The Environmental Assessment Sourcebook (the EA Sourcebook) was published by the World Bank during its fiscal year 1991-92. It is expressed in the EA Sourcebook that the Bank's environmental assessments emphasize identifying environmental issues early in the project cycle; designing environmental improvements into projects; and avoiding, or mitigating or compensating for adverse impacts. Figure 12.5-2 shows the EA activities in a project cycle, which are required by the World Bank.

The EA Sourcebook also provided guidelines for environmental assessment on each sectorial basis, in which mitigating measures are suggested for various potential negative impacts. In addition to the mitigation measures, the EA Sourcebook also made recommendations as to what a Borrower should do, such as project management, training and environmental monitoring.

To understand the details of the requirements of the World Bank, it is recommended to study the contents of the EA Sourcebook and its related documents at initial stage of planning a project, if the World Bank's loan is intended.

12.5.3 Environmental Issues and Protection Measures

In this section, some typical environmental issues of thermal power and hydropower plant projects are discussed and concerned protection measures are described.

(1) Regarding thermal power projects

As every one knows, releases of fly ash, sulfur oxide (SOx) and nitrogen oxide (NOx) to the atmosphere are the key issues of thermal power plants. To limit the releases, permissible emission standards are set force and various flue gas treatment systems are being used in many countries. In addition, environmental standards of ambient air are also stipulated to protect against air pollution. As an example, the standards and flue gas treatment systems being adopted in Japan are described below for references.

(a) Environmental and permissible emission standards in Japan

Attached Table 12.5-6 shows the environmental quality standard for ambient air set force in Japan to protect the public health, and conserve the living air environment, and serve as the target to be achieved by the environmental administrations. On the other hand, permissible emission standard is stipulated to control the exhaust gas emissions from each facility, so that air pollution can be controlled.

In case of permissible emission standard, a term called "Soot and Smoke" is used to mean the following substances:

- 1) Sulfur oxide (SOx) generated as a result of the combustion of fuels
- 2) Soot and smoke generated as a result of the combustion of fuels
- 3) Soot and smoke generated as a result of the use of electrical heat as a heat source
- 4) Cadmium and other harmful substances generated as a result of the combustion, synthesis, decomposition and other treatment of matters (excluding mechanical treatment)

Attached Table 12.5-7 shows the overall table of maximum permissible standards of the soot and smoke. The following tables are also attached to show some details of the standards:

- Table 12.5-8 Regulation on Sulfur Oxides Emission (K Value)
- Table 12.5-9 Emission Standard for Nitrogen Oxides (abstract)
- Table 12,5-10 Examples of the Emission Standard for Soot and Dust

As shown in Table 12.5-8, the permissible emission standard for sulfur oxides is defined by using K values which are the factors to be used in the following equation:

$$q = K \times 10^{-3} x He^2$$

q = the hourly volume of sulfur oxides emitted (in unit of Nm³/h)

k = the factor as defined by the Table 12.5-8, which is classified into 16 areas

He = effective height of the stack emitting flue gas (m)

Therefore, the emission standard of sulfur oxides is different from one area to the other. This is called "K-value regulation". The special standard is applicable to some areas as shown for new projects only.

(b) Flue gas treatment systems

1) Dust collector system

The soot and smoke presented in the form of particles is called aerosol. Dust collectors are used to separate and collect particulates from the flue gas. There are several types of dust collectors as below:

- Gravitational dust collector
- Inertial dust collector
- Centrifugal dust collector
- Scrubbing dust collector
- Filtering dust collector
- Electrostatic precipitator

For thermal power plants, the ESP is commonly used. ESP with high performance can remove about 80% of unburnt carbon from oil-fired thermal power and about 99% of fly ash from coal-fired plants. The installation cost of such high performance ESP is about 3% of the total plant construction cost. Others like cyclone type centrifugal dust collector or bag filter type dust collector is also common, but their efficiencies are quite sensitive to various parameters, such as flow velocity, particle size, filter mesh and so on. To achieve higher performance of a bag filter type dust collector, electrostatic bag filter has also been developed, by which the particles with about 0.1 micron diameter can be removed more efficiently. Selection of the dust collectors will depend on manufacturer's recommendations and the cost allocations.

2) Flue gas desulfurization (FGD) system

There are mainly two types of FGD processes, i.e. wet type and dry type processes. There are various processes developed as wet type or dry type FGD system, respectively as below.

- a) Wet type FGD process
 - Lime/limestone gypsum process
 - Magnesium hydroxide desulfurization process
 - Basic aluminum sulfate desulfurization process
 - Sodium sulfite desulfurization process
- b) Dry type FGD system
 - Activated carbon absorption process
 - Metal oxide absorption process

As far as thermal power plants are concerned, lime/limestone gypsum process has been most commonly used in Japan, especially for coal-fired thermal power plants. Recently, dry type activated carbon absorption process is being introduced to new projects. Brief discussions on these two processes are provided below.

a) Lime/limestone gypsum process

Main chemical reactions of the process are as follows:

$$\begin{array}{cccccccc} Ca(OH)_2 + SO_2 + Aq & \rightarrow & CaSO_3 \bullet 1/2 \ H_2O + Aq \\ CaCO_3 + SO_2 + Aq & \rightarrow & CaSO_3 \bullet 1/2 \ H_2O + CO_2 + Aq \\ CaSO_3 \bullet 1/2 \ H_2O + 1/2 \ O_2 + Aq & \rightarrow & CaSO_4 \bullet 2H_2O + Aq \end{array}$$

or

$$\begin{array}{cccc} \text{CaSO}_3 \bullet 1/2 \text{ H}_2\text{O} + \text{SO}_2 + \text{Aq} & \rightarrow & \text{Ca(HSO}_3)_2 + \text{Aq} \\ \text{Ca(HSO}_3)_2 + 1/2 \text{ O}_2 + \text{Aq} & \rightarrow & \text{CaSO}_4 \bullet 2\text{H}_2\text{O} + \text{H}_2\text{SO}_4 \end{array}$$

Where, Aq means water. The chemical reaction process is quite complicated and its details are not still very clear.

The advantage of lime/limestone gypsum process is that lime and limestone are available everywhere in most countries, and sulfur is recovered as useful by-products such as gypsum, sodium sulfate, sulfuric acid, elemental sulfur, etc. Disadvantage of lime/limestone gypsum process is that a lot of water is used and that large space for waste water treatment system will be needed. Scaling within the absorption tower and piping walls caused by crystallization of gypsum (CaSO₄ • 2H₂O) was used to be a key problem of this process. However, this problem was resolved by manufacturers in Japan and therefore the process became a major SOx removal system.

Attached Figure 12.5-3 shows a schematic flow diagram of a limestone gypsum process system. To satisfy specific applications, many variations can be made to the system, such as adding magnesium sulfite to obtain appropriate low pH value in the process, and depending on what by-product will be produced, etc. Figure 12.5-4 shows the picture of an EPDC's coal-fired thermal power plant (Isogo Thermal Power Plant, 265 MWe), which is equipped with the limestone-gypsum desulfurization system.

In general, installation cost of a lime/limestone gypsum process being adopted in Japan is quite high and influenced by site characteristics, it reaches about 10% of the total plant construction cost. Therefore, the process being used in Japan may need to be redesigned and modified into some simplified system, if it would be applied in developing countries.

b) Activated carbon absorption process (dry type)

Activated carbon acts as a catalyst in the process, through which SO₂ is converted into SO₃ by combining with oxygen, and then into H₂SO₄ by reacting with steam. The reaction temperature needed is as high as 100 to 130°C, and therefore its outlet gas will not need to be heated again for guiding to the stack. The advantage of this process is that it is a dry type and no waste water treatment will be required. However, it is a solid-gas reaction and that it needs large volume for the reactor. Moreover, the price of activated carbon will be higher compared with that of limestone. Therefore, detailed tradeoff among various factors will be needed when making selection of a desulfurization system. EPDC's Isogo Thermal Power Plant will be upgraded in near future. The upgraded plant will utilize the activated carbon absorption system, which will be integrated into a combined system of having both functions of desulfurization and NOx removal. For the purpose of NOx removal, ammonia will be added to react with NOx. Again, activated carbon will also act as a catalyst for this reaction.

Besides the FGD system, fluidized bed combustion (FBC) method has also been developed to remove SOx in boiler. SO₂ is removed by adding powdered limestone into boiler bed. Usually, 60% to 80% removal of SOx can be achieved.

3) Flue gas NOx removal system

NOx is generated from oxidation of nitrogen included in fuel itself and in air. The amount of NOx will be rapidly increased when burning temperature becomes higher than 800°C. Improvement of burning method, use of fuel with low nitrogen content and so forth can reduce 5% to 10% of NOx in flue gas. The remained NOx in flue gas has to be removed by flue gas NOx removal system.

The method of "catalytic reduction of NOx with ammonia" is considered most effective and is most commonly used in thermal power plants. The chemical reactions in this process are as below:

```
6NO + 4NH_3 \rightarrow 5N_2 + 6H_2O

6NO_2 + 8NH_3 \rightarrow 7N_2 + 12H_2O

4NO + 4NH_3 + O_2 \rightarrow 4N_2 + 6H_2O
```

There are various base metal catalysts, such as V₂O₅, CuO, Fe₂O₃, MnO₂ and Cr₂O₃, can be used for this purpose. Activated carbon is also useful as the catalyst. Actual application and selection of a catalyst will depend on the fuel and other factors of a power plant. As already indicated in the above, EPDC's Isogo Thermal Power Plant will install this process, using activated carbon as the catalyst, in its upgrade coal-fired thermal power plant in future.

Using the catalytic reduction of NOx with ammonia method, about 80% of NOx in flue gas can be removed. For more details, it is recommended to consult with the system manufacturers/suppliers.

(2) Regarding hydropower projects

There are various potential environmental issues which may be incurred in connection with hydropower developments. Use of agriculture and forest lands, potential impact to flora and fauna in project area, resettlement of people, potential conflict with the water uses needed by the people and industrial activities in downstream areas, eutrophication of the reservoir water and so on are all to be considered and evaluated. Necessary measures will have to be planned and taken for potential impacts. The extent of each potential environmental impact has to be investigated through EIA.

Based on the case studies made on Da River and Dong Nai River basins, as well as the field survey made during the site visits in the country, a few key existing environmental issues are discussed and some recommendations are provided as below.

(a) Resettlement issue

The case study report prepared on Da River Basin has described in details the resettlement issue of Hoa Binh hydropower plant. Discussions are made in the report regarding the causes of the existing problems and the further actions to resolve the problems being planned by the concerned governmental organizations. From the contents of the case study report, it is understood that the concerned governmental organizations are aware of the importance of the unresolved issue and preparing plans to resolve the problems. It is desirable that the following points are to be considered in the course of the problem resolution and for other development projects in future:

- Prepare an integrated program to cover the whole remaining issues and planned resolution measures, if such program has not yet been established. Also to organize a steering committee which is composed of the representatives from concerned governmental organizations, project owner and local authorities. The steering committee will be responsible for the whole program implementation. Enough budget shall be allocated for the program implementation. It is recommended that PC1 will take the key role and final responsibility for this issue.
- Taking the resettlement issue of Hoa Binh hydropower plant as the important lessons to be learned for future projects.
- It should be noted that resettlement program is one of key issues to be covered in the EIA of a project. A good and realistic resettlement program shall be established and

clearly described in an EIA. The program shall also cover the follow-up activities to monitor the livelihood of resettled habitants. Power companies shall be responsible for establishment of such a program. It is noted that the World Bank and other ODA donor countries are all paying much attentions to the resettlement issue of a project plan which will receive funding assistance from them.

(b) Eutrophication of reservoir water

Because of the inflow of large amount of substances having nitrogen and phosphorus from upstreams of a reservoir, water quality of the reservoir could degraded due to breeding a lot of aquatic organisms, such as plankton. BOD and COD values will gradually be increased. It was reported that water quality of the Tri An reservoir is getting degraded. In spite of not having water quality data available at the time of the site visit, this issue is usually expected. The issue of organic fouling to plant components is also an issue which is related with the water quality. However, key problem of the eutrophication is that no more fishes could survive at the ultimately degraded condition and offensive odor would be generated by the degraded water. If such condition would be reached, discharged water will raise severe negative impact to downstream areas.

This is an issue which can not easily be resolved and is also happening in industrialized countries. Many nature lakes are also suffering from this problem.

The countermeasures which can be taken to decelerate the progress of eutrophication are as below:

- Waste water treatment at the sources of upstream area
- Cut out the trees and clean up other organic substances existing in the reservoir area before water filling
- Forced circulation of reservoir water, if possible
- Perform periodic plant maintenance to clean up the plant components

(c) Water reduction area issue

Water reduction area will usually generated between the water intake point to the water discharge point of a hydropower station. If the distance between the two points will be long, significant negative impacts may occur in the water reduction area, including those on nature environment and agriculture. Figure 12.5-5 and Figure 12.5-6 show the pictures of the scenery of the water reduction area taken at the downstream side of the spillway of Tri An hydropower plant. These pictures were taken in July 1994, which is the beginning of raining season in this area. It can be seen that no any flowing water is existing in the areas and it was said that such water reduction section has about 7 km long.

To understand the extent of possible negative environmental impacts occurring in this section, it is recommended to perform an environmental survey covering both of nature and social environments. If it would be found that some significant negative impacts are existing or occurring, a certain amount of water discharge should be considered to mitigate the impacts.

In general, a certain amount of water should be discharged from spillway to protect the environment in such a water reduction area. The amount of the discharge water may be up to

about 3% power reduction of a hydropower station. This is the current practice being recommended in industrialized countries.

Table 12.5-1 Checklist for Preliminary Environmental Study on Dam Construction Projects (JICA)

(page 1/2)

		Environπ	nental	Factor	Evaluation	Basis
	Populat	ion	1	Change of population in the region (including racial minority problems)		
			2	Resettlement (including racial minority prob- lems)		
			3	Agriculture and forestry		
	T - J		4	Fisheries		
Social	Industry		5	Secondary industry (including mining, mineral resources)		
Envi- ron-			6	Tertiary industry (including tourism, recreation)		
ment	Commu	nications	7	Regional disruption (including racial minority problems)		
	_		8	Impact on land transportation		
	Transpo	rtation	9	Impact on water transportation		
	Water a	eas and their	10	Impact on water and fishing rights		
	Sanitation		11	Water-born diseases and their spread		
			12	Deterioration of sanitation during work		
	Landscape		13	Deterioration of landscape		
	Cultural assets, etc.		14	Impact on cultural assets		
		Geological phenomena	15	Induction of earthquakes		
			16	Slope collapse		
Natural Envi-	Litho-	Topography	17.	Sedimentation in the backwater section		
ron- ment	sphere	10000	18	Impact on downstream waterways		
·			19	Impact on coastal areas		· . ·
		Soil Condition	20	Soil erosion		
· ·			21	Soil contamination		

Table 12.5-1 Checklist for Preliminary Environmental Study on Dam Construction Projects (JICA)

(continued)

(page 2/2)

		Environmental	Facto	r	Evaluation	Basis
			22	Inter-basin diversion		
	Hy- dro-	Water phenomena	23	Impact on the groundwater		
	sphere		24	Change of flow regime		
			25	Change in water temperature		
		Water condition	26	Eutrophication		
Natural			27	Turbidity		
Envi- ron- ment		Bottom condition	28	Change in composition of bottom		
		Flora	29	Impact on flora		
	Bio- sphere	Fauna	30	Impact on fauna		
		Aquatic organisms	31	Impact on aquatic organisms		
		Ecosystem	32	Disruption of ecosystem		
			33	Air pollution		
	At- mos-	Air	34	Changes in micro-climate		
	phere	Offensive odors	35	Offensive odors		
		Noise, vibration	36	Noise and vibration		

Note 1: Evaluation Codes

- A: Great impact
- B: Moderate impact
- C: Little impact
- D: Unclear (Need for further study. It may so happen that the impact becomes clear as the survey progresses.)
- X: No impact and negligible impact
- Note 2: When evaluating items, refer to the corresponding sheet of the explanatory notes
- Note 3: Except in very large-scale dam projects, the induction of earthquakes is extremely rare. Furthermore, this evaluation is difficult in a feasibility study, so judgement should be made as carefully as possible.

Table 12.5-2 Study Items of a Full-scale EIA on Dam Construction Projects (JICA)

1. Outline of dam construction project Project name Location of the project 1.2 Objective and necessity of the project Reasons for the selected project site 2. Details of the project 2.1 Utilization plan 2.2 Work plan and schedule 3. Environmental conditions of the region and project site area 3.1 Social environment Population Customs and cultures Industries Transportation network Land utilization Water area and utilization Public health and hygiene Historical and culture assets Landscape (10)Specific regulations on the above items 3.2 Nature environment Geosphere (topography, geography, soil, sedimentation) Aquasphere (flow regime, water quality, sediment condition) Atmosphere (weather, air quality, offensive odrs, noise, vibration) Biology (flora, fauna, aquatic organisms, ecology) 4. Establishment of key environmental factors during costruction and operation 4.1 Establishment of key environmental factors during construction4.2 Establishment of key environmental factors during operation 5. Predictions and evaluations of potential environmental impacts Process of predictions and evaluations, and environmental conservation goals 5.2 Predictions and evaluations during construction work Resettlement of residents (2) Customs and culture
(3) Water and sediment qualities
(4) Biology (flora, fauna, aquatic organisms, ecology) 5.2 Predictions and evaluations for the time of operation (1) Topography and soil (2) Flow regime, water quality, sediment quality
(3) Biology (flora, fauna, aquatic organisms, ecology)
(4) Landscape 6. Environmental conservation measures, environmental monitoring and control plans 7. Comparison of alternative proposals

Remarks: The contents of this table have been slightly rearranged from the original JICA table.

8. Overall evaluation

Table 12.5-3 Environmental Checklist for Hydropower Projects (GECF, Japan)

Action & Countermeasures Planned					
Problems					
Not Clear					
Mone					·
Sea		· .			
Ma jor					
Check Items	Deterioration of mater quality (including detrimental changes in water (emperature) in the dam reservoir and downstream		Effect of construction of the facility on the ecology. Effect on landscape	1. Effect of construction of the facility on the historical and cultural heritage 2. Effect on existing infrastructure 3. Relocation 4. Effect on traffic 5. Effect on ther downstream utilization 6. Occurrence of diseases, such as malaria, carried by insects or water	1. Effect on the environment during construction period 2. Environmental Monitoring
3	L. Get (In		٦ ٨		

Table 12.5-4 Environmental Checklist for Thermal Power Projects (OECF, Japan)

	12.	Haine Cas	Sus 1: No	X BOOK	10	۱	Problems		berion & Cou	in } 6 7 8 6 11 5 0 5	-		
כונטי ניונטי	1				Clear	-	C#13:00		Planned	Planned		Remarks	
Pollution through the canission of soot and dust. sulphur oxides, and attrost	a 1.												
	ombust lon			•									
2. Offensive odours 3. Effect of thermal efficient and land reclassition on annuit	one in					-					<u> </u>		
organisms. fisheries, and other water utilization	1 other				·			-					-
	from				 -								
5. Noise and vibration 6. Ground subsidence													
	1 of 2	···											
8. Effect of industrial waste								·					
Matural Environment facility on the ecolory 2. Effect on Indicate	a u								-				
				·	-								
		· ——•	········	·				,					
	-		· · · · · · · · · · · · · · · · · · ·										
		-	\dashv	1									
Ilyman Eavironment facility on the historical and cultural heritage 2. Effect on existing infrastructure 3. Effect on land-use	the it and it ucture		· · · · · · · · · · · · · · · · · · ·										
											72		
1. Effect on the environment during construction period 2. Environmental Monitoring	during								•				
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				:				:	٠.				

Table 12.5-5 Classification of Projects/Components for Environmental Assessments by the World Bank

BOX 1.1: CATEGORY A PROJECTS/COMPONENTS

The projects or components included in this list are likely to have adverse impacts that normally warrant classification in Category A.

- dams and reservoirs
- forestry and production projects
- industrial plants (large-scale)
- irrigation, drainage, and flood control (large-scale)
- land clearance and leveling mineral development (including oil and gas)
- port and harbor development
- reclamation and new land development
- resettlement and new land development
- river basin development
- thermal and hydropower development manufacture, fransportation, and use of pesticides and other hazardous and/or toxic materials."

Category A. in general, certain types of projects either have or are likely to have "adverse impacts that may be sensitive, irreversible, and diverse." Category A includes projects which have one or more of the following attributes that make the potential impacts "significant": direct pollutant discharges that are large enough to cause degradation of air, water or soil; large-scale physical disturbance of the site and/or surroundings; extraction, consumption, or conversion of substantial amounts of forest and other natural resources; measurable modification of hydrologic cycle; hazardous materials in more than incidental quantities; and involuntary displacement of people and other significant social disturbances.

BOX 1.2: CATEGORY B PROJECTS/COMPONENTS

The following projects and components may have environmental impacts for which more limited. analysis is appropriate.

- agro-Industries
- electrical transmission
- aquaculture and mariculture
- irrigation and drainage [small-scale]
- renewable energy
- rural electrification
- tourism
- rural water supply and sanitation
- watershed projects (management or rehabilita-
- rehabilitation, maintenance, and upgrading projects (small-scale)

Category B projects often differ from A projects of the same type only in scale. Large irrigation and drainage projects are usually Category A; however, smallscale projects of the same type may fall into Category B. Similarly, a 50-meter hydroclectric dam is clearly large in scale and will usually require Category A classification, while low-head power dams are normally Category B. Construction of a 50-km expressway would also require thorough environmental analysis (Category A) due to scale, while rural road rehabilitation will tend to raise only minor environmental issues (Category B).

Projects entailing rehabilitation, maintenance or upgrading rather than new construction will usually be in Category B. A project with any of these characteristics may have impacts, but they are less likely to be "significant" to the environment. However, each case must be judged on its own merits.

***BOX 1.3: CATEGORY C** PROJECTS/COMPONENTS

These projects are likely to have negligible or minimal environmental impacts. No environmental assessment or analysis is required.

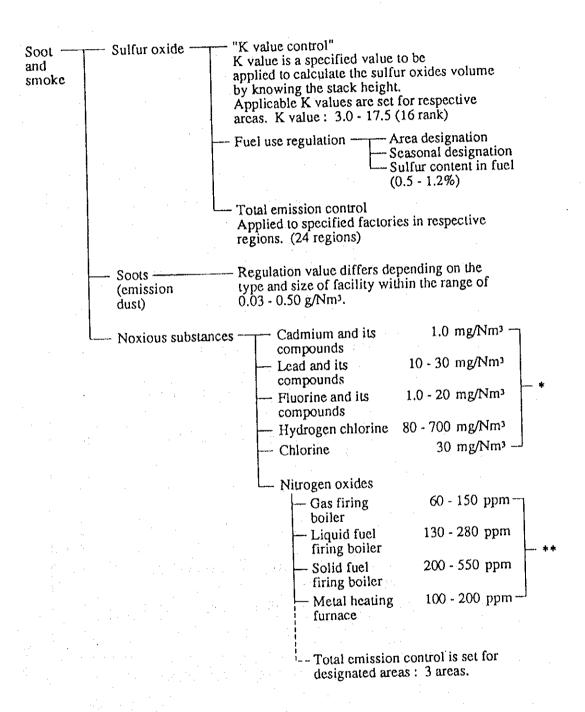
- education
- family planning
- health
- nutrition
- institution development
- technical assistance
- most human resource projects

Category C projects generally do not require an environmental analysis because they have negligible or minimal direct disturbance on the physical setting. However, not all Category C projects are entirely devoid of environmental impacts. For example, in a health project, the design may have to provide for disposal of medical wastes.

Table 12.5-6 Environmental Quality Standard for Ambient Air (Japan)

Substance Content Standard value School of measurement Remark Content Standard value Under 0.19 ppm Particulate Barbour average of one hour value Daily average of one hour value Daily average of one hour value Daily average of one hour value hour value hour value hour value Daily average of one hour value h					
Content Standard value One hour value Daily average of one hour value Daily average of one bour value Daily average of one between 0.04-0.06 ppm Daily average of one between 0.04-0.06 ppm Light absorption using neutral potassium iodine solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method.	Substance	Cond	ition	Method of measurement	Remarks
Daily average of one bour value Under 0.04 ppm bour value bour value Daily average of one Under 10 ppm Calizman reagent. Daily average of one Under 10 ppm Calizman coefficient being 0.84) bour value Daily average of one Daily average of one bour value coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method.		Content	Standard value		
baily average of one bour value 8-hour average of one bour value 8-hour average of one bour value Cone hour value Cone hour value Daily average of one bour value Daily average of one vith the range Daily average of one bour value Daily average of one vith the range Daily average	SO2	One hour value	Under 0.1 ppm	Electric conductivity method	
8-hour average of one Under 20 ppm Saltzman reagent. Daily average of one Under 10 ppm (Saltzman reagent. Daily average of one hour value Under 0.10 mg/m² which values hour value between 0.04-0.06 ppm (Saltzman coefficient being 0.84) Daily average of one Under 0.10 mg/m² which values having a linear relationship with the former method. Daily average of one with the range hour value between 0.04-0.06 solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method.	8	Daily average of one hour value	Under 0.04 ppm		
bour value One hour value Under 0.20 mg/m³ Weight and concentration Weight and concentration Weight and concentration Meight and concentration method by filter collection, or light-scattering method by which value bour value Daily average of one with the range hour value between 0.04-0.06 ppm Light absorption using neutral potassium iodine solution, or coulometric method. Under 0.06 ppm Light absorption using neutral potassium iodine solution, or coulometric method.		8-hour average of one hour value	Under 20 ppm	Absorption photometry using Saltzman reagent.	
ded One hour value Under 0.20 mg/m² Weight and concentration method by filter collection, or light-scattering method by hour value Daily average of one with the range hour value between 0.04-0.06 ppm botometric method. Daily average of one with the range hour value ppm or below solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method.		Daily average of one hour value	Under 10 ppm	(Saltzman coefficient being 0.84)	
Daily average of one hour value Daily average of one with the range hour value Daily average of one with the range hour value ppm or below Daily average of one hour value ppm or below Coulometric method. Light absorption using neutral potassium iodine solution, or coulometric method. Coulometric method.	Suspended	One hour value	Under 0.20 mg/m³	Weight and concentration method by filter collection, or	Suspended Particulate Matter means the particulate matter
Daily average of one with the range Absorption photometry using hour value between 0.04-0.06 neutral potassium iodine ppm or below solution, or coulometric method. Under 0.06 ppm Light absorption using neutral potassium iodine solution, or coulometric method.	Matter	Daily average of one hour value	Under 0.10 mg/m³	light-scattering method by which values having a linear relationship with the former method.	suspended in air whose diameter is under 10 µm.
coulometric method.	NO,	Daily average of one hour value	with the range between 0.04-0.06 ppm or below	Absorption photometry using neutral potassium iodine solution, or coulometric method.	
	Photo-chemical oxidant	One hour value	Under 0.06 ppm	Light absorption using neutral potassium iodine solution, or coulometric method.	Photo-chemical oxidants are oxidizing substances such as ozone and peroxyacetyl nitrate produced by photo-chemical reactions (only those capable of isolating todine from neutral potassium iodine, excluding nitrogen dioxide).

Table 12.5-7 Maximum Permissible Emission Standards for Soot and Smoke (an overall table, Japan)



* : Value varies by the type of facility

^{**:} Value varies by the size of facility and date of its installation

Table 12.5-8 K-value Regulation on Sulfur Oxides Emission (Japan)

a) General standards

		Area	K value
1	6 areas:	Central Tokyo, Yokohama-Kawasaki, Nagoya, Yokkaichi, Osaka-Sakai, Kobe-Amagasaki	3.0
2	21 areas:	Chiba, Fuji, Kyoto, Himeji, Mizushima, Kita-kyushu and others	3.5
.3.	1 area:	Sapporo	4.0
4	4 arcas:	Hitachi, Kashima and others	4.5
5	3 areas:	Toyama-Takaoka, Kure, Tohyo	5.0
6	9 areas:	Annaka, Niigata, Okayama, Shimonoseki and others	6.0
7	3 areas:	Tomakomai, Hachioji, Kasaoka	6.42
8	6 areas:	Sendai, Fukui, Hiroshima and others	7.0
9	8 areas:	Asahikawa, Utsunomiya, Mihara, Tokushima and others	8.0
10	8 areas:	Akita, Kanazawa, Otsu, Fukuoka, Nagasaki and others	8.76
11	6 areas:	Takasaki, Urawa, Narita, Naha and others	9.0
12	4 areas:	Shizuoka, Sasebo and others	10.0
13	15 areas:	Hakodate, Gifu, Takamatsu, Minamata and others	11.5
14	6 areas:	Mishima, Kurume and others	13.0
15	20 areas:	Aomori, Morioka, Yamagata, Nagano, Kagoshima and others	14.5
16	Others:		17.5

b) Special standards

	Area	K value
6 areas:	Central Tokyo, Osaka-Sakai, Yokohama-Kawasaki, Kobe-Amagasaki, Yokkaichi, Nagoya	1.17
8 areas:	Chiba, Fuji, Himeji, Mizushima, Kita-kyushu and others	1.75
14 areas:	Kashima, Toyama, Kyoto, Fukuyama, Ohmuta, Ohita and others	2.34

Note: Special standards are applied to newly constructed facilities only.

Table 12.5-9 Emission Standard for Nitrogen Oxides (Japan)

							alue (ppm)			
	Type of facility	Stack gas volume (Unit: 1,000 Nm³/h)	Date of instal- lation	before Aug.	after Aug. 10, 1973 before Dec. 9, 1975	after Dec. 10, 1975 before June 17, 1977	9, 1979	9, 1983	after Sep. 10, 1983 before Mar. 31, 1987	after Apr 1, 1987
	Gas firing	500 and above 100500 40100 1040 510 less than 5	5	1	30 50 Aug. 10, 198	4:150		00		
Boiler	Solid material firing (including coal)	700 and above 500 - 700 200 - 500 40 - 200 5 - 40 less than 5	6	400 420 450 from	350 380 Aug. 10, 198	4:480		80	350	250
	Liquid firing	500 and above 100 - 500 40 - 100 10 - 40 5 - 10 less than 5	4		180 30 250 Aug. 10, 198	4:250	150	30		
(excli	ring furnace uding pellet ing furnace)	100 and above 10-100 less than 10	15		260 270	300]	220	
Calci	nation furnace	,	10			200			200	
	ting furnace		14			250		<u> </u>	220	
	furnace		15			120	····	ļ	100	
Mota	l melting fumace		12			200		L	180	
Meta	l heating furnace	100 and above 40 -100 10 - 40 5 - 10 loss than 5	li		70 200	150			130 <u>J</u> 50 180	
Pedo	ocum heating fumace	40 and above 10 - 40 5 - 10 less than 5	6		170 180 200	150			130 150 180	
	ent calcination furnace luding wet types)	100 and above less than 100	10	:	480			250	350	
torie	ting furnace used for m a and fire bricks		18			450			400	
glass	ing furnace used for m see and glass fibers	anufacturing plate	15	<u> </u>		400			360 230	
Was	ng furnace te incinerator	40 and above less than 40	16			250 300	:	1 1	250	
	tinuous type) ic soid production facil		Os	 			200			
Cok	e oven luding Otto type)	100 and above	7		350	200	-		170	

Notes: 1. Reference to unit, the symbol "-" means "and above/less than": e. g. a-b means a and above/less than b.

2. NOx emission concentration shall be converted through the following equation

 $C = \frac{21 - O_0}{21 - O_s} \times Cs$ C : Nitrogen oxides emission concentration On : Oxygen concentration in flue gas (set values in the above table) Os : Actual oxygen concentration in flue gas Cs : Actual nitrogen oxides emission concentration

Examples of the Emission Standard for Soot and Dust (Japan) Table 12.5-10

(Unit: g/Nm³)

Name of facilit	facility	Ordin	Ordinary emission standard	sion sta	ndard	Speci	ial emiss	Special emission standard	ard	
(excerpt)	rpt)	Large scale	cale	Sm	Small scale	Large scale	ale	Small scale	scale	ő
Boilers	Gas	0.05	35		0.10	0.03	3	0.	0.05	5
	Oil	0.05	0.15	0.25	0.30	0.04	0.05	0.15	0.15	4
	Coal	0.10	0.	0.20	0.30	0.05	0	0.10	0.15	9
Gas generating furnace	ng furnace		0.	0.05			0	0.03		7
Blast furnace			0.	0.05			0	0.03		ဝိ
Cement kiln			0	0.10			ō	0.05		91
Waste incinerator	rator	0.15	5	-	0.50	0.08	~	0.	0.15	12
Coke oven			.0	0.15			0.10	10		7

Prefectures may, by decree, set more stringent standards. Notes: 1.

The gas emission rate of 40,000 Nm³/h is the criterion used for scale classification. However, heavy oil boilers and coal boilers are classified into four and three scales respectively. The criteria for the former ones are 200,000 Nm²/h, 40,000 Nm²/h, and 10,000 Nm³/h, 200,000 Nm³/h and 40,000 Nm³/h are for the latter ones.

The emission concentration shall be converted through the following equation. (except in the case of blast fumace).

 $C = \frac{21 - O_n}{21 - O_s} \times C_s$

: Soot and dust emission concentration

Oxygen concentration in flue gas (set values in the above table)

: Actual oxygen concentration in flue gas ర ర

: Actual soot and dust emission concentration

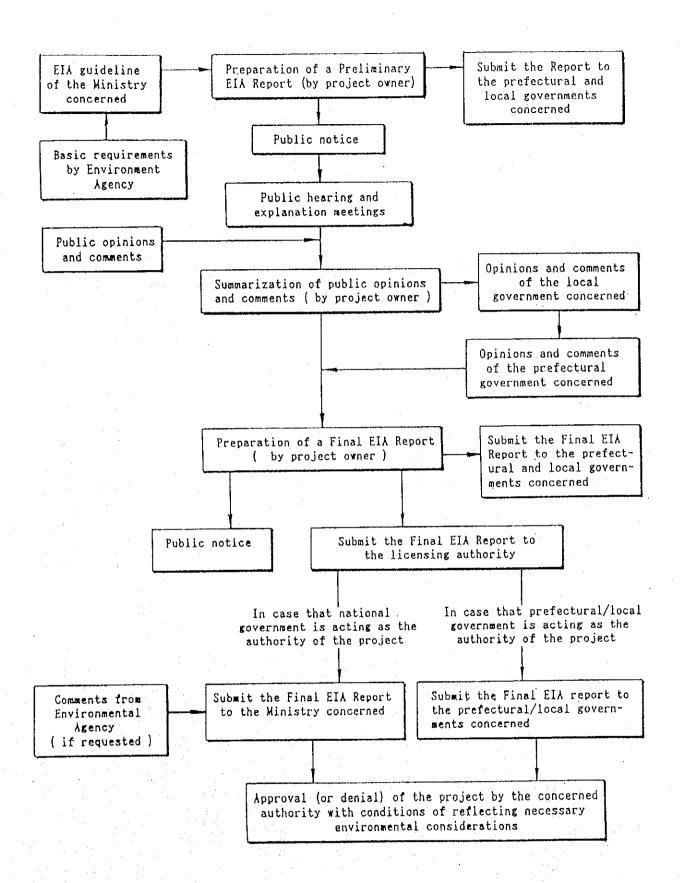


Figure 12.5-1 Licensing Procedure of Implementing an EIA in Japan (based on the Cabinet Decision of August 1984)

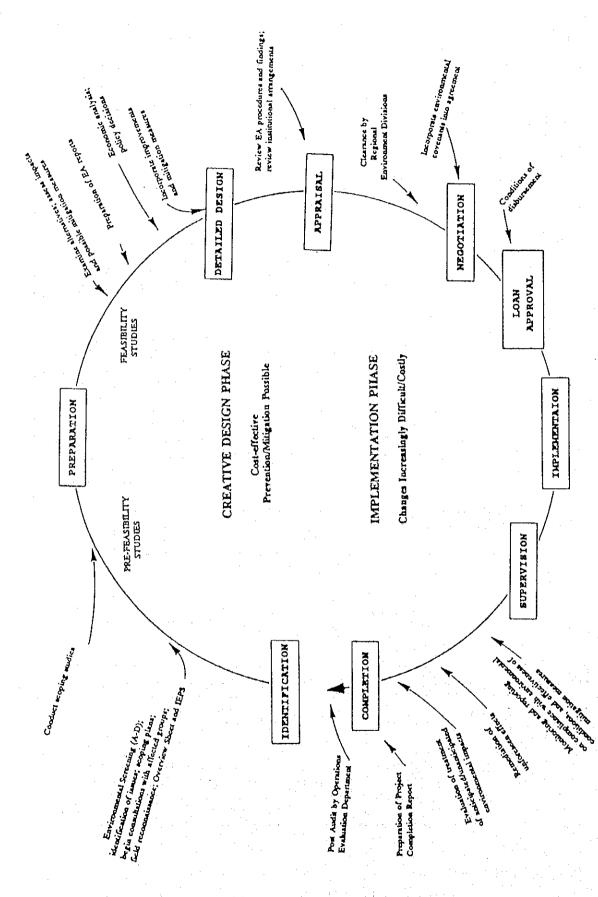
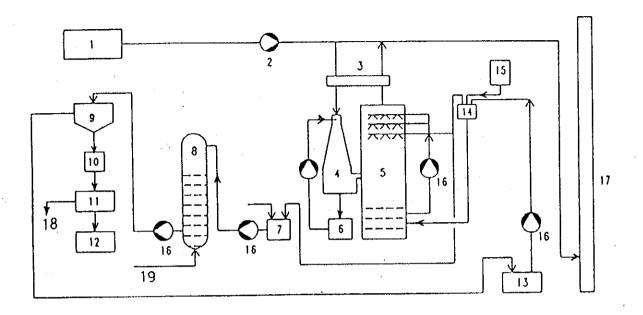


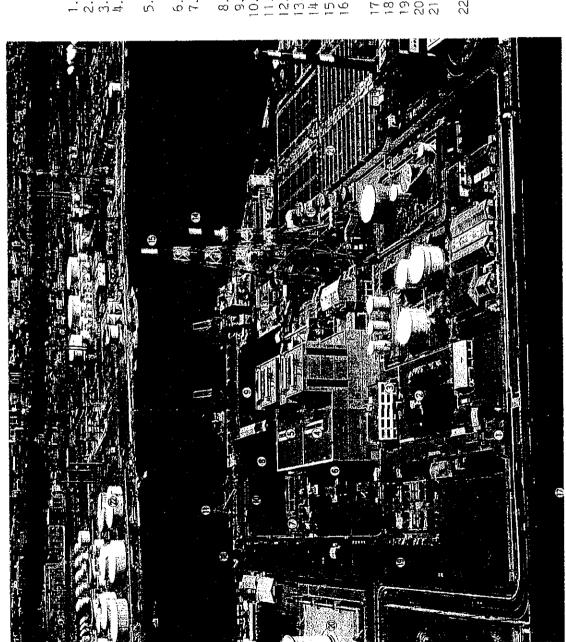
Figure 12.5-2 Envronmental Assessment Activities in a Project Cycle (Source : EA Sourcebook of the World Bank)



- 1. Flue gas source
- 2. Blower
- 3. Gas to gas heater
- 4. Pre-scrubber
- 5. Absorption tower
- 6. Recirculation tank
- 7. pH adjusting tank
- 8. Oxidation tower
- 9. Sedimenting tank
- 10. Gypsum slurry

- 11. Centrifugal separater
- 12. Disposal tank
- 13. Returned water
- 14. Limestone slurry
- 15. Limestone powder silo
- 16. Pump
- 17. Flue gas stack
- 18. Produced gypsum
- 19. Air

Figure 12.5-3 Schematic Flow Diagram of a Limestone-gypsum Process for Flue Gas Desulfurization



Power station gate

- Green hall
- Service Building Power plant unit No.1
- ower plant unit No.2
- Main transformer
- Intefrated waste water treatment system
 - Coal stock yard
- Coal unloading facility Gypsum storage house
- Cypsum loading facility
- Fly ash sedimentation pond Fertilizer plant
- Flue gas stack(Unit 2: 140m) Flue gas stack(Unit 1: 120m)
 - Flue gas desulfurization system (Two units)
 - Cooling water discharge Cooling water intake Green belt
- TEPCO LNG thermal power plant Sewerage treatment system
- Nisseki oil refinery plant of Yokohama City

Picture of an EPDC's Coal-fired Thermal Power Plant Figure 12.5-4

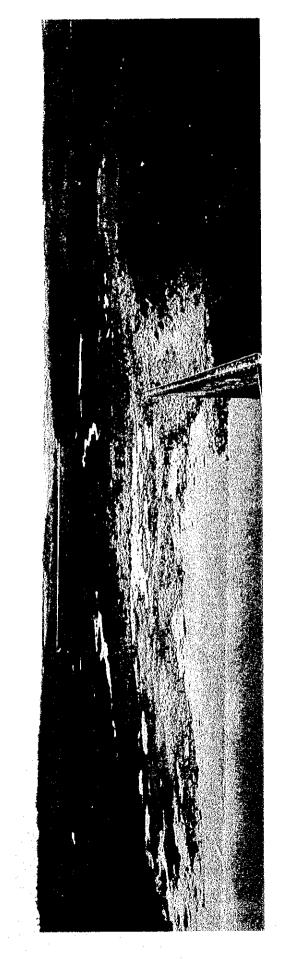


Figure 12.5-5 The Sceney of the Water Reduction Area at the Point Right after the Spillway of Tri An Hydropower Reservoir (July 1994)

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The Scenery of the Water Reduction Area at about 2 km Downstream Side of the Spillway of Tri An Hydropower Reservoir (July 1994) Figure 12.5-6

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