantity of Electric	$Q_1 = W_1 \times 860$
Power, Qt (kcal/t-output)	W ₁ . Unit consumption of electric power
	(kWh/t-output)
	860: Conversion factor (kcal/kWh)
(2) Potential Heat of Hot Metal,	$Q_2 = Q_3 + Q_{2b} + Q_{2c}$
Hot Heel of Raw Materials and	(a) Potential Heat of Hot Metal, Q2a (kcal/t-output)
Slag, Q2 (kcal/t-output)	$Q_{2a} = M_{2a} \times H_{2a}$
	M2a: Unit consumption of hot metal (kg/t-output)
	H _{2a} : Heat content of metal (kcal/kg)
	Note: Hot metal was not used in this study.
	(b) Potential Heat of Hot Heel, Q26 (keal/t-output)
	$Q_{2b} = M_{2b} \times H_{2b}$
	M _{2b} : Unit consumption of hot heel (kg/t-output)
	H _{2b} : Heat content of hot heel (kcal/kg)
	Note: 1) The amount of hot heel was estimated
	to be 10t/heat in this study.
	2) Heat content depends on the hot heel
	temperature which was estimated to be 1,550 °C i
	this study.
名《新聞》。 第4日本教	(c) Potential Heat of Residual Slag, Q2. (kcal/t-
	output)
	$Q_{2e} = M_{2e} \times H_{2e}$
	M _{2c} : Unit consumption of residual slag (kg/t-output)
	H _{2c} : Heat content of residual slag (kcal/kg)
	Note: Small amounts of residual slag were not
	considered in this study.
(3) Sensible Heat of Raw	$Q_3 = M_3 \times (II_{3h} - II_{3a})$
Materials, Q3 (kcal/t-output)	M ₁ : Unit consumption of raw materials (kg/t-output)
	H _{sh} Heat content of raw materials at temperature after
	preheating (kcal/kg)
	H _{3a} Heat content of raw materials at indoor
	temperature (kcal/kg)
	Note: 1) H ₃ was assumed 0 kcat/kg of heat conter

	at 0 °C as basis in this study.
(4) Calorific Power of Fuel Oil,	$\mathbf{Q_4} = \mathbf{M_4} \times \mathbf{q_4}$
Q4 (kcal/t-output)	M4: Unit consumption of fuel oil (kg/t-output)
	q4: Low heating value of fuel oil (kcal/kg)
(5) Sensible Heat of Fuel Oil,	$Q_5 = M_5 \times C_5 \times (Z_{51} - Z_{52})$
Q ₅ (kcal/t-output)	Ms: Unit consumption of fuel oil (kg/t-output)
	C ₅ : Mean specific heat of fuel oil (kcal/kg)
	Z _{sf} : Temperature of fuel oil (°C)
	Z _{5a} : Indoor temperature (°C)
	Note: Small sensible heat of fuel oil was not
	considered in this study.
(6) Oxidation Heat of Electrode,	$Q_6 = M_6 \times C_c \times 10^{-2} \times (q_{6CO2} \times CO_2)/(CO_2 + CO)$
Q6 (kcal/t-output)	$+ q_{6CO} \times CO/(CO_2 + CO)$
	Ms: Unit consumption of electrode (kg/t-output)
· 新闻《自己》 (1) [4] [6] [6] [6]	Ce. Carbon content of electrode (%)
	q _{CO2} : Oxidation heat of electrode in CO ₂ formation
	(kcal/kg)
	q _{6CO} : Oxidation heat of electrode in CO formation
	(kcal/kg)
	CO ₂ : CO ₂ content of exhaust gas (%)
	CO: CO content of exhaust gas (%)
	Note: Unit consumption of electrode was estimated
	1.9 kg/t-output.
7) Oxidation Heat of Charge,	$Q_7 = Q_{7a} + Q_{7b} + Q_{7c} + Q_{7d} + Q_{7c} + Q_{7l} + Q_{7g}$
Q7 (kcal/t-output)	(a) Oxidation Heat of Charged Carbon,
	Q _{7s} (kcal/t-output)
	$Q_{1a} = M_{7a} \times (q_{7CO2} \times CO_2/(CO_2 + CO) + q_{7CO} \times CO$
	/(CO ₂ + CO))
	M7a = M7aPig iron + M7aScrap + M7aCarbon position
	$-1,000 \times C_{Tap} \times 10^{-2}$
	$M_{7aPig iron} = M_{7Pig iron Charge} \times C_{Pig iron} \times 10^{-2}$
	$M_{7aScrap} = M_{7Scrap Charge} \times C_{Scrap} \times 10^{-2}$
	M _{7aCarbon powder} = M _{7Carbon powder Charge x C_{Carbon powder} x 10⁻²}
	M7. Oxidation amount of charged carbon (kg/t-output)
	1071. Oxidation amount of charged carbon (kg/courput)

M7aScrap: Carbon in scrap (kg/t-output)

M_{7aCarbon powder}: Carbon in charged carbon powder (kg/t-output)

M_{7Pig iron Charge}: Unit consumption of pig iron (kg/t-output)

M78crap Charge: Unit consumption of scrap (kg/t-output)

M_{7Carbon powder Charge}: Unit consumption of charged carbon powder (kg/t-output)

Cpig iron: C content of pig iron (%)

C_{Scrap}: C content of scrap (%)

Ccarbon powder. C content of charged carbon powder (%)

C_{Tap}: C content of molten steel before tapping (%)

1,000. Output (kg)

q_{7CO2}: Heat of carbon oxidation in CO₂ formation (kcal/kg)

q_{2CO}: Heat of carbon oxidation in CO formation (kcal/kg)

CO₂: CO₂ content of exhaust gas (%)

CO: CO content of exhaust gas (%)

Note: Charged carbon powder was not used in this study.

(b) Oxidation Heat of Charged Silicon, Q_{7b} (kcal/t-output)

 $Q_{7b} = M_{7b} \times q_{7b}$

 $M_{7b} = M_{7b \text{Fig iron}} + M_{7b \text{Scrap}} - 1,000 \text{ x Si}_{\text{Tap}} \text{ x } 10^{-2}$

 $M_{7bPig iron} = M_{7iSg iron Charge} \times Si_{Pig iron} \times 10^{-2}$

M76Scrap = M7Scrap Charge x Siscrap x 10-2

M7b: Oxidation amount of charged silicon (kg/t-output)

M7bFig iron: Silicon in pig iron (kg/t-output)

M7bScrap: Silicon in scrap (kg/t-output)

M_{7Pig iron Charge}: Unit consumption of pig iron (kg/t-output)

M7Scrap Charge: Unit consumption of scrap (kg/t-output)

Sirigiron: Si content of pig iron (%)

Siscrap: Si content of scrap (%)

Si_{lap}: Si content of molten steel before tapping (%)

1,000: Output (kg)

q76: Heat of silicon (keal/kg)

(c) Oxidation Heat of Charged Manganese, Q_{7c} (keal/t-output)

Same as (b).

(d) Oxidation Heat of Charged Phosphorus, Q_{7d} (kcal/t-output)
Same as (b).

(e) Oxidation Heat of Charged Chromium, Q7. (kcal/t-output)
Same as (b).

(f) Oxidation Heat of Charged Aluminum,
Qn(kcal/t-output)
Same as (b).

(g) Oxidation heat of Charged Iron, Q7g (keal/t-output)

 $Q_{7g} = M_{7g} x$ (FeO x 0.777 x $q_{7gFeO} + Fe_2O_3 x 0.699 x$

 $q_{7gFe2O3}$) x 10^{-2}

M_{7g}: Unit weight of slag (kg/t-output)

FeO: FeO content of slag (%)

Fe₂O₃: Fe₂O₃ content of slag (%)

q₇₆FeO: Heat of iron oxidation in FeO formation (kcal/kg)

q_{7gFe2O3}: Heat of iron oxidation in Fe₂O₃ formation (kcal/kg)

0.777; Ratio of Fe in FeO

0.699: Ratio of Fe in Fc₂O₃

Note: Unit weight of slag was calculated as follows in this study:

 $M_{7g} = M_{7gl,ime} \times CaO_1/CaO_8$

M_{7gl inc}. Unit consumption of burnt lime (kg/t-output)

CaOs: CaO content of slag (%)

CaO_L: CaO content of burnt lime (%)

Note: Mean value of seven heats was used for slag

(8) Oxidation Heat of Additives, Q₈ (kcal/t-output)

 $Q_8 = Q_{8a} + Q_{8b} + Q_{8c}$

(a) Oxidation Heat of Carbon in Additives, Q82 (kcal/t-output)

 $Q_{8a} = M_{8a} \times (q_{8CO2} \times CO_2/(CO_2 + CO) + q_{8CO} \times CO$ /(CO₂ + CO))

 $M_8 = M_{8aCarbon\,injection} + M_{8aSi-Mn} + M_{8aFe-Si} + M_{8aFe-Mn}$

 $M_{8aCarbon injection} = M_{8Carbon injection Additives} \times C_{Carbon injection}$ $\times 10^{-2}$

 $M_{\text{SaSi-Mn}} = M_{\text{SSi-Mn Additives}} \times C_{\text{Si-Mn}} \times 10^{-2}$

 $M_{8aFe-Si} = M_{8Fe-Si~Additives} \times C_{Fe-Si} \times 10^{-2}$

 $M_{8aFe-Mn} = M_{8Fe-Mn \text{ Additives}} \times C_{Fe-Mn} \times 10^{-2}$

M8a: Carbon of additives(kg/t-output)

MsaCarton injection: Carbon in carbon injection (kg/t-output)

M_{8aSi-Mn}: Carbon in Si-Mn (kg/t-output)

M_{8aFe-Si}: Carbon in Fe-Si (kg/t-output)

M8aFe-Mn Carbon in Fe-Mn (kg/t-output)

M_{8Carbon injection Additives}: Unit consumption of carbon injection (kg/t-output)

M_{88i-Ma Additives}: Unit consumption of Si-Mn (kg/t-output)

MsFe-Si Additives: Unit consumption of Fe-Si (kg/t-output)

M_{8Fe-Mn Additives}. Unit consumption of Fe-Mn (kg/t-output)

Ccarbon injection: C content of carbon injection (%)

C_{Si-Mn}: C content of Si-Mn (%)

CFe-Si C content of Fe-Si (%)

CFe-Mn: C content of Fe-Mn (%)

q_{8CO2}: Heat of carbon oxidation in CO₂ formation (kcal/kg)

 q_{sco} . Heat of carbon exidation in CO formation (kcal/kg)

CO2: CO2 content of exhaust gas (%)

CO: CO content of exhaust gas (%)

Note: Si-Mn Fe-Si and Fe-Mn were not used in this

study.

(b) Oxidation Heat of Silicon in Additives,Q_{8b} (keal/t-output)

18

 $Q_{8b} = M_{8b} \times q_{8b}$

 $M_{8b} = M_{8bSi-Mn} + M_{8bFc-Si} + M_{8bFc-Mn}$

 $M_{888i-Mn} = M_{88i-Mn Additives} \times Si_{8i-Mn} \times 10^{-2}$

 $M_{8Fe-Si} = M_{8Fe-Si \ Additives} \times Si_{Fe-Si} \times 10^{-2}$

 $M_{8NFe-Mn} = M_{8Fe-Mn \ Additives} \times Si_{Fe-Mn} \times 10^{-2}$

M_{8b}: Silicon in additives (kg/t-output)

M_{8bSi-Mn}: Silicon in Si-Mn (kg/t-output)

M_{SbFe-Si}: Silicon in Fe-Si (kg/t-output)

Mshre-Ma. Silicon in Fe-Mn (kg/t-output)

M_{8Si-Mn Additives}: Unit consumption of Si-Mn (kg/t-output)

Mare-Si Additives: Unit consumption of Fe-Si (kg/t-output)

M_{SFe-Mn Additives}: Unit consumption of Fe-Mn (kg/t-output)

Sisi-Mn: Si content of Si-Mn (%)

Sire Si Si content of Fe-Si (%)

SiFe-Mn: Si content of Fe-Mn (%)

1,000. Output (kg)

q_{8b}: Heat of silicon (kcal/kg)

Note: Si-Mn, Fe-Si and Fe-Mn were not used in this study.

(c) Oxidation Heat of Manganese in Additives,Q_{8c} (kcal/t-output)

Same as (b).

(9) Heat of Stag Formation, Q2 (kcal/t-output)

$Q_9 = M_9 x (SiO_2 x q_{9SiO_2} + P_2O_5 x q_{9P2O_5}) x 10^{-2}$

M9: Unit weight of slag (kg/t-output)

Note: Sec (7) (g).

q_{98iO2}: Heat of SiO₂ reaction in Ca₂SiO₄ formation (kcal/kg)

q_{9P2O5}: Heat of P₂O₅ reaction in Ca₃P₂O₃ formation (kcal/kg)

SiO₂: SiO₂ content of slag (%)

•		
· · · · · · · · · · · · · · · · · · ·	P2O5: P2O5 content of slag (%)	
	Note: The mean value of seven he	ats was used for
	slag analysis in this study.	
Heat Input,	$Q_{ilest loput} = Q_1 + Q_2 + Q_3 + Q_4 + Q_5 + Q_$	$Q_6 + Q_7 + Q_8$
QHeat Input (kcal/t-output)	+ Q ₉ (kcal/t-output)	· · · · · · · · · · · · · · · · · · ·
the state of the s		



(10) Potential Heat of Molten Steel, Q₁₀ (kcal/t-output)

 $Q_{10} = Q_{10a} + Q_{10b}$

(a) Potential Heat of Output (excluding hot heel),

Q102 (kcal/t-output)

 $Q_{10a} = 1,000 \times H_{10a}$

1,000: Output (kg)

H_{10a}: Heat content of molten steel before tapping (kcal/g)

Note: Heat content is dependent on temperature.

(b) Potential Heat of Hot heel, Q(Da (kcal/t-output)

 $Q_{10b} = M_{10b} \times H_{10b}$

M_{10b}: Unit weight of hot heel (kg/t-output)

H_{10b}: Heat content of hot heel (kcal/g)

Note: 1) Amount of hot heel was assumed 10 t/heat in this study.

2) Heat content is dependent on temperature

(11) Potential Heat of Slag, Q₁₁ (kcal/t-output)

 $\mathbf{Q_{H}} = \mathbf{M_{H}} \times \mathbf{H_{H}}$

M₁₁: Unit weight of slag (kg/t-output)

H₁₁: Heat content of slag (kcal/g)

Note: 1) Concerning unit weight of slag, see (7) (g) in Heat Input.

2) Heat content is dependent on temperature which is same as that of molten steel in this study.

(12) Heat of Decomposition, Q₁₂ (kcal/t-output)

 $\mathbf{Q_{12}} = \mathbf{Q_{12a}} + \mathbf{Q_{12b}}$

(a) Heat of Limestone Decomposition,

Q12. (kcal/t-output)

 $Q_{12a} = 10^{-2} \text{ x M}_{12a} \text{ x CaO x } q_{12aCaCO3}$

M_{12a}: Unit consumption of limestone (kcal/t-output)

q_{12aCaCO3}: Heat of decomposition of limestone (kcal/kg)

CaO: CaO content of limestone (%)

Note: 1) 30 % of limestone was included in burnt lime in this study.

(b) Heat of Iron Ore Decomposition,

Q_{12b} (kcal/t-output)



 $Q_{12b} = 10^{-2} \text{ x M}_{12b} \text{ x (FeO x } q_{12bFeO} + Fe_2O_3 \text{ x } q_{12bFe2O_3})$

M₁₂₆: Unit consumption of iron ore (kcal/t-output)

q_{12N/co}: Heat of decomposition of FeO in iron ore (kcal/kg)

q_{12bFe2O3}: Heat of decomposition of Fe₂O₃ in iron ore (kcal/kg)

FeO: FeO content in iron ore (%)

Fe₂O₃: Fc₂O₃ content in iron ore (%)

Note: Iron ore was not used in this study.

(13) Etectrical heat Loss, Q₁₃ (keal/t-output)

$$Q_{13} = Q_{13a} + Q_{13b}$$

(a) Heat Loss in Secondary Conductors,

Q17. (keal/t-output)

$$Q_{13a} = (R_0 \times I_0^2 \times T \times 860) \times 3.6/\Gamma$$

$$R_0 = R_S + O_X (E_1 + E_2 + E_3)/S$$

$$I_0 = (W_0 \times 10,000)/(3^{1/2} \text{ VO } \times \cos \text{ U})$$

$$W_0 = W_0 / \Gamma$$

$$V_0 = V \times T/\Gamma$$

$$\cos U = W_p/(W_0^2 + W_0^2)^{1/2}$$

R_o: Combined resistance of secondary conductor and electrode (Ohm)

Io. Mean current (A)

t: Output (ton)

R_S: Combined resistance of secondary conductor resistance and contact resistance between electrode and holder (Ohm)

O. Specific resistance of electrode (Ohm-cm)

S: Sectional area of electrode (cm²)

 E_1 , E_2 , E_3 : Average length of electrode in each phase exposed from roof

Wo: Mean electric power (kW)

V₀: Mean voltage (V)

cos U: Mean power factor (-)

Wn: Electric power consumed (kW)

Wo: Reactive energy (kVar)

- V: Secondary voltage of transformer tap(V)
- T': Conducting period of each tap in transformer (hr)
- T: Power-on to power-off time (hr)

 Note: Heat loss in secondary conductor was not measured.
- (b) Heat Loss of Transformer,

Q_{13b} (kcal/t-output)

$$Q_{136} = (W_1 - W_2) \times 860/t$$

- W₁ = Electric power on primary side of transformer (kWh)
- W₂ = Electric power on secondary side of transformer (kWh)

Note: Heat loss in secondary conductor was not measured.

(14) Heat in Cooling Water, Q₁₄ (keal/t-output)

$$Q_{14} = Q_{14a} + Q_{14b} + Q_{14c} + Q_{14d}$$

(a) Heat in Cooling Water for Elbow, Q₁₄, (kcal/t-output)

 $Q_{14a} = M_{14a} \times C_{14} \times (t_{a14aO} - t_{14aI})$

 $M_{14a} = F_{14a} \times T/t$

M14a: Average quantity of cooling water (kg/t-output)

 C_{14} : Specific heat of water (kcal kg, °C) = 1

t_{14aO}: Mean outlet temperature of cooling water (°C)

t_{14at}: Mean inlet temperature of cooling water (°C)

F_{14a}. Flow rate of cooling water (kg/hr)

T: Power-on to power-off time (hr)

t: Output (ton)

(b) Heat in Cooling Water for Roof, Q146 (kcal/t-output)

Same as (a).

(c) Heat in Cooling Water for EBT, Q14c (kcaVt-output)

Same as (a).

(d) Heat in Cooling Water for Shell, Q14d (keal/t-output)

Same as (a).

Q15 (kcal/t-output)

1

H₁₅ = Heat in average flow of exhaust gas (kcal/min)

T = Power-on to power-off time (hr)

t = Output (ton)

Note: Heat in average flow of exhaust gas is described in another pages (3. Calculation of Heat Content of Exhaust Gas).

(16) Heat Loss at Furnace Body, Q16 (kcal/t-output)

 $y_1 = Q_{16} = Q_{16} + Q_{16} + Q_{16}$

(a) Heat Loss at Roof, Q164 (kcal/t-output)

 $Q_{16a} = T x (q_{16aR} + q_{16aC}) x Mt$

 $q_{16aR} = 4.88 \text{ x r x} \left[(T_0/100)^4 - (T_a/100)^4 \right]$

 $q_{16aC} = p \times (t_0 - t_a)^{1.25}$

q16ak: Radiation heat loss at roof (kcal/m2, hr)

q_{16aC}: Convection heat loss at roof (kcal/m₂, hr)

T: Power-on to power-off time (hr)

A: Surface area of roof (m₂)

t: Output (ton)

r. Degree of blackness on furnace surface due to radiation (0.8)

To: Surface temperature of roof (°C)

Ta: Indoor temperature (°C)

p. 2.8 for horizontal wall facing upward, roof

2.2 for vertical wall facing sideways, shell

1.5 for horizontal wall facing downward, bottom

Note: Degree of blackness of 0.8 is based on "Heat

Calculating Figures for Iron and Steel Making

(1966)" by the Japan and Steel Association, Society

of Japan Academic Development).

(b) Heat Loss at Shell, Q166 (kcal/t-output)

Same as (a)

(c) Heat Loss at Bottom, Qtec (kcal/t-output)

Same as (a)

(17) Other Heat Loss,

Q17 (kcal/t-output)

 $Q_{17} = Q_{heat Input} - (Q_{10} + Q_{11} + Q_{12} + Q_{13} + Q_{14} + Q_{15}$

+ Q16

Table 12-29 Heat Content (Iron, Steel and Stag)

									Unit	kcal/kg
Tempera-	Pig iron	Pure iron	Mild steel	0.23 %	0.4 %	0.8 %	1.2 %	18 Cr	13 Cr	Slag
ture (°C)				СС	c	C	<u> </u>	- 8 Ni	·	
. 0	: -	0.0	-	0.0	0.0	0.0	0.0	0.0	0.0	-
50	· · · · •	· -	•	5,6	5.6	5.4	5.4	5.9	5.2	-
100		11.0	.	11.4	11.4	11.2	11.2	12.0	10.8	19.1
150		•	-	17.4	17.4	17.4	17.4	18.3	16.8	
200	- : :	23,0	· • •	23.6	23.5	23.8	23.9	24.7	23.0	39.9
250				30,0	29.8	30.3	30.4	31.1	29.3	
300	* :	35.0		36.6	36.4	37.1	37.1	37.7	35.9	60.0
350	-	-		43.5	43.2	44.1	44.0	44.3	42.8	
400		49.0	•	50.6	50.2	51.3	51.1	51.1	50.0	81.0
450		•		58,1	57,5	58.8	58.8	58.1	57.6	· · · · · ·
500	<u> </u>	64.0		66.0	65.3	66.8	66,1	65.2	65.8	101.9
550				71.4	73.5	75.1	74.0	72.7	74.4	
600	- <u>-</u>	82.0	· . · · · <u>-</u>	83.3	82.0	83.6	82.3	80.5	83.8	129.0
650	: •	·		92.7	90.7	92.3	91.2	88.0	93.5	· .
700	•	102.0		102.8	99.9	101.5	101.0	95.5	104.0	151.9
750		•	•	119.9	118.8	126.8	125.9	102,9	114.8	7
800	-	125.0	-·	131.3	126.2	133.6	133.7	110.6	123.0	177.0
8 50			-	140.1	132 3	141.5	141.5	118.3	132.6	
900		145.7	_	147.8	138.8	148.9	148.9	126.0	140.6	201.8
950		4 - 1 - 1 <u>-</u>		155,6	146.3	156.3	156.4	133.8	148.4	· : · · <u>-</u>
1000	<u>.</u>	163,0	<u> </u>	163.3	153,7	163.9	163.8	141.5	156.1	227.9
1050	·	•		171.1	161,3	171.5	171.5	149.3	163.9	
1100	•	178.0		178 8	168.8	179.3	179.1	157.2	171.7	254,8
1150		· .	-	186.7	176.5	187.2	186.9	165.2	179.5	
1200	264.9	194.0	•	194.6	184.3	195.2	194.7	173.2	187.2	286.9
1250	: -	1 1		202.7	192.3	203,3	202.7	181.3	195.0	
1300	280.9	209.0		210.9	200.5	211.4	210.7	189.4	202 8	321.7
1350	•		-	(219.2)	(208.9)	(219.5)	(218.7)	(197.5)	(210.6)	_
1400	301.9	231.0		(227.5)	(217.5)	(227.5)	(226.7)	(205.6)	(218.4)	365.7
1450	¹⁰ , 2	÷ .	: ·	(236.1)	(226.3)	(235.6)	(234,7)	(213.6)	(226.2)	
1500	322.0	247.9	(311.0)							406.8
1550	·			•					•	431.6
1600	343.0	331.0	332.0				1 1 2		·	459.8
1650	-		•	-	•			11.		489.6
1700	361.0	349.9	353.0						•	519.7

Basic stag of steelmaking process, CaO = 43.55 %, SiO₂ = 34.22 %, FeO = 10.27 %, Fe₂O₃ = 3.68 %, Al₂O₃ = 4.63 %,

MgO = 11.84 % and MnO = 6.60 %

			Ont. Kcankg
	item	Reaction Heat	Reaction
Oxidation Heat	Graphite carbon	7,829 (Graphite carbon)	$C + O_2 = CO_2$
	Graphite carbon	2,200 (Graphite carbon)	$C + 1/2 O_2 = CO$
	C	8,075 (C)	$C + O_2 = CO_2$
	\mathbf{c}	2,448 (C)	$C + 1/2 O_2 = CO$
	S i	7,459 (Si)	$Si + O_2 = SiO_2$
	Mn	1,674 (Mn)	$Mn + 1/2 O_1 = MnO$
÷ .	Þ	5,811 (P)	$P + 5/4 O_2 = 1/2 P_2 O_5$
	Cr	2,620 (Cr)	$Cr + 3/4 O_2 = 1/2 Cr_2O_3$
		7,419 (AI)	$A1 + 3/4 O_2 = 1/2 Al_2O_3$
	Fe	1,151 (Fe)	$Fe + 1/2 O_2 = FeO$
	Fe	1,756 (Fe)	$Fe + 3/4 O_2 = 1/2 Fe_2O_3$
Formation Heat	Slag	502 (SiO ₂)	2 CaO + SiO2 = CaSiO ₄
	Slag	1,070 (P ₂ O ₅)	$3 \text{ CaO} + P_2O_5 = \text{Ca}_3P_2O_3$
Decomposition	Iron ore	896 (FeO)	$FeO = Fe + 1/2 O_2$
Heat	Iron ore	1,228 (Fe ₂ O ₃)	$Fe_2O_3 = 2 Fe + 3/2 O_2$
	Lime stone	757 (CaO)	$CaCO_3 = CaO + CO_2$

(2) Calculation of Heat Content of Exhaust Gas

1) Measurement of Velocity, Flow Rate and Composition of Exhaust Gas

At C/C-outlet point, velocity and flow rate of exhaust gas are continuously measured based on JIS Z 8808 -7 ("Method of measuring dust concentration in flue gas" 7. Measurement of Velocity and Flow of Flue Gas).

Gas composition is also measured continuously both at C/C-outlet point and C/C-inlet point.

2) Calculation of Heat Content of Exhaust Gas

Combined gas flow rate at C/C-outlet with change of gas composition from CC-inlet point to CC-outlet point, mass balance equilibrium gives gas flow rate at C/C-inlet. To calculate the heat content of exhaust gas at C/C-inlet point, linearized specific heat is referred to JIS G 0703 ("Method of heat balance calculation of are furnace")

$$g_0 = \{ (44 \times CO_{20H} + 32 \times O_{20H} + 28 \times (100 - CO_{20H} - O_{20H}) \} \times (1 - H/100) + 18 \times H \}$$
/(22.4 x 100) (kg/Nm³)-------(1)

$$g = (g_0 \times 273/(273 + T_{OUT})) \times \{(P_{Alm} \times 100/9.81 + SP_{OUT})/(13.6 \times 760)\} \quad (kg/m^3) - \cdots$$
----(2)

$$V = 0.854 \text{ x} \sqrt{2 \times 9.81 \text{ x} DPour / g}$$
 (m/s)(3)

$$F_{OUT} = V \times p \times (1840/2000)^2 \times 60 \times \{273/(273 + T_{OUT})\} \times (P_{Atm} \times 100/9.81 + SP_{OUT})/(13.6 \times 760) \text{ (Nm}^3/min)------(4)$$

$$F_{IN} = F_{OUT} \times (CO_{IN} + CO_{2IN}) / (CO_{OUT} + CO_{2OUT}) (Nm3/min) ------(5)$$

$$q_{IN} = F_{IN} \times T_{IN} \times \{q_{CO} \times CO_{IN} + q_{CO2} \times CO_{2IN} + q_{O2} \times O_{2IN} + q_{N2} \times (100 - CO_{IN} - CO_{2IN} - O_{2IN})\} / (4.186 \times 100)$$

$$q_{CO} = 0.00013 \text{ T x T}_{IN} + 1.28 \text{ (kj / Nm}^3 \text{ C)}$$

$$q_{CO2} = 0.000397 \text{ T x T}_{IN} + 1.826 \text{ (kj / Nm}^3 \text{ C)}$$

$$q_{02} = 0.000148 \text{ T x T}_{IN} + 1.33 \text{ (kj/Nm}^3\text{C)}$$

$$q_{N2} = 0.000128 \text{ T x T}_{IN} + 1.271 \text{ (kj/Nm}^3\text{C)} \text{ (kcal/min)}-----(6)$$

Whereas:

go: Gas density at 0°C and 1 atm. pressure. (kg/Nm³)

g: Gas density at actual state. (kg/m³)

V: Gas velocity at actual state. (m/s)

Note: Measurement was done using a Pitot tube, of coefficient 0.845.

Four: Gas flow rate at C/C-outlet. (Nm³/min)

Note: Duct diameter is 1840 mm.

F_{IN}: Estimated gas flow rate at C/C-inlet. (Nm₃/min)

q_N: Estimated gas heat content at C/C-inlet. (kcal/min)

DP_{OUT}: Gas dynamic pressure at C/C-outlet (mmAq)

SPour: Gas static pressure at C/C-outlet. (mmAq)

P_{Atm}: Atmospheric pressure. (hPa)

Tour: Gas temperature at C/C-outlet. (C)

CO_{OUT}: CO content of exhaust gas at C/C-outlet. (%)

CO_{20UI}: CO₂ content of exhaust gas at C/C-outlet. (%)

O_{2OUT}: O₂ content of exhaust gas at C/C-outlet. (%)

N_{20UI}: N₂ content of exhaust gas at C/C-outlet. (%)

 T_{IN} : Gas temperature at C.C. outlet. (C)

CO_{IN}: CO content of exhaust gas at C/C-inlet. (%)

CO_{2N}: CO₂ content of exhaust gas at C/C-inlet. (%)

O2IN: O2 content of exhaust gas at C/C-inlet. (%)

N_{2IN}: N₂ content of exhaust gas at C/C-inlet. (%)

p: Pi, 3.14

H: Moisture (%)

12-14-4 Data Sheets for Measurement

Results of measurement of the exhaust gas, outlet temperature of cooling water and surface temperature of furnace body are automatically recorded in the personal computer. Other data are recorded on the data sheets.

Examples of data sheets for measurement are shown below:

Table 12-31 Operational Data: Electric Power

Date					 				
Heat No.	:			:		:		:	
Name							·		<u>.</u>

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Time	1		Vol	tage				c power			Remark	S	1
hr m				V)			(k'	Wh)		· · ·			
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Date	
Heat No.	
Name	

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hr	min	(Nn	n3)	(kg		(kg		Kg)
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Table 12-33 Operational Data: Oxy-Fuel Burner

Date						 				٠
Heat No.		 <u>:</u>		·		:	-		-	
Name			:						1	

	·		No. 1 Furn	ace burner		No. 2 Furna	ce burner
Fim	ie		O ₂	Oil		O ₂	Oil
11	min		(Nm3)	(kg)		(kg)	Kg)
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Table 12-34 Operational Data: Scrap

Date				
Heat No.				
Name	 		:	

1st bucket	2nd bucket	
Bucket No.	Bucket No.	
Time	Time	
Temperature (°C)	Temperature (°C)	·
Upper	Upper	· · · · · · · · · · · · · · · · · · ·
Middle	Middle	
Lower	Lower	
Raw materials (kg)	Raw materials (kg)	
Scrap	Scrap	
Pig iron	Pig iron	
Total	Total	

 	 	
3 rd bucket		·
Bucket No.	•	
Fime	-	
Temperature (°C)		
Upper	•	
Middle	-	
ower	-	
Raw materials (kg)	Raw materials (kg)	
Scrap	Scrap	
Pig iron	Pig iron	
l'otal	Total	

Table 12-35 Operational Data: Cooling Water

Date		,				
Heat No.	 		;			
Name			· · · · · · · · · · · · · · · · · · ·	٠.,		
				<u> </u>	<u> </u>	:
Time		4 14 <u>41</u>	:		:	
Flow rate (m³/hr)						
EBT	1 4					<u> </u>
Elbow	111			· 		
Roof	: 1			· · · · · · · · · · · · · · · · · · ·		
Shell-1	: : :-	·	· ·	<u>:</u>		
Shell-2	:					
Total			<u> </u>		· .	
Inlet temperature (°C	:)					

12-14-5 Measurement Results

Following data shall be recorded.

- 1 Profile of facilities
- 2 Long-run operation results of each five heats of same grade steel which are operated before and after sampled three consecutive heats for heat balance.
- 3 Concerned personnel for measurement
- 4 Operation results of sampled three heats for heat balance
- 5 Results of measurement of sampled three heats for heat balance

Examples are shown as follows:

Name of compan	y	IDC (IZMIR DEMIR CELIK
		SANAYI A. S.) Foca Celik Fabrikasi 35807
Address		Aliaga, IZMIR, TURKEY
Furnace manufac	<u></u>	NKK Corporation (Japan)
	Type of furnace	AC arc furnace
•	Charging method	Top charge
	Tapping method	EBT system
	Other facilities (bottom stirrer, bottom	Water sprayed electrode, Scrap
	bubbling, water sprayed electrode, scrap	preheater
	preheater, etc.)	
	Nominal capacity (ton)	
Molten steel	Bath diameter (mm)	
Monen seed	Bath area (m²)	
	Bath depth (mm)	
	Distance between sill level and roof (mm)	
Roof	Thickness (mm)	
	Radius (mm)	
	Diameter of electrode hole (mm)	
	Pitch circle diameter (mm)	
	Ratio of water cooling area (%)	
Shell	Inside diameter (mm)	
	Thickness (mm)	
	Height (mm)	
	Height from bottom to roof (mm)	
	Ratio of water cooling area (%)	
Hearth	Diameter (mm)	
	Thickness (mm)	
	Height of bank (mm)	
Working door	Width (mm)	
:	Height (mm)	
Tapping hole	Diameter (mm)	
(EBT)	Depth (mm)	

Transformer	Capacity (MVA) Primary voltage (kV)								
	Secondary voltage (V)								
<u> </u>	Connection								
Reactor	Capacity (kVA) Reactance (Ohm)	:							
Electrode	Diameter (mm)								
Oxygen injection	Type and number Capacity (Nm³/hr)	 .							
Burner	Type, number and capacity Kind of fuel and capacity (kg/hr)	 .							
Carbon injection	Capacity and number (kg/hr)								

Table 12-37 Summary of Long-Run Operation Results of Before and After Heat Measured

Date	Aug. 13 Aug. 14 Aug. 14 Aug. 14 Au	Aug. 14 Aug. 14 Aug. 14 Aug. 14		Aug. 14 Aug. 14 Aug. 14 Average	erage
Heat Mo.	965746 965747 965748 965749 96	965750 965754	965755 965756	965757 965758	٠,
1. Charged raw material (excluding hot					
1.1. 1st bucket (t) Scrap					
1.2. 2nd bucket (t)					٠
Scrap		1			
Pig iron					
1.3. 3rd bucket (t) Scrap					
2. Output (Production) (t)				:	
2.1. Good billet (t)					
2.2. Crop end, skull (t)					
3. Steel yield (%)					:
3.1. Molten steel yield					
3.2. Good billet yield					
4. Operation time (min.)					
4.1. Tap-to-tap time		1.			
4.2. Power on-to-power off time					
5. Consumption					
5.1. Electric power (kWh/t-output)					•
5.2. Fuel oil (kg/t-output)					
5.3. Oxygen gas (Nm3/t-output)					
Burner					
Lancing		- -			
5.4. Burnt lime (kg/heat)					
5.5. Carbon injection (kg/heat)				A SECTION ASSESSED.	
6. Composition and temperature of molten	U				
steel before tapping					
6.1. C(%)					
6.2. Si (%)					

	ature (deg. C) f hot heel (t)	on of billet (%)										
6.4. P (%) 6.5. S (%) 6.6. Cr (%)	6.6. Tempera 7. Weight of	8. Composition 8.1 C (%)	 8.3. Mn (%)	8.4 P (%)	8.5. S (%)	8.6. Cu	8.7. Ni	8.S. Cr	8.9. Mo	8.10. Sn	8.11 Al	

Note: 1) Heat No. 965757 is excluded for average of pig iron. 2) Heat No. 965755 is excluded for average of fuel oil.

Table 12-38 Concerned Personnel for Measurement

Date	August 14, 1996				
Heat No.	96	5751, 965752,	965753		
Company	Study team	EIE	IDC		
	- I. Kawakami - H. Tokano		- Suleyman Eldem - Necati Unsal		
Measuring person Flow rate, temperature and composition of exhaust gas Outlet temperature of cooling	- T. Kawai - S. Kinoshita - N. Honda	- (Bora Omurtay) - (Birgul Duman) - (Mehmet Sezer)			
water for furnace Electric power Oxygen gas by lancings Carbon injection			- Sibel Ozkan - Uguri Egeli		
Burnt lime Oxygen gas and fuel oil by 4 furnace/1 door burners Flow rate and inlet temperature of cooling			- Hikmet Yuzuak - Talip Bostanci		
water for furnace Temperature of scrap bucket Surrounding conditions	- (H. Tokano) - H. Tokano		- Duzgun Elitas		

Note: Bracket means co-working with specialists.



Date	Aug. 14	Aug. 14	Aug. 14
Heat No.	965751	965752	965753
1. Charged raw material (excluding hot heel) (t)			
1.1. 1st bucket (t) Scrap			
1.2: 2nd bucket (t)			
Scrap			
Pig iron		:	4
1.3. 3rd bucket (t) Scrap			
2. Output (Production) (t)			· · · · · · · · · · · · · · · · · · ·
2.1. Good billet (t)			
2.2. Crop end, skull (t)	•		
3. Steel yield (%)			
3.1. Molten steel yield			
3.2. Good billet yield			•
4. Operation time (min.)			
4.1. Tap-to-tap time			
4.2. Power-on-to-power-off time			
5. Unit consumption			
5.1. Electric power (kWh/output)		The second second	
5.2. Fuel oil (kg/t-output)			
5.3. Oxygen gas (Nm³/t-output)			- t
Burner	-		
Lancing			
5.4. Burnt lime (kg/t-output)		The second secon	
5.5. Carbon injection (kg/t-output)			
6. Composition and temperature of molten steel			
before tapping			
6.1. C(%)			
6.2. Si (%)	1		
6.3. Mn (%)		·	
6.4. P(%)			
6.5. S (%)			
6.6. Cr (%)			
6.6. Temperature (°C)			
7. Weight of hot heel (t)			
8. Composition of billet (%)			
8.1. C (%)			
8.2. Si (%)			
8.3. Mn (%)			
8.4. P(%)		$= \{ \sum_{i \in \mathcal{N}} \lambda_i - \lambda_i \geq \epsilon$	•
8.5 S (%)			
8.6. Cu (%)			
8.7. Ni (%)			

8.8. Cr (%) 8.9. Mo (%) 8.10. Sn (%) 8.11 Al (%)

Table 12-40 Measurement Results

;	1000	100	

Date	Aug. 14	Aug. 14 Aug. 14
Heat No.	965751	965752 965753
Amount of raw materials in 1st bucket (t)		
Amount of scrap in 2nd bucket (t)		
Amount of pig iron in 2nd bucket (t)		
Amount of raw materials in 2nd bucket (t)	1 1 × 1	
Amount of raw materials in 3rd bucket (t)		
Charged raw material (excluding hot heel) (1)		
Amount of scrap (t)		
Amount of pig iron (t)		
Output (excluding hot heel) (t)	:	
Consumption of electric power (kWh)		
Unit consumption of electric power, w ₁ (kWh/t-output)		
Hot heel of raw material (1)	:	
Unit weight of hot heel of raw material, m2 (kg/t-output)		
Temperature of hot heel, m ₂ (°C)	4 1	
Unit consumption of scrap, (kg/t-output)		·
Unit consumption of pig iron, (kg/t-output)		
Unit consumption of raw materials, m3 (kg/t-output)		
Temperature of top of 1st bucket (°C)		
Temperature of middle of 1st bucket (°C)		
Temperature of bottom of 1st bucket (°C)		
Mean temperature of 1st bucket (°C)		
Temperature of top of 2nd bucket (°C)	14 Table 1	
Temperature of middle of 2nd bucket (°C)		
Temperature of bottom of 2nd bucket (°C)		
Mean temperature of 2nd bucket (°C)		
Temperature of top of 3rd bucket (°C)		
Temperature of middle of 3rd bucket (°C)	1.	
Temperature of bottom of 3rd bucket (°C)		
Mean temperature of 3rd bucket (°C)		**
Mean temperature of raw materials after SPH (°C)	· .	
Consumption of fuel oil at No. 1 burner (kg)		
Consumption of fuel oil at No. 2 burner (kg)		
Consumption of fuel oil at No. 3 burner (kg)		
Consumption of fuel oil at No. 4 burner (kg)		
Consumption of fuel oil at door burner (kg)		
Consumption of fuel oil (kg)		
Unit consumption of fuel oil, m4 (kg/t-output)		
Mean CO ₂ content in exhaust gas at elbow (CO ₂) (%)		
Mean CO content in exhaust gas at elbow (CO) (%)	 	·
C content of molten steel before tapping (%)		
Si content of molten steel before tapping (%)		

		:
Mn content of molten steel before tapping (%)		:
P content of molten steel before tapping (%)		
Cr content of molten steel before tapping (%)		
Al content of molten steel before tapping (%)		
CaO content of slag (%)		
Consumption of burnt lime (kg)		
Amount of slag (kg)		
Unit weight of slag, m11 (kg/t-output)		
FeO content of slag (FeO) (%)		_
Fe ₂ O ₃ content of slag (Fe ₂ O ₃) (%)		
Consumption of carbon injection (kg)		_
Unit consumption of carbon injection (kg/t-output)		
P_2O_3 content of slag (P_2O_3) (%)		
SiO ₂ content of slag (SiO ₂) (%)		
Temperature of molten steel before tapping (°C)		
Weight of hot heel of molten steel (t)		-
Unit weight of hot heel of molten steel (kg/t-output)		
Power-on-to-power-off time (hr)		-
		_
Flow rate of cooling water for roof (m³/hr)		
Average quantity of cooling water of roof, m14b (kg/t-output	,	
Mean outlet temperature of cooling water for roof (°C)		
Mean outlet temperature of cooling water for roof-1 (°C)		
Mean outlet temperature of cooling water for roof-2 (°C)		
Inlet temperature of cooling water for roof (°C)		_
Flow rate of cooling water for EBT (m³/hr)	· ·	
Average quantity of cooling water for EBT, m _{14c} (kg/t-output	,	
Mean outlet temperature of cooling water for EBT (°C)		
Inlet temperature of cooling water for EBT (°C) Flow rate of cooling water for elbow (m³/hr)		_
	it).	
Average quantity of cooling water for elbow, m ₁₄₃ (kg/t-outpu		
Mean outlet temperature of cooling water for elbow (°C) Inlet temperature of cooling water for elbow (°C)		
Inlet temperature of cooling water for elbow (°C) Flow rate of cooling water for shell-1 (m³/hr)		
Average quantity of cooling water for shell-1, m ₁₄₄ (kg/t-outp	art\	
Mean outlet temperature of cooling water for shell-1 (°C)	uij	
Inlet temperature of cooling water for shell-1 (°C)		
Flow rate of cooling water for shell-2 (m³/hr)		-
Average quantity of cooling water for shell-2, m ₁₄₄ (kg/t-outp	ort).	
Mean outlet temperature of cooling water for shell-2 (°C)		
Inlet temperature of cooling water for shell-2 (°C)		
Indoor temperature (°C)		
Surface area of roof (m ²)		
Mean surface temperature of roof (°C) Surface area of shell (m²)		
Mean surface temperature of shell-1 (°C)		

Mean surface temperature of shell-2 (°C)	
Surface area of furnace bottom (m²)	
Mean Surface temperature of furnace bottom (°C)	
Heat in average flow of exhaust gas (1,000 kcal/min)	
Consumption of limestone, m ₁₂ (kg/t-output)	