THE FIRST PROGRESS REPORT
FOR
THE STUDY RELATED TO
THE REGIONAL DEVELOPMENT PLAN
OF THE GREAT CARAJAS PROGRAM
OF
THE FEDERATIVE REPUBLIC OF BRAZIL

Vol. 3 NGRICULTURAL PRODUCTS PART II

NOVEMBER 1983

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Vol. 3 AGRICULTURAL PRODUCTS PART II

NOVEMBER 1983

JAPAN INTERNATIONAL COOPERATION AGENCY

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[3] ETHYL ALCOHOL

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A. GENERAL DESCRIPTION

I. Historical Background of Ethyl Alcohol Production

Ethyl alcohol has many useful properties, such as intoxicating, dissolving, reacting, germicidal and combustible properties. And these properties have been made use of since early days for application in alcoholic beverages, medicinals, industrial uses and as fuel. Ethyl alcohol has been produced in various countries of the world, since it can be made comparatively easily from agricultural products. The latest United Nations statistics list 43 alcohol producing countries.

While ethyl alcohol is such a popular substance, it has been treated as an object of government involvement in most countries, as a source of tax revenue when used in alcoholic beverages and as an energy source to secure handy liquid fuel which is domestically producible.

Production of ethyl alcohol was commenced with the fermentation process using farm products or saccharine and starchy substances, which are by-products of farming, as raw materials. Alcohol production has greatly been boosted between 1920 and around 1940 in various countries of the world for industrial and fuel uses from the need for national defense or for promotion of agriculture. As a result, outputs of ethyl alcohol rapidly expanded.

However, ethyl alcohol as fuel has practically been ignored after 1945, when international oil supplies became plentiful and oil prices remained low.

With regard to alcohol for industrial application, it began to be produced largely in the developed countries, when the synthesis of alcohol using ethylene as raw material was industrialized between 1930 and 1948. And on account of the raw material supply situation and agricultural policies followed in the respective countries, the use of both the synthesis and fermentation processes has continued to this day.

The 1973 oil crisis took place amid such a situation, and ethyl alcohol production under the fermentation method was limelighted again, and alcohol again began to attract attention for industrial application and as fuel.

Ethyl alcohol production technology have also been spotlighted at every encounter with oil crisis in recent years, and as production of alcohol for industrial use and for fuel increased, ethyl alcohol production technology also achieved progress. Further technological development efforts are also being made.

II. Properties of Ethyl Alcohol

More important physical and chemical properties of ethyl alcohol are shown in Table A-1 in comparison to those of methyl alcohol, gasoline, diesel oil and fuel oil. Data in this table are quoted from those by the American Petroleum Institute (API). For more detailed physical, chemical and pharmacological properties refer to Appendix Table 1.

III. Uses of Ethyl Alcohol

As stated above, ethyl alcohol has many highly useful properties, which make it usable for a large variety of purposes. As details of use will be presented in subsection C-I later, the historical development of uses will be briefly introduced here.

It is said that man has already known how to brew alcoholic drinks by the time he learned to leave records in writing. Ethyl alcohol, as a consequence, was first used for drinking purposes. And after distilled alcoholic beverages made from the drinking liquer became available, alcohol came to be valued as a substance for medicinal use. With further progress made in the distillation technology, dehydrated ethyl alcohol became available, and this reportedly began to be used as fuel and solvent for perfume refining. Meanwhile, use of ethyl alcohol in food and beverage and for hygienic purposes also expanded. And with rise of the chemical industry, use of alcohol as solvent, auxiliary agent and raw material commenced. Around 1910, 20,000 to 100,000 kl of ethyl alcohol was being used for industrial purposes in the leading countries of Europe and America.

After World War I, since 1920 in particular, the demand for fuel and industrial use ethyl alcohol expanded. From 1950 towards the first half of the 1960s, ethyl alcohol-based chemical industry using

Table A-1 Main Physical/Chemical Properties of Alcohol and Hydrocarbon Fuels

Property	Ethyl Alcohol	Methyl Alcohol	Gasoline	Diesel	Fuel Oil
Formula	сн ₃ сн ₂ он	сн 30н	c_4 to c_{12} Hydrocarbons	$c^{}_{14}$ to $c^{}_{19}$ Hydrocarbons	C ₂₀₊ Hydrocarbons
Molecular weight	46.1	32.0	100-105 avg.	240 avg.	
Composition (weight %)					
Carbon	52.2	12.5	88-58	80 - 80 - 80 - 80 - 80 - 80 - 80 - 80 -	85-87
Hydrocarbon Oxygen	13.1	12.5	12-15 Neg.	12-15 Neg.	10-11 Neg.
Specific gravity	0.79	0.79	0.72-0.78	0.83-0.88	0.88-0.98
Boiling temperature, °C	78	92	27-225	240-360	360+
Flash point, °C	e (1.0 4.1 W1	38	99
Autolgnition temperature, 'C	423	ند ۵ ۵	157		
Flammability limits (volume %)					
Lower	4. w.	ı	1.4	1. 1.	1
Higher	0.61	1	7.6	•	1 1
Octane Number (Research) (Motor)	106-111	106-115	79-98	f t	N.A. N.A.
Cetane Number	ιο •	N.A.	5-10	45-55	N.A.
Solubility in water	Infinite	Infinite	0	0	

Note : N.A. - Not Applicable Source: American Petroleum Institute (API)

ethyl alcohol as raw material began to develop mainly in Brazil and India, and commercial production of ethylene, polyethylene and ethylbenzene commenced.

In parallel with the above development, manufacture of synthetic chemical products based on petrochemical processes commenced in the United States from the 1940s. And in 1948, production of synthetic ethyl alcohol under direct hydration of ethylene commenced, and by the latter half of the 1960s the ethyl alcohol-based chemical industry using fermented ethyl alcohol as raw material has lost competitiveness in the international market.

In the fuel ethyl alcohol sector, on the other hand, use of ethyl alcohol as fuel almost ceased since 1945 as low-cost stable oil supplies became available.

The oil crisis in 1973 amid such a situation and the repeated price hikes in the subsequent years forced the countries of the world to reassess their energy policies, and since ethyl alcohol made under the fermentation process uses biomass as raw material, it has lately been attracting attention not only as an energy source but also as raw material for chemical products.

As described in the foregoing, uses of ethyl alcohol can be broadly classified into three main categories, namely, for beverage, as industrial material and for fuel. However, use of alcohol as beverage has been excluded from this report because of the aims of this Study. Uses of ethyl alcohol for industrial purposes and as fuel only have been taken up.

B. PRODUCTION

I. Product Specification

1. Purity of Product

In the specification applicable to ethyl alcohol, the specification for product purity is applicable both to industrial and fuel ethyl alcohol. For instance, the two types of ethyl alcohol include the hydrous ethyl alcohol with alcohol content of 94 to 95% by volume and the anhydrous ethyl alcohol with alcohol content of 99.4 to 99.6% by volume. However, some countries do not produce anhydrous alcohol.

2. Other Items of Specification

The industrial ethyl alcohol specifications of developed countries include, as other items specified, specific gravity, evaporation residue, free acid (as acetic acid) content, methyl alcohol content and the fusel oil content.

For reference, the industrial ethyl alcohol standard in selected countries are given in Appendix Table 2.

Meanwhile, specifications for fuel alcohol have not yet been established except for product purity and organic acid content. As ethyl alcohol for fuel is normally produced with top priority given to cost reduction, use of fuel ethyl alcohol for industrial purposes like making cosmetics may require further refining to a certain extent.

II. Raw Materials

1. Raw Material Production and Regional Distribution

There are two manufacturing processes of ethyl alcohol. One is fermentation process and the other is synthesis process, each using raw material widely different from the other.

The fermentation process consists of obtaining alcohol from fermentable sugars through fermentation. And as the starchy and fibrous substances are fermented after saccharification, the raw materials for the fermentation process may be classified as hereunder.

- Saccharine raw materials: Molasses, sugar cane, etc. - Starchy raw materials: Grains, potato, sweet potato,

etc.

- Cellulosic raw materials: Lumber, etc.

The synthesis process consists of producing alcohol by causing chemical reactions between ethylene from the petrochemical plant and water.

- Synthesis process materials: Ethylene (from petroleum or natural gas).

While raw materials cited in the foregoing are used extensively for making industrial ethyl alcohol, the production of fuel ethyl alcohol is naturally limited to the fermentation process, and the raw materials used in Brazil largely consist of molasses, sugar cane and cassava. On the other hand, corn is the principal material used in the United States.

In the following the production of principal raw materials and the regional distribution of such materials will be outlined.

1.1 Saccharine Raw Materials

1.1.1 Molasses

When sugar is manufactured using sugar canes and sugar beets as raw materials, molasses is obtained as a by product. As it contains 50 to 60% of sugar and is low priced, it is used for a variety of purposes; it is also used as raw materials for making both industrial and fuel-use ethyl alcohol. Molasses production and their regional distribution are given in Table B-1.

Table B-1 Molasses Production of the World

(1,000 MT)

July-June	1976/1977	1977/1978	1978/1979	1979/1980
EC.	(3,565)	(3,276)	(3,033)	(2,995)
Western Europe	5,077	4,638	4,380	4,174
Eastern Europe	4,371	5,227	5,226	4,557
North & Central America	6,743	6,971	7,153	6,393
South America	5,452	7,106	7,195	7,390
Africa	2,087	2,099	2,102	2,197
Asia	6,698	6,931	6,676	5,332
Oceania	728	746	683	750
World Total	31,156	33,718	33,415	30,803

Note:

- 1) Refined molasses included.
- 2) Figures for EC given in () are part of total for Western Europe.

Source: The third estimate by Licht

1.1.2 Sugar cane

Sugar cane has originally been used as a raw material for making sugar, but they began to attract attention as raw material for ethyl alcohol, when ethyl alcohol began to be used in large quantities as fuel. In Brazil sugar cane is already used as a raw material for fuel ethyl alcohol. Sugar cane production and regional distribution are shown in Table B-2.

In addition to sugar cane, saccharine raw materials include fruit juice and sweet sorgham.

1.2 Starchy Raw Materials

Any substances with high starch content may serve as raw material, but raw materials for making ethyl alcohol become limited when the yield, ease of production and economic efficiency are taken into consideration. Typical materials include corn and potato, and in Brazil cassava is also used for making fuel ethyl alcohol.

Table B-2 Sugar Cane Production of the World

	AREA HARV		1	TH COC	AILTO			CG/HA	PADDUCTI	<u>ри</u>	-	1000 HT
	1969-11	1414	1980	1981	1969-71	1979	1980	1981	1969-71	1979	1980	[981
WORL D	10747	13584	13249	13819	52833	55547	54328	56102	567817	754536	11,9800	175285
AFRICA	698	864	887	930	67394	70744	64568	69582	47036	61095	57293	6468
	15	15F	156	LSE	\$1602	29000	50000	27333	776 224	120F	300F	
ANGOL A CAHERDON	į ti	206	526	175	20394	36000 (5000	35909 15000	34118 15000	<u> </u>	155	15F	1
CAPE VERDE	l l	1 F 3 F	. 18 3F) F	55918	86000	88000	6,6667	22	215F	220F 203F	
CHAO:	20	5F	715	86	33899	24600 84203	29000 84060	28175 83575	1096	123f	87910	907
EGYPT	77	104	105€ \$E	109F	92300 [48197	165412	160976	150000	1097	1405	13206	
ETHICPLA GABON		• • • • • • • • • • • • • • • • • • • •	3	31	22928	53884	51339	51852 21111	13	1904	134*	140
GHAH A	6	9#	9* 3년	9F 3F	28712	21111 81491	21333	81481		\$ 20F.	(201	
GUINEA CUIN BISSAU	(3F 2F	· 26,	2F	56250	13333	12500	12500	5	20F ₁	25F	
IVORY COAST		10	21	26 41*	62878	80317 108492	75551 112767	76453	1645	3678F	45176	462
KENYA LIBERIA	26	34* 10F	40* 10F)	101	10000	14737	15000	15500	45	140F	150F 1388	15
HADAGASCAR	11	35 '	34	345	55741	100000	100000	100000	1153	1432 1120F	14805	
MALAKI	4	11F	15F) 3F)	1.7F 3F	74967 56738	77077	78571	17586	56	513*	5501	
MALI MAURITIUS	80	80	79	908	67552	79178	57674	68750 74419	5400	6313 1	375	550 321
MOROCCO,		50F	5 0 F	4F 55F	53716	13729 44000	93277 40000	35455	2767	2200F	2000F	195
HOZANSTQUE NIGER	52	4 !	46	48	27432	53414	52778	50000	3.9	189 815F	190F	
NIGERIA	12	165	16R	185	46306 50659	52581 66800	34375 66176	1. 55556 65556	356 2026	2405	2250	536
reunton Ryanda	40	36년 16	34F] 1F:	36F 16	20808	51667	49231	47143	5	315	321	
SENEGAL	- {	4	5*	6.6		84445	103140	100000		367*	473+	60
SEACHEFFER	3	3	7 1	35	126738	87063	63561	58571	436	261	420	41
SOMALIA SOUTH AFRICA	188	2224	2154	217F	77495	82833	65182	77086	14561	18412	14014 1500F	1672 240
SUQAN	13	26 22	765	30F	65458	66341 194014	58594 107722	80000 108333	1548	2280	2782	325
SHAZILANO TANZANIA	39	426	428	42F	29598	30476	31861	30643	1341	1280F	1339F 550F	
UGANDA	31	30F	29F: 48	. 30F	54055	17667 00000	18944 87500	18333 87500	1667	530F 360F	350F	35
UPPER VOLTA ZAIRE	11:	4F 14F	14F	157	58997	49265	43662	45426	519	6105	6205	65
AIBHAS	3	10	10	175	102160	88752 104209	91972 103121	£ 83333 £ 119206	1861	2555	920 2528	100 352
SIMENSKI	2698	25	25 2983	30* 3036	104338 57088	61236	54788	58535	154022	17878		17162
I G AMERICA INTIGUA	20,0				31033		1	ļ .	-59			
BAHAHA5	1	?F	7.6	8F	32857	30833	30270 76071	29737 54430	230 1375	222F	224F	860 251
BARBAQOS BELTZE	12	25	16 25	16* 25	66121 51685	66247 60053	41012	38804	622	989	[013	97
COSTA RICA	36	450	49.	5 i s	55891	58521	51684	49448	1005	2615 77311	. 2516 62374	252 6700
CUBA	1273	1313	1361	1 400F	4749 <u>3</u> 21615	58890 20000	45843 20000.	2000D	60467		45	
DOMINICA OOKINICAN RP	147	173*	180	1854	60770	57823	50309	62703	5986	10304	9056	1140
EL SALVADOR	31	37	34	28	53224 39290	86816 33330	33750	68501	1662	3214	2201 14F	191
GRENADA GUADELDUPE	28	23	21.	235	63982	55330	47509	36457	1785	. 🧓 1163 🖟	1007*	83
GUAT EMAL A	38	748	796	83*	71713	59153 - 38667	68811 37500	68511 37500	2692	5190F	54094 3000F	
HA I 7 I HONDURAS	75	75A 75A	80F	80F 90F	36186 27821	34667	33176	33333	1368	2690F;	20201	300
JAKAICA	58	4.7	49#	45*	70655	63082	58000	55124	1098	2965	2615	
MARTINIQUE MEXICO	518	538:	5 46 }	5 545	65352 64261	57983 64288	52083 66869	48636 65076	33271	771 ' 34587	250 36480	. 24 3546
ni caragua	32	42	32+	37*	58143	65029	76838	75072		54.84	2431*	
PANAMA PUERTO RICO	18	39	48 34 :	48F 32F	68609 72373	59408 53897	49868 59350	53250 59375	1242 4952	2076	2386 2028	255 190
ST KITTS ETC	3	4	36	3F	70291	100392	101515	96875	285	384	3350	31
SAINT LUCIA	1				4			: :			7	
SI VINCENT IRINIDAD TOB	35	390	34	28*	70354	10858	41154	42857	2481	1600*	1400	120
USA US.VIRGIN IS	238	297	291	305	89769	81173	82497	89802	21494	24069	24460	2107
SOUTH AMERIC	2477	3530	3616 °	1886	51.735	58028	59704	57230	128121	294853	215912	21924 [.]
	1 :								1	:	. 4	
ARGENTINA BOLIYTA	200	306 57	314	323 65*	51195 ; 37293 ;	46204 46349	. 54760. 46978	57701 50039	10213	14120 3120 :	17200	1526 325
BRAZIL	1703	2537	2695	2803	45926	54750	56769	54898	78460	138879	146065	15385
COLORS IA ECUADOR	841	282 103	290F	300F	54709 17689	87539 64019	90000	86333	6500	24700F 6599 ;	76100F	2590 665
FR GUIANA	***	. 103	103	105,	50056	32000	34000	34500	5	3,7	74	
GUYANA	50	37*	57⊭	581	81563		66316	68766	1081	4200 °	3790	
PARAGUAY PERU	35 56	35 - 54	37 : 49 :	409	35137 141926	36940 ! 130507 !	37511 113928	37805 104680	1215	1257 1034	1373 5598	195 416
SUR ENAME	3	2	2	3/-	77632	80343	15158	44000	713	164	146	16
UR UGUAY. YENE ZUE LA	62	10 71•	30 m	10*	38960 75025	31355 57143	49008 48750	55000 50000	145 4883	\$23 4400	5+9 5500	400
ASIA	4599	5935	5402	5633	47369	47836	47295	51548	217931	284207	255463	27035
AFGHANISTAN	3	4F	46	45	22000	16763	16250	16250	.55	64	654	
BANGLADESH	165	155	151	149	45847	55787	44155	44355	1551	6031	6676	459
BURMA China	43	50 . 622 !	42F 652F	4 3F 700F	33907 41306	36346	35474 49019	35835 47143	1458 19581	1812	1477 i 31978	152 330µ
Algni	2632	3088	2610	2648	48879	49114	47358	56844	124689	151655	120911	
INOUNES 14 IRAN	69	166	110	1774	148941	96355	100500	99208	10357	15995F	170857	1756
PAG] z	101. 5 <i>F</i>	1 SF	61	125145 (39197 (83333 : 46667 :	100000	100000	556 78	15004 2104	1400f: 260f	140

Table B-2 (cont'd.)

		AREA HAR	ŧŸ;		1000 HA	YIELD	<u>, , , , , , , , , , , , , , , , , , , </u>	i	(GZILA	PRODUCT	ION	<u> </u>	1000 HT
		1969-71	1979	1980	1981	1969-71	1979	[980	(981	1969-71	1979	1980	1981
	JAPAN	39	35	34	37	62313	65653	61982	66216	2407	2311	2095	2450
	KANPUCHEA OH LAO	6 1	2F		, 1	70833 1756	52273 8182	48429 8534	484 <i>62</i> 27369	397 8	}15F 9F 1F	130F 9	126F
	LEBANDN HALAYS EA	. 4	20	18	194	21750 30623	46269	33889	54737	122	930#	610F	850F
	NEPAL	14	201		217	17185	18497	18750	17868	237	370	384	379
	NARG					30303	27027	25641	24390	ı	l F	1 F	l F
	PAKISTAN	599	753	719	825	39799	36313	38271	39238	23836	27326	21498	32359 20450F
ĺ	PHILIPPINES SINGAPORE	. 384	4244	425	4 201	12421	47209	46738 6233	48690 6333	16272	2000 7F 2	1 9846F	2043QF
	SRI LANKA	4	78	51	58	42433:	50000	50000	55135	168	3256	265F	276F
	SYRIA]		53947	45000	50000	-	1			1 1 1
	THATLAND	144	480		4804	40653	42175	29952	38750	5856	20244	12460	18600*
	VIET HAM	1.5	84	110	aor	27799	40828	39854	48750	333	3446	4388	3900F
	YEMEN AR	2	: 2F	SL.	ŞF	2941	2941	2941	2941	5	5F	5 F	
	EUROPE	7	6	6	δ	72132	63910	66520	64000	486	374	371	384
	PORTUGAL	1	1	11	15	38058	18906	26923	26154	49	24	355	34F
	SPAIN	5	5		5.	80198	16616	78140	74468	437	350	336	3500
	OCEANIA	269	329	355	384	75065	16570	76907	75437	20220	25222	27321	28983
	AUSTRALIA	552	267	288	310	79185	79159	83141	81161	17607	21151	23948	25160
	FIJI	47	62	67	746	55554	65452	50149	51486 81818	2603	4056 3F	3360 3F	3810F
	FR POLYNESIA PAPUA H GUIN			i í	· (74227 60714	80645 60606	81250 60000	61176	6	1 OF	LOF	
	AOHAZ	4				14000	15000	15000	15000				. 10.
	DEV.PED H E	694	827	839	875	81356	80186	77346	82064	56465	66317	64889	71790
	N AMERICA	238	297	291	305	89769	51173	82497	88802	- 21404	24069	24460	27076
	₩ EUROPE	7	٥	6	6	72132	93910	88250	64000	196	374	371	384
	DCEINTA	222	192	208	310	19185 74906	79159 80485	83141 64747	81161 75502	17607	21151 20723	23948 16109	25160 19170
	OTH DEV.PED	553	257	249	254				1.				
	DEV.PING H E	9562	12048	11645	15165	51355	54260	53106	54800	491042	653714	618416	666470
	AFRICA	421	- 511	542	.574	55397	62961	60846	63555	24555	32193	32988	36491
	LAT AHERICA	4936	6153	6303	6562	52824	38435	56305	55442	280740	359568	354892	363798
	HEAR EAST	100	158	155	164	86041	11622	17352	80610	8616	12271 243611	12022	13207 249130
	FAR EAST OTH DV.PING	4058	5163 62	4577	4788 74	47936 55577	47573 65446	- 47003 50188	52038 51522	194518 2613	4071	3373	3823
	CENTR PLANNO	492	109	765	763	41314	16669	47691	47312	20310	34505	36495	37026
	i		, ,					ł }					37026
	ASTAN CPE	492	709	765	783	41314	48669	47691	47312	20310	34505	36495	
	DEA'BED WIF	674	827	8 39	875	81356	88108	17346	B2064	56465	66317	64889	71790
	DEV.PING ALL	10053	12757	12410	12944	50864	53950	52772	54347	511352	688219	654911	703496
,									ì				

^{*} Unofficial figures
F: Estimates by FAO

1.2.1 Corn

Large quantities of corn in the United States as raw material for fuel ethyl alcohol is well known. Corn production and regional distribution are shown in Table B-3.

1.2.2 Potato

Potato is cultivated widely in various parts of the world. In northern Europe, potato is an especially important root crop and raw material for making ethyl alcohol. Potato production and regional distribution are shown in Table B-4.

1.2.3 Cassava

Cassava is used principally as a raw material for starch, but lately it is attracting attention as raw material for alcohol. Cassava production and regional distribution are given in Table B-5.

1.2.4 Sweet potato

It has been used principally in Japan as one of the raw materials for ethyl alcohol, but only in small quantities lately. The production and regional distribution are given in Table B-6.

1.3 Cellulosic Raw Materials

A total of some 30 billion tons of cellulosic are estimated to be produced annually in the world. It is the largest reproducible resource on the earth. Today alcohol is produced by acid saccharification in some countries, but with regard to production of alcohol by bio-saccharification method, research work on the preparatory saccharification and fermentation processes with a view to improving the efficiency thereof is being undertaken.

1.4 Raw Materials for Synthesis Process

Raw materials used for obtaining ethyl alcohol by synthesis process include natural gas, naphtha and ethylene produced from associated gas of oil fields. Ethylene is a basic raw material for the petrochemical industry, since ethyl alcohol and various other petrochemical products are manufactured from it. Ethylene production and regional distribution are shown in Table B-7.

Table B-3 Corn Production of the World

:		AREA HA	RY		1000 HA	YIELD			KG/HA	PRODUCT	104		1000 HT
		1969-71	1979	1980	1981	1969-71	1979	1980	1981	1969-71	1979	1980	1981
	WORLD	114485	125796	128764	134024	2412	3326	3060	3370	283031	418357	394056	451 704
	AFRICA	18549	20019	21373	22583	1170	1203	1271	1455	21710	24974	27173	32860
	ALGERIA ANGOLÁ	540	1 600*	600*	1F 600F	953 864	1117 500	966 533	1000	467	1 300F	3 2 0 F	1F 250#
	BENIN	359	424	427*	427F	559	724	017	817	201	307	349*	349F
	BOTSWANA BURUNDI	112	21 125F	130F	50# 130#	309 1071	107	1000	1077	120	1405	130*	140+
11	CAMERDON CAPE VERDE	309	545 10F	54 OF	540F 10F	1151 431	148	907 467	926 300	355	408	490F	. 500F
1 1	CENT AFR REP	62	103	108	1.00F	711.	328	379 1400	400 1500	12	34 158	41 14F	40F
	CHAD CONOROS	3	10F	10F	LOF 5F	1943 1061	1500	1000	1000	4	.5	5F	3F
	CONGO	634	26 791	26F	27F 714*	3741	654 3714	4038	556 3782	2370	2938	3231	2700¢
	ETHIOPIA GABON	869	100) + 5F	1106*	1100F 6F	1071 1319	1066 1481	1034 1500	1000 1667	709	1067* 8F	1144* 9F	1100F
4.	GANBIA	12.54	105	12*	124	566	1000	1000	1075	4	10	126	1 34
	GHANA GULNEA	387	358 - 1	3404 50F	350F 55F	1078 1153	1061	1147	1145	417 68	380 7	3904 57F	4200 63F
	GUIN BISSAU IVORY COAST	333	56 584	400	5F 615F	628 773	800 471	400 467	80D 488	257	4F ≥75	2F 280	4. F 300F
	KENŸA	1383	LACOF	1120	1500	1485	1286	1579 957	L875	2060 93	1800	1768 106	2250 130*
	LESOTHO LIBYA	143	112	110	110*	651 1234	1112 1054	950	1182	1	125 1	1	. 1F
•	HADAGASCAR HALAWI	1039	116 1000F	128 (1100F	126* 1100*	1004 1025	1004	1000	1000	1066	12004	127 1100F	126# 1600#
	MALI	78	90F 9F	90F	90F 10F	866 580	944 556	367 571	889 600	67	85 5F	.33* 5F	80F 6F
	MAURITANIA Mauritius	1))		1F	1984	2502	2245	2500		1 1	1	25
11.	HOROCCO HOZAHBIQUE	363	416 620	411 600F	360± 550F	801 1003	150 183	809 417	250 364	380 364	312 300F	333 250F	90° 200F
	HANIBIA NIGER	63	110F	110F	100F	368 600	31 B	364 750	300 750	23	35F	40F	30F 9F
	NIGERIA	1398	16655	1710F	1746F 3F	869 3564	901 5400	906 5385	905 5000	1215	1500F	1550F 14F	1580F 15F
	REUNTON RHANDA	50	31 78	775	80F	1084	1012	1039	1063	54	83	80F	65P
	SAO TONE PRN SENEGAL	52	68	70*	491	1526 814	1535 684	1523 760	1556	42	45	1F 53¢	1F 55*
	STERRA LEONE SONALIA	10	13F 106F	13F 114F	13F 150F	984 793	923 984	962	1077 800	10	104	13* 111	146 120F
	SOUTH AFRICA	5290	50001	60005	7000F	1265	1648	1798 738	2093 769	6691 31	8240 454	1 0790 456	14650 50F
	SUDAN SWAZILAND	39	60F	61F 71	65F 650	780 846	750 1000	1360	1462	. 25	55*	97	95+
	TANZANLA TOGO	1005	1300F	1300F	1300F 120F	1109	692 1383	1283	577 1143	160	900F	750+ 154	750# 137
٠.	UGANDA UPPER VOLTA	343	500F	258 100F	260 100F	1221 659	809 8401	1109	1315 1000	419	453 104	286	342 100F
	ZAIRE	595	7067	.7105	7 1 5F	713	723	727	727.	424	509 4 7007	516* 800F	520F
	ZAMBIA ZIMBASWE	992	900F 800	1100F	1150F 1350	792 1495	778 1500	727 1343	870 2133	186 1366	1500	1539	2880
	N C AMERICA	33570	37536	39441	41424	4062	5787	4798	5599	136352	217230	189227	231937
	ANTIGUA BAHAHAS	1	16	1	16	2296 985	2000 1063	2055 1141	2083 1138	1	16		16
	BARBACOS	1	1 F	เห	· 16	5341	2614	2500 1690	2500	2	2F	2F 19	2F 19
	BEL1ZE CANADA	490	11 893	958	1058	1437 5078	1437 5583	5671	1686 5874	13 2487	4983	5434	6214
	COSTA RICA CUBA	100	50* 76F	50¢ 77F	47 77F	1123 853	1458 1250	1492	1876 1234	56 85	73 95F	75 95F	88 951
	DONINICA DONINICAN RP	27	194	240	240	1235	1419	1438 2042	1438 2083	46	38	49	501
	EL SALVADUR	203	276	292	259+	1670	1896	1806 1100	1882 976	340	523	527 1F	487
	GRENADA GUADELOUPE			"	:	856 1778	1200	1200	1200	1	-		1053
	GUATERALA HAITI	671 231	6224 234	2456	680 250£		. 1512 .784	1568 714	1,547. 720.	751 245	941 183	1041* 175F	1052 1808
	HONDURAS JAKATEA	283	341 40	3404 35	340* 4F	1198	1037	992 1327	994 1795	339	354	337*	3381 71
•	HEX 1CO	7512	5502	6955	8150	1218	1477	1780 1000	1812 1000	9025	8124	12383	14766
	MONT SERR AT MCCARAGUA	260	140*	1970	250*	912	1196	1164	1000	238	158# 63	229* 63*	2501 681
	PANANA PUERTO RICO	71	66	66.*	70*	859 1043	959 920	955 920	971 920	66	03		
	SATHY LUCTA					828 3207	700 3333	700 3313	727 3235		1.6	1.F	i.i
	TRINIDAD TOO	23749	27300	. เศ 29555	1F 30200	4153 5164	4167 6883	4167 5711	- 4167 - 6898	122649	201655	5¢ 168787	56 208314
	SOUTH AREP IC	16525	14690	16383	17361	1553	1741	1857	2212	25668	29058	30430	38406
	ARGENT INA	3880	2800	2490	3500	2247	3107	2570	3857	8717	8700	6400	13500
	BOLIYIA	10021	278 11319	293 11438	11491	1306 1365	1360	1306	1316 1835	13683	378 [6306	20374	250: 21098
	CHILE	70 884	130	116 614	629	3111 1251	3752 1414	3487 1323	4128 1399	217 856	489 870	505 813	518 880
	ECUADOR	315	219	256	2250	767	996 4000	1070	1094 4000	239	218	54.5	246
	FR GUTANA GUYANA		51	3F	.3F	1632	1158	1200	1000	2	ž	3.5	31
	PARAGUAY PERU	162 373	353 361	317 258	400F 316	1245 1621	1560	1553	1500 1986	201 605	550 621	585 443	628
	SURTHAME URUGUAY	194	94	132	145	1906 832	1671 758	1585 904	1611	161	71	119	196
			1 1	,,,,	- 14	1		I	1				·

Table B-3 (cont'd.)

		AREA HAR	ŧγ		1000 HA	AIFFD			KG/IIA	PROQUET	10%		1000 41
···		1969-71	1979	1980	1991	15-6961	1979	080	1861	1469-71	1979	1 480	1981
		606	519	436	335+	1152	1834.	1519	1451	698	848	645	4861
	A S J J J J J J J J J J J J J J J J J J			36926	37510	1644	2250	2308	2308	50413	82460	85243	86570
AS I	[Å	30558	36653		4 70F	1560	1652	1733	1698	707	760	797	798
	SHANTSTAN NGLADESH	453	460	460F	2F	670	1110	715	100	9	151	1 12F	11 121
atti	UTAN NATU	9	115	11F	11F 80F	1108 619	1219	1250	1250 2999	32376	60149	61105	1001
CHI	RMA LNA	15965	20167	200351	2053 TF	2028 819	2983	3050		15			10001
EA:	ST TIMOR BIA	5794	5721	5983	5800F	1051	- 979 1390	1137	1207	6087 2575	3605 3606	6804 4012	3991
180	DONES IA	2667 25	2594	2771 42F	· · 2735 426	965 1400	1357	1129	1190	35	57+ 100F	40F	501 901
18) 18)	AQ .	6	35F	35f	35F	1495 4804	2857 4038	2571 4893	5000	5	11	14	15! 3'
	RAEL Pan	12	2	. 3	Į.ė	2674 290	2591 4041	1989 7650	1000	33		19.51	4 4 4 4
108	RDAN HPUCHEA BH	94	75	99	95	1332	933 5210	1010 5789	1032 5789	1493	70F 1950F	100F	86
KO	REA DPR	302	370f	380F	380F 33	4950 1450	4587	4162	4383	63	149	154	145
	REA REP WAIT	ì		28	.31	1711	3000 1123	3000 1000	1062	26	32	28	33
LAU	o Bangn	15	28 2F	21	2F	904	1373	1333 1143	1333	15	2 • 8	8	
HAI	LAYSIA	ė	7	1	7	1942 4500	3333	3667	3667	198	554	743	720
NE	LOIVES . Pal:	439	450F	4508	450F 727	1812	1231	1651 1271	1500	697	875	947.	1.004
	KISTAN IL IPPINES	840 2356	701 3327	745 3281	3319	813	1600	450 (600	957 1600	1915	3167 4F	3117 4£	3116
SA	UDI ARABIA	19	3 F)	19 19	3F 23F	5010 775	1139	1166	1087	15	22 34	23 47	25 89
SY	I LANKA RIA	ه	18 1509	27 1562	36 1716*	1403 2567	1835 2187	2093 2017	2512	1979	3300	3150	3700
	A ILÀND RKEY	711 646	585	583	580*	1637	2308 1281	2127 1701	1897	1058 272	1350 475	1240	1100 540
V3 I	HAN TE HAN TE	239	371 314	246 314	350F 34 F	1136 1975	1548	1548	1441	16 5	48 * 15F	48* 15F	49 15
	MEN DEN	2	6F	1 of	1 OF	2528	2419	1500	1500	38635	56808	52214	53627
EU	ROPE	11478	12154	11586	11520	3366	4674	4507 2941	4655 2941	220	250+	250F	250
	BANIA	111	. 85F	85F 193	85F 189	1988 3547	2941 3110	6700	1270	154	1347	1293	1374 25
85	STRIA EGIUK-LUX	2	6	585	- 44 571	5510 3913	6130 4840	6270 3859	6250 4331	2436	37 3223	394 2256	2477
	LGARTA ECHOSLOVAK	623	190	159	LAGE	4021	5008	. 4685 5326	4211 5796	511 7394	10413	745 9358	800 9100
FR	ANCE RMAN OR	1436	1995 L	1757	1570 1*	5148 2631	5220 4692	4490	2000	9	. 6	672	2 832
GE	RHANY FR	99	115 123	119	129 159	5052 3073	6423 5804	5653 6812	6455 7862	590 498	741	1158	1250
	EECE NGARY	162	1367	1253	1350F	3570	5911	5324 6936	4815 7318	4542	7396 6197	6673 6403	6500 7290
	ALY THERLANDS	986	937	937 1	991	4630	3333	3922	4500	5 (2 M	2* 58	79
20	LAND RTUGAL	5 432	46 (394 (16 394	20 388	2449 1385	3900 1284	3544 1355	3965 1035	17 599	508	534	417
RG	Hansa	31.70	3311 {	3286 448	3150° 430	2320 3432	1752 4793	3392 4931	3556 5002	7354 1834	12425 2205	11153 2208	11200 2151
	AIN ITZERLAND	526 10	460 18	.14	17	6168	7542	4472	6882	٥i	134	1.6	117
ÚK		7391	2251	1F) 2161F)	2275F	4666 3095	4490	785 4312	1000 4308	3399	10084	9317	9800
	EAN! A	89	77	78	82	2911	4619	403Z	3724	260	356	316	304
AU	STRALIA	11	50	54	56	2387	3380 2000	2791 2000	2122	184	169 4f	. 151 4*	1 I B
FI. GU		2	2F	25	. 25	2130 1130	1570	1500	1500				2
NE:	W ZEALAND	و	22	l 19	1F 21*	2222 7753	8031	2925 6076	3500	10	1 179	156	177
PA	CIFIC IS		18	1F	15	1125 888	1171 2003	1195 2000	2900		28	ž.	2
	PUA N GUIN NUĀTU	l i	if	iř	1.5	509	500	520	520		1 F		1
US.	SR	3617	2667	2977	1545	2763	3139	3176	2257	9993	8373	9454	6000
	3 K 039, Y	35196	41157	42190	44491	4349	5930	5057	5885	155622	247620	216410	261810
	EUROPE	24238	30192 6488	30513 6200	31257 6153	3163 3618	6844 4991	5710 5012	5253	125137 23552	206638 32378	31075	314528
. 0	CEANIA TH DEV.PED	86 5303	5005	6005	7004	2951 1269	4814 1450	4185 (800	3851	255 6723	348 8256	10807	295 14550
	V.PING # E	53162	54723	56850	59259	1279	1376	1,54	1613	68016	15191	83230	96148
	FRICA	12586	14167	14511	14803	1003	907	903	1044	81651	12850	13107	15459
1.	AT AMERICA	25857	24034	25311	27521	1420	1650	1795	2028	36883	39648	45436	55816
	EAR EAST AR EAST	1824	2034	2050	1991 14934	2328	2633 1203	2723 1275	2486	14263	5154 17430	19999	19716
, Ø	TH DY.PING	3	, 5	5	. 5	1790	1613	1710	1737	6	8	8	9
	NTR PLANNO	25527	29316	29124	30274	2325	3255	3242	3097	59343	95447	94416	93746
	STAN CPE EURHUSSA	8927	20983 8333	20763 8364	9915	2309	2995 3936	3014 3659	3017	34265 25076	62644 32803	63873	29301
DE	Y.PED ALL	44723	50090	51154	53403	4041	5594	4829	5451	180745	280423	247003	.291117
	V.PING ALL	69761	75706	17610	15308	1466	1872	1895	1992	192284	(37936	(47053	160546

^{*} Unofficial figures
F: Estimates by FAO

Table B-4 Potato Production of the World

			AREA HAR	ı v		1000 HA	ALETO		1	KG/HA	PRODUCT	ION		1000 NT
			1989-71	tala	1980	1861	1969-71	1919	1980	1981	1969-71	1979	1980	1981
	MONTO	~~~~	20014	18515	17978	17861	1 3855	15571	12800	14387	277286	288294	230122	256978
1, 4	AFRICA		367	568	582	585	7976	8591	8870	8764	2931	4891	5161	5127
	ALGERIA	- 1	43	17	76	80F	5906	6493	7770	7771	253	501	591	6166
	ANGOLA BURUNOT		14	6F	6f 12F	6F 12F	6296 5435	7273	7273 1933	7273 1833	34	40F	40F 22F	40F
	CAMEROON		20	25F	25F	25	2672 10250	96B 16667	2000 16667	2400 16667	54 l	24 2F	50F 2F	hof 2F
	CAPE VERDE CENT AFR REP	į					1968	2000	2048	2091				
1	CHÀD CONGO	17	3	35	35)	4747 7083	5200 4815	5200 5867	5200 6000	12	13F	13F 2#	13F 2#
	EGYPT		30 30	60 38F	70+ 39f	63* 39F	5303	17061	17279 6182	17886	161	1019 235F	1214¢ 238F	1120* 240F
``	TYORY COAST	14.5	29	1 F 4 8 F	1 <i>F</i> 48F	2F 48F	7145	7500	10714 7292	11111	206	LOF 360F	15F 350F	20F 365F
	AYBLI		. 2	165	175	175	6450	564 7 6010	6050 5462	6053	15 106	90 183	103 159	103F 258*
	HADAGAS CAR HACA ¥ I		16 25	30 30F	29 32F	40F 34F	6428 3387	3667	3594	3493	85	110F	215F	117#
	MAURITANIA MAURITIUS			1	ı	1F	20670 15027	16000 15539	16000 16564	16000 16216	2 7	4 F 8	4F 12	4F 12F
:	HOROCCO HOZ AMBI QUE		28 6	40+ 5	38* 6F	28• 6F	10119	14200	14300 10909	14000	283	568* 57	543# : 60F	396 ¢ 62F
i i	NIGER		2	3F		3F	12514	15000	13333 14000	13333	25	35F	35F	358
	HIGERIA REUNION	. 1				- 11	11121	20000	19048	19048	134	45 213F	4F 221F	4.5
	RHANDA SENEGAL		19	31 F	32F 1F	33F 1F	7153 6543	5.854	5714	5581		. 5F	5F	5F
	SOUTH AFRICA		44	50F	50F	50F LF	13374	13920	13720	14000 19231	583 25	696 25F	686 25F	700F 25F
	SWAZILANO TANZANIA		2 13	3F 25F	3F 25F	3F 2 SF	2960 3873	2000 5600	2000 5600	2000 5600	50	6F	6F 140F	6F 1+0F
	TUNESTA		17	10 45F	1.1 45F	11F	7654 8800	12500 7333	11215 7333	11367 7447	69 147	125 330F	120 330F	124F 350F
	UGANDA ZAIRE		5	6F	64	6F	5377 8823	4921	4921	4906 9000	29	31F 3F	31 % 3F	
	ZAMBIA ZIMBABWE		2	۶F	2F	ŞF	11167	10000	10000	10000	22	224	23F	
	N C AMERICA		162	71.7	675	705	22954	26996	25918	26851	17482	19354	17501	18938
	BERMUDA CANADA	- 1	121	113	107	108	15565 19113	13446	6397 23417	6410 23624	2312	2760	2509	1 F 2555
	COSTA RICA		3	3 F		3 13F	9557 9359	10299	10024 18828	9333 18850	27 83	27 ZGL	27 239	28 239F
	CUBA DOMENTCA			5.0			4164	5518	5500	5688	1	1,5	15	15
	DOMINICAN RP		2	11	15	2F	12703 7367	10886	11385 19954	12857 19677	23	11	15 6	27# 6F
- :	GUADELOUPE GUATEHALA		1	145	8 1	98	7000 4209	3614	4750	4353	28	51	38*	3 T F
	HAIT L HONDURAS		1	1.F		1.F	15217 6608	15000 5412	15000 5618	15000 5667	7	9F 5F	9.F	9F 5F
	JAHATCA MEXICO	2 1 24	1	1P 56		1F 67	9826	9443 12982	9131 12500	8889 12945	10 489	11 727	7 892*	1 2F 868
	MONTSERRAT		10	- 10			2500 4000	2500 4250	2529 4250	2553 4250	1	25	215	2F
	NI CARAGUA PANANA		. 1	Şə	20	25	10057	6667	8000	8667	ð	10*	12*	138
	ST KITTS ETC		571	514	467	498	7133 25386	30220	8333 29405	8333 3036#	14483	15535	13737	15135
ļ .	SOUTH ARERIC		1601	980	904	947	8392	9879	9940	11116	8706	9633	9001	10531
	ARGENTINA BOLLVIA		190 95	110	105 131	117 150f	11617 6968	15470 5616	14874 6009	19255 6333	2212	1694 730	1568 787	2247 950*
	CHILE		214 76	204 81	181	171	7260 9312	9520	10745	11175	1557	2154 770	1948	1911
	COLOMBIA ECUADOR		84	148	142	160 31*	10407	13284	12160	13169	87 L 560	1966	1727 323	2100 349*
	PARAGUAY		1	1.0	194	189		8828 7005	8900 7106	9000 8598	6 1877	1695	9 1380	9F 1627#
	URU ÇUAY		293	242 21	- 13	50*	6040	6493	5668	6500	135	135	110	130*
	ASTA		2603	17 3308	1? 3150	3227	8 2 5 8 10 2 4 1	13366	13000	10526	121 26651	225 38441	247 36718	200* 37929
	AF GHANISTAN	-	16	215				12619	17619	12759	172	265	265F	
	BAHRAIN	.				1.35		25000	25000	25000 9181		909	917	999
	BANGLADESH BHULAN		85 1	91 2F		102 2F	9863	9406	9506 6661	6750	842 5	134	13F	14F
	WRMA CHINA		9 1336F	1556F		14 1505F	4576 9254	4851 9986	4703 9996	4702 9996	40 12362F	54 15536F	575 15036F	15039F
	CYPRUS: EAST TIMOR		10	8*	9+	98	18641 9519	23256	23043	24888	183 L	190*	208*	
	GAZA STRIP		501	807	685	732	8950	15000 12555	16667 12152	16667 13113	4482	10133	5F 8327	5F 9599
	INDONES IA		16	22 317	23F 817	25F 81F	0 6111: 9072	9456 8494	9395 8519	9368 8702	99 617	204 688F	219F 69D¥	234F
	IRAN IRAQ		ı	71	7.5	7 F	14256	15000	L5000	14568 35455	16	105F	105F	
	ISRAEL JAPAN		164	125	123	1264	25869 21238	35533 27031	36170 27723	25896	131 3490	211 3381	3421	3250
	JORDAN Korea opr		90	120F	1 125F	1F 130F		14877	17293 12400	17297	960	1500F	13 1550F	
	KOREA REP		54	34	37	41	11100	15000	11931 15000	13512 15000	598	356	446	554
	LAO LEBANON	1	3	4.5 76		4F 7F	6667 10694	7750 16471	7907	7955 20431	- 17 96	31 F	34F 145	35F
	LEBANOA		,	["	'	"	1,007	1			t "		• •	[

Table B-4 (cont'd.)

1090-71 1979 1980 1981 1990 1991 1980 1991 1990											An artista		
		AREA HARV	,		1000 HA	YIELO			(G/HA	PRODUCT	LON	,	1000 HT
MORDELLA 0		1969-71	1979	1980	1981	1969-71	1979	[980]	1981	1969-71	1979	1980	1981
MOREGILA 3 1 1 1 1 1 1 1 1 1							0018	5270	7143	18	72	: 39	5 !
Separation 1	HONG OL LA										279		275
PARLES 10 2 2 2 7 7 7 7 7 7 7													39
SMILE MARAELA SMIL MARAELA S								1108		5.5	5.0	31	3(
SAID ARABIA		1 1	• •	7.	"								ĺ.
SET LANKA 3		1	. !							10			50
SYMIA 10 100 11		1 3 1	4	. 5	. 5E	9545							27
THALLAND 1			14		18								
UA CHIRATES 1.00													2900
U RERIGATES 7 103 82 100F 15449 6473 10485 7500 103 084 862 177 118 118 127 1494 17218 17200 131 171 131 118 127 1494 17218 17200 131 171 131 118 17200 131 171 131 13		160	169	183	180*	12377							
VIETE ANA **TEREN ANA **TEREN DEN **TEREN		1	1		1005	15440				103	684		75
TERRIL DEM TYSEN DEM	VIET NAH									35			
ALBANIA		,		11.					4000	1	15	15	'
ALBANIA ALBANIA ALBANIA 100 108 108 109 109 109 109 109		7172	5928	5684	- 5500	17665	10011	16453	20232	156501	123963	93515	111270
ALBRITA 100 100 100 100 100 100 100 1					105	5747	7697	7667	7135	109	1376		140
AGLOLIN-LUX 166 197 198 198 198 198 198 198 198									26386				131
SALE CARELLAND 30 39 35 36 12781 1095 8598 11930 8598 11930 8598 1292 2005 3391 211 195 1978 14688 1298 2005 8598 11930 868 5124 2005 8599 11930 8183 1213 1213 1213 1213 1213 1213 1213												1450	
CZECIOSLOWAK 331 213 195 1956 14683 17508 13804 12709 4867 COEMARR 34 32 38 31 2478 26818 2904 2804 11 15 15 15 15 16 16 16 16 16 16 16 16 16 16 16 16 16							10876		11145				
DEMMARR 33 32 34 31 24188 2018 1200 1150 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					1954	14683							
ERREAD 56 43 41 37 1400 12449 1000 12449 1000 1146 6446					3.7								•
FINALNO		1 1		4 1									47
FRANCE 376 275 230 215 2220 1795 20664 10412 12223 9214 1056 6EENAN OR 64.3 54.9 513 513 177 17512 31414 27952 2000 15804 10205 1970 600 6EENAN OR 64.3 54.9 513 513 177 17512 31412 27952 2000 15804 10205 1970 600 6EENAN OR 64.3 54.9 513 513 100F 11077 14477 14908 1600 1874 1512 1372 1616 1616 1617 1447 14908 1617 1900 1000 1874 1512 1372 1616 1617 14478 1618 1618 1618 1618 1618 1618 1618 16													648
GERMAN DE ALS SAP STORM TO STO													1050
CERMIN No. 1.00	GERNAN DR								29000		10205	7970	804
SREECE 129 105 93 1009 11097 14477 14498 16000 1874 1512 1392 166 1898GARD 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1						12727		15844					95
STREET	GREECE					11097				1874	1512		160
TRELAMD	HUNGARY												
1													
The content of the													
MALTA MALTA METHERLANDS 153 166 172 165 39169 39177 39187 39169 207 212 227 24919 20300 24976 24976 24919 24976 24977 24977 24978 24919 24970 24976 24977 24977 24978 24979 24977 24977 24978 24979 24977 24978 24979 24979 24977 24978 24979 24978 24979 24978 24979 24978 24978 24979 24978 24979 24978 24978 24979 24978 24978 24979 24978 249	LIECHTENSTEN												
NETIMERIAMOS 153 166 172 165 35160 37772 36413 30710 3207 6700 5218 520 521 722 24919 20900 27076 24071 3807 40912 3007 2401 2344 2258 16503 20911 11260 11550 20911 11260 1155 16503 20911 11260 11550 2091					- 2F								
NORMAY 32 20 21 2244 2258 18530 2021 11200 11200 2030 11200 2030 2030	NETHERL ANDS			172									
POLAHO		32											
PORTICIAL 117 118													93
SPAIN 389 355 356 377 1281 15879 16139 16391 4985 56377 5737 555 59AIN SIGNER 527 42 51 43 27511 30718 16139 16391 4985 56377 5737 555 59AIN SIGNER 527 42 51 43 27511 30718 16139 1													450
SPAIN SHOP							15010						557
SMITZERLAND 31 204 205 102 207 206 102 207 207 206 102 207 207 207 208 207 208 207 208 208 208 208 208 208 208 208 208 208							30738					1084	127
VINCOSTANTA 270 204 205 27223 31867 31476 31862 7356 6485 7109 641													104
TUGOSLAVIA 328 296 287 3004 9198 9203 8502 8333 3020 2774 2440 251 0CEANIA 52 43 45 43 20392 24707 25252 23319 1069 1066 1143 1143 1143 AUSTRALIA 43 35 37 35 18384 22937 25252 27319 1069 1069 1066 1143 1143 1141 1141 AUSTRALIA 43 35 37 35 18384 22937 25252 27319 1069 1069 1066 1143 1143 1141 1147 114										7355			610
AUSTRALIA 43 35 37 35 18384 22937 23387 25862 782 795 851 6711 1 275 275 851 851 851 851 851 851 851 851 851 85								8502	8333	3050	2724	2440	250
AUSTRALIA FIJI FR POLYNESIA REMCALEDONIA REM ZEALAND PAPUA N GUIN USSR 8019 6970 6936 6854 11669 13050 9663 10505 93739 90956 67023 7200 DEV.PED M E 3921 3121 2996 2922 21278 24172 23667 24306 83425 75431 70913 7100 N AMERICA 592 627 574 607 24289 291170 26288 29166 16796 18296 16247 1769 OEV.PED M E 2964 2270 2200 2093 20699 22812 2291 22949 61360 51788 49250 480 OTH DEV.PED 213 181 178 181 19754 23687 24023 22001 4204 4229 4250 480 DEV.PING M E 2430 2981 2843 2936 8782 10716 10705 11326 21337 31940 30433 332* AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 314 DEV.PING M E 2430 2981 2843 2936 8782 10716 10705 11326 21337 31940 30433 332* AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 314 LAT AMERICA 1108 1070 1006 1046 8480 9994 10189 11260 9397 10691 10255 117 FR PEAST 745 1072 944 1017 8898 11604 1129 12905 5744 6203 6204 1129 12905 6204 1129 1129 1205 5024 1129 1205 5024 1129 1205 5024 1129 1205 5024 1129 11295 112	OCEANTA	52	43	45	43	20392	24707	25252	27319	1069	1066	. .	118
FI 10 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	ATISTO AL TA	43	35	37	35	18384	22937			782			89
REPOLYMESTA REW ZEALAND REW ZEALAND RAPPUA N. GUIN USSR 8019 6970 6936 6854 11689 13050 9663 10505 93739 90956 67023 7200 DEV.PED N. E. 3921 3121 2996 2922 21278 24172 23667 24306 83425 75431 70913 710 N. AMERICA M. EUROPE 2964 2770 2200 2093 20699 22812 27391 22949 61360 51788 49250 OCEANIA DTH DEV.PED DEV.PING N. E. 2430 2981 2843 2936 8782 10716 10705 13328 21337 31940 30433 3322 AFRICA LAT AMERICA MERICA 108 390 428 418 1221 1449 1648 19794 10189 11260 9397 10691 10255 117 REW.PING N. E. 2430 2981 2843 2936 8782 10716 10705 1328 21337 30940 30433 3322 AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 3140 AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 3140 AFRICA 108 396 428 418 12211 14498 1448 1449 1449 1449 1070 1006 1046 880 9994 10189 11260 9397 10691 10255 117 FAR EAST 745 1072 964 1017 8898 11604 11239 12005 6626 12444 13602 57466 6203 600 CENTR PLANND 13663 12413 12139 12003 12627 14575 10608 12721 172523 180924 128775 15266 ASIAM CPE 1436 1786 1719 1742 9362 9964 10174 10013 13443 17792 17487 1748 E EUR.PUSSR 12227 10628 10420 10261 13010 15349 10680 13181 159080 163131 11288 1352 DEV.PED ALL 16148 13768 13417 13183 15018 17352 13580 15647 242506 238562 182702 2062		· ·	- 1	, et .									
NEW CALLEDONIA NEW ZEALAND PAPULA N GUIN USSR 8019 6970 6936 6854 11689 13050 9663 10505 93739 90956 67023 7201 DEV.PED M E 3921 3121 2996 2922 21278 24172 23667 24306 83425 75431 70913 7100 N AMERILA 692 627 574 607 24289 29170 26288 29166 16796 18296 16247 1764 M EUROPE 2964 2270 2200 2093 200699 32812 27391 22949 61360 51788 49250 800 OCEANIA 52 42 42 20498 25126 25628 27763 1066 1059 11356 11360 11360 1070 11328 27331 31940 30433 3322 AFRICA 290 441 443 454 6245 6916 7068 7007 1813 3051 3133 3140 DEV.PING N E 2430 2981 2843 2936 8782 10716 10705 11328 21337 31940 30433 3322 AFRICA 290 441 443 454 6245 6916 7068 7007 1813 3051 3133 3140 LAT AMERICA 1108 1070 1006 1046 8480 9994 10189 11260 9397 10691 10255 117 PARE EAST 745 1072 964 1017 8898 11604 11230 12035 6626 4 7 9 ASIAM CPE 1436 1786 1719 1742 9362 9964 10174 10013 13443 17792 17487 1756 E EUR-USSR 12227 10628 10420 10261 13010 15349 10680 13181 159080 163131 111288 13522			·]										
New Teal and New	NEW CALEDONIA	(1.1						1 - 1	1	
USSR 8019 6970 6936 6854 11689 13050 9663 10505 93739 90956 67023 7200 DEV.PED M E 3921 3121 2996 2922 21278 24172 23667 24306 83425 75431 70913 710 N. AMERILLA 592 627 574 607 24289 29170 26288 29166 16796 18296 16247 1764 OEBROPE 2944 2270 2200 2093 20699 22812 27391 22449 61360 51788 49250 480 OCEANIA 52 42 44 42 20498 25126 25628 27763 1666 1059 1135 11 OTH DEV.PED 213 181 178 181 19754 23689 24023 22901 4204 4289 4782 41 OEV.PING M E 2430 2981 2843 2936 8782 10716 10705 11328 21337 31940 30433 332 AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3054 3133 314 LAT AMERICA 1108 1070 1005 1046 8480 994 10189 11260 9397 10189 10255 117 FAR EAST 745 1072 964 1017 8898 11604 11239 12055 6626 12444 10834 122 OTH DV.PING 1 1 1 1 1 1 1 1 1	HEN ZEALAND PAPUA N GUIN	9	8	8	88						(077	, ,,,,	
NAMERICA 1092 1074 10013 13443 17792 17487 1756 1074		8019	6970	6936	6854	11689	13050	9663	10505	93739	90956	67023	7200
# EUROPE 2964 2270 2200 2093 20699 22812 22991 22949 61360 51788 49250 480 OCEANIA 52 42 44 42 20498 25126 25126 27763 1066 1059 41350 1135 113 113 181 181 19754 23689 24023 22901 2666 1059 4482 418 181 19754 23689 24023 22901 2668 1059 4482 418 181 19754 23689 24023 22901 2668 1059 4482 418 181 19754 23689 24023 22901 2668 1059 4482 418 181 1970 1006 1046 8480 9994 10189 11260 9302 10691 10255 117 1108 1070 1006 1046 8480 9994 10189 11260 9302 10691 10255 117 107 107 107 107 107 107 107 107 107	DEV.PED H E	3921	3121	2996	2922	21278	24172	23667	24306	83425	75431	70913	710
# EUROPE 2964 2270 2200 2093 20699 22812 22949 61360 51788 49250 0CEARLA 52 42 44 42 20498 25926 25628 27763 1066 1059 1135 11 0T 0EV.PED 213 181 178 181 19754 25889 24023 22901 4204 4289 4289 4282 41 0EV.PING M E 2430 2981 2843 2936 8782 10716 10705 11328 21337 31940 30433 332 AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 3140 1108 1070 1006 1046 8480 9994 10189 11260 9392 10691 10255 117 NEAR EAST 287 396 428 418 12211 14498 14489 14441 3502 5746 6203 60 FAR EAST 745 1072 964 1017 8898 11604 11239 12035 6626 4 7 9 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	N AMERICA	892	627	574	607	24289	29170	26288					1769
OTH DEV.PED 213 181 178 161 19754 23689 24023 22901 4204 4289 4282 41 DEV.PING N E 2430 2981 2843 2936 8762 10016 10705 11326 21337 31940 30433 332 AFRICA LAT AMERICA LITOR TO TOUGH		2964	2270	2200	2093								
DEV.PING N E 2430 2981 2843 2936 8782 10716 10705 11328 21337 31940 30433 332 AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 314 LAT AMERICA 1108 1070 1006 1046 8480 9994 10189 11260 9397 10691 10255 117 NEAR EAST 287 396 428 418 12211 14498 14489 14441 3502 5746 6203 60 THE PLANNO 1012 964 1017 8898 11604 11239 12055 6626 12444 19834 122 CENTR PLANNO 13663 12413 12139 12003 12627 14575 10608 12721 172523 180924 128775 1526 ASIAN CPE 1436 1786 1714 1742 9362 9964 10174 10013 13443 17792 17487 1756 E EUR-USSR 12227 10628 10420 10261 13010 15349 10680 13181 157080 163131 11288 1352	OCEANLA												
AFRICA 290 441 443 454 6245 6916 7068 7002 1813 3051 3133 31 LAT AMERICA 1108 1070 1006 1046 8480 9994 10189 11260 9397 10691 10255 117 NEAR EAST 287 396 428 418 12211 14498 14489 14441 3502 5746 6203 60 PAR EAST 745 1072 964 1017 8898 11604 11239 12035 6626 4 7 9 CENTR PLANNO 13663 12413 12139 12003 12627 14575 10608 12721 172523 140924 124775 1526 ASTAM CPE 1436 1786 1719 1742 9362 9964 10174 10013 13443 17792 17487 1756 E EUR+USSR 12227 10628 10420 10261 13010 15349 10680 13181 157080 163131 111288 1352 DEV-PED ALL 16148 13748 13417 13183 15018 17352 13580 15647 242506 238562 182702 2062	et ger	1 1	. 1	- 1		100		4 . 4]			
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FAR EAST 745 1072 964 1017 8898 11604 1239 12055 6526 12444 10834 1227 0TH DY.PING 1 1 1 8413 7084 8159 8066 4 7 9 9 1208 1208 1208 1208 1208 1208 1208 1208													
OTH DY-PING 1 1 1 8413 7084 8159 8066 4 7 9 CENTR PLANND 13663 12413 12139 12003 12627 14575 10608 12721 172523 180924 128775 1526 ASIAN CPE 1436 1786 1714 1742 9362 9964 10174 10013 13643 17792 17487 1748 12227 10628 10420 10261 13010 15349 10680 13181 157080 163131 111288 1352 DEV.PED ALL 16148 13748 13417 13183 15018 17352 13580 15647 242506 238562 182702 2062													
CENTR PLANNO 13663 12413 12139 12003 12627 14575 10608 12721 172523 180924 128775 1526 ASIAN CPE 1436 1786 1719 1742 9367 9964 10174 10013 13443 17792 17487 1756 E EUR+USSR 12227 10628 10420 10261 13010 15349 10680 13181 159080 163131 11288 1352 DEV.PED ALL 16148 13748 13417 13193 15018 17352 13580 15647 242506 238562 182702 2062		143											12.6
ASIAN CPE 1436 1786 1719 1742 9362 9964 10174 10013 13443 17792 17487 1746 E EUR+USSR 12227 10628 10420 10261 13010 15349 10680 13181 157080 163131 111288 1352 DEV-PED ALL 16148 13748 13417 13183 15018 17352 13580 15647 242506 238562 182702 2062	*	13663	ł						18.5%				15269
DEV.PED ALL 16148 13748 13417 13193 15018 17352 13580 15647 242506 238562 182702 2062	ASTAN CPE												1751
		1 . 1										,	13524
DEV.PING ALL 3866 4766 4562 4678 8997 10434 10505 10838 34780 49732 47920 506	DEV. PED ALL	1	. [[į	{ !	()	2062

^{*} Unofficial figures

F: Estimates by FAO

Table B-5 Cassava production of the World

	AREA HAI	RY		1000 HA	YIELO			KG/HA	PRODUCT	ON		1000 HT
	1969-71	1979	1980	1981	1969-71	1979	1980	1981	1969-71	1979	1780	1981
NORL D	10899	13680	13777	14054	8880	8601	8674	9055	96695	117651	119506	127261
AFRICA	5194	7179	7334	7433	6617	6266	6346	6433	38339	44979	46539	47816
ANGOL A	121	1 3 OF	130F	1305	13196	13846	14231	14615	1597 533	1 800F	1850F 971*	1700
BEN IN BEN IN	88 80	113	125F	125F	6084 10542	6492 14935	7768 15584	7800 15584	843	735 1150F	1200F	1200
CAMERODA CAPE VEROE	161	230f	233F	2365	3965	4348 24000	4292 24000	4283 24000	637	1000 °	1000F	1011
CENT AFR REP	262	320F	327F	334F	2930	3031	2999	3061	767	970F	980F	1021
CHAD COMOROS	23	50f 28f	50F	51F 29F	3162 3088	3600 3080	3700. 3071	3723 3084	140	180F	185F 86F	186 88
CONGO EG GUINEA	99	80F 22F	80f 22F	80F 23F	4660 2745	6575 2364	6500 2353	6625 2310	461 43	526F) 52F	- 520¢ 536	530 54
GABON	44	405	. 41F	426	3504	2625	2439	2381	155	105F	100F	100
GAMBIA GHANA	105	219	2F,	25 230F	3820 7419	3500 8032	3500 7826	3500 8043	1533	1759	1800F	1850
GUINEA	70	68	78F 219	85F 220F	6848 3300	6998 3443	7051 3425	7059 3545	482 546	475 730F	550F 750F	600 780
IVORY COAST	166	212 80f	61F	816	7969	7875	7888	7901	510	630F	635F	640
LIBERIA HADAGASCAR	19 188	82F 267	85¥ 278	07F	33\$6 6514	3659 5685	3529 6098	3621 5873	1227	300 %	300F	31 ! - 174!
HALAHI .	15	136	15F	15F	5921 7213	6154 8523	5000 8585	6000 8646	90 36	80F	90F 56F	90 58
HALI HAURITIUS	?	75	75	7.F	16467	20222	17200	16667				
MOZAMBIQUE NIGER	450 25	602 28	600F	1006 185	5669 7155	8005	4667 8036	4750 7923	2549 182	2700F5	2800F 225F	285
NIGERIA	894	1150	1200F	12006	10592	9130	9167	9167	9473	10500F	11000F	1100
REUNION RWANDA	29	438	AAF	4.5F	6333 11358	10000	10000	10000	333	4 <i>F</i> 450F	4F 472F	49
SAD TONE PRN	1 3	8	87	75	10515 4365	11111 3166	11111 3500	11111 3571	159	3F)	3FJ 28FJ	z
SENEGAL SEYCHELLES	36				6062	5000	5000	5000				
STERRA LEGNE SONALIA	16	186	22F 3F	22F 3F	4676 19573	11071	4318 11034	4316 11000	75 25	85F	95 F 32 F	3
SUDAH	39.	50	45	45F 950F	3402 4854	2541 4892	2689 4894	2687 4895	133	127 4550F	122 46008	12 465
TANZANIA TOGO	695 31	930s 28s	288	. 28F	17926	15821	16071	16766	562	543	\$50 ₽	47
UGANDA UPPER VOLTA	277	400E	410F 6F	420F	3825 6000	3375 6774	3415 6774	3381 6774	1058	1350F	1400F	142
ZAIRE	1493	18005	1818#	1 B53F	5851	6667	6711	7017	10232	120004	12200	1300
ZAMBIĂ Zimbabwe	52 15	55E 18F	55F 19F	57F 18F	3064 3033	3182 3000	3182 3243	3123 2778	159 46	175F	175F 60F	. 17
N C AMERICA	125	147	148	157	6784	5995	5855	6091	783	088	865	95
ANTI CUA BARBADOS			Ì		3625 26355	2353 25000	2353 25128	2353 25000		គេ	1 F	
CAYHAH IS					3600	4500	4500	4500	l' ., l	15	18	. 1
COSTA RICA CUBA	33	2 F 4 6 F	3F 47F	3F 48F	5566	6600 6957	6696 6915	6667 6899	12 217	320	325	32
DOMINICA DOMINICAN RP	15	116	LIF	1.8F	9958 11490	10000	10482	10000	1 173	119	115	18
EL SALVADOR	[.]	2	2	2	9102	11829	L1843	11840	12	20	20	2
GRENADA GUADELOUPE			}		11250	5367 8000	9000	4000 9000	. 5	1	1	
GUAT EHALA	2	. 36	38	3F	3000	2600 4038	2567 3968	2567 4048	6 205	8F 254	814 250 H	25
HAIT! HONDURAS	51 4	63	43F	63F 2F	6984	2871	3281	3273	28	9 [7	
JAHATCA MARTINIQUE	, 2	21	2.5	ŞΕ	8241	10892	9326 5000	10833 5000	16	24	20	
MEXICO	2	2	5	2F	20000	13390	9961	10000	43 17	33 24 F	20 26#	. 2
HICARAGUA Panama	5 6	. 6f	7F 5F	76 58	4111 8547	4000 8473	4000 8333	8212	32	40 [408	4
PUERTO RICO SAINT LUCIA	1				3474	8228 3462	8252 3500	7678 3500	4	រុម	3 16	
THEORY IZ		·			13516	1 2000	12000	12000 12500	3	3 F 5 *	36 56	1 -
TRIHIDAD TOB SOUTH AMERIC	2483	2591	2489	2577	13873	12500	12500 11663	11905	34444	30089	29034	3067
ARGENT IN A	26	21	23	2 3F	11555	8551	9639	10000	296	183	222	23
. BOL 1 VI A.	18	16	18	1 9F	12671	12737	12150	12432	223	201	219	2 250
ERAZ11 COLOH81A	2042	2111	2006 208	2093 207	14655	11825 6610	11669	11968 10386	29922	24962 1909	23411	211
ECUADUR	35	20	25 IF	26 ×	10753	9115	9109	9100 9877	382	183	229 8F	2.
FR GUIANA Paraguay	100	126	136	135F	34365	14936	14967	14815	1442	1888	2031	504
PERU Suriname	37	35	350	360	12716	11487 6995	11429 6897	11389. 7073	417	503	. 400* 3F	
VENEZUELA	39	. 38	37	378	8080	9117	9663	9130	317	350	361	36
ASIA	2470	3743	3786	3866	9289	11083	11316	12307	27943	41484	42843	475
BRUNEI]		.		7107	8571	7778	7838	2	. 3	35	
BUR MA CHINA	1618	3 215#	3 24 3F	251F	12062	12693	8991 13101	9283 13061	10	23 2726F	28F 3185F	321
EAST TIMOR	7	362	352	346	2816 14176	16736	16611	16803	4993	6050	5845	581
1110 TA 1110 QNE 5 TA	352 1424	1434	1414	.1412	7512	9587	9571	9718	10695	13751	13532	137
KO ABHOUGHAX GAB	1	221	254 5F	25F 5F	13636	6500 15000	6000 15111	5800 15217	24	143F 60F	150F	
HALAÝSIA	30	407	350	35F	9096	10000	10000	10286	271 436	400F 2769	350F 2276	34 239
PHILIPPINES SINGAPORE	82	192	204	2006	5297 10744	10310	10000	10000] ",	15	22,16 LF	
•		1	1 1		I		i	I '	1		I	1

Table B-5 (cont'd.)

	AREA HAR	Y		AH 0001	AIEFD			KG/HA	PRODUCT	E OR	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1000 HT
	1969-71	1979	1980	1981	1969-71	1979	1980	เจลเ	1969-71	1979	[980	1981
	 		. 51	45¢	5155	9974	6876	8703	376	535	199	537
SRI LANKA	65	54.		1050*	15205	12737	13414	17048	3208	12100F	\$3615F	
THA IL AND	211	950F	10150		7153	7367	7476	7158	950	3444	3290	3400
VIET NAM	[133	468	439	475F	11.33	'7"	1775			[100	. 17.
OCEANIA	17	20	50	21	10972	10894	11035	11124	187	219	225	259
CCEATITA			.1			14167	13462	12057				
AMER SANDA	i i	1	. 1		26800	32258	32258	32259		4.5	. : AF	1
COOK ISLANDS	1		1		27445		15000	12141	85	936	946	9
FIJI	1 7	8#	8F	-8F	12305	12000	18630	18356	. 6	7.6	7.6	
FR POLYNESIA	1	· .	i i		17877	18611		30000		36	38	
NEWCAL EDGNIA	1	·i			22699	56000	28000	3846	1 T			
NIUE	1 . 1	į			5200	3846	3846		6	91	98	
PACIFIC 15	1 1	. 15	LF	₹F	9032	9000 (9300	9600	72	928	948	9
PAPUA N GUIN	1 71	96	9.5	9F	10188	10698	10682	10667	. 12	420	346	
		- 1	- 1		18750	10867	10667	10667	12.2		ام و و	ı
SAHOA TONGA	2	2 5	35	ZF	5742	5238	6413	6667	10	115	134	
DEV PING H E	10593	12975	13070	13303	8853	8501	8636	9054	93184	111339	112881	12044
	1 : 1			7000	4430	6292	6369	6456	38206	44852	46417	4169
AFRICA	5755	7129	7288	7386	6639	11312	11338	11572	35227	30970	29900	3163
LAT AMERICA	5603	2738	2637	2733.	13510	2541	2689	2687	133	127	122	15
NEAR EAST	39	50	+5	45	3402		11762	13083	20031	35172.	36218	+016
FAR EAST	2174	3039	3079	3116	9213	11574		11124	187	219	225	22
OTH OV.PING	17	20	5.0	. 51	10972	10894	11035	11124	191			
CENTR PLANNO	296	704	707	751	9844	8984	9372	9085	2912	6313	6623	662
ASTAN CPE	296	704	707	751	9844	8964	9372	9085	2912	6313	6625	582
DEV PING ALL	10889	13680	13777	14054	8880	8601	8674	9055	96695	117651	119506	12726
		į				į						

Unofficial figures Estimates by FAO

Table B-6 Sweet Potato Production of the World

Detail 17 3 3 3 4 30 300 350 350 66 11 14 14 14 14 14 14		AREA HAI	l V		1000 HA	YICED			KG/HA	PRODUCT	10н		1000 HT
AFFERS 677 775 775 770 791 5917 644 645 645 4607 4004 679 5005 5111 AUGULA 10 107 177 177 177 177 178 179 5917 644 645 645 645 645 645 645 645 645 645		1969-71	1979		1981	1969-71	1979	1 490	1991	1969-71	1979	1980	1981
MARCOLLA 10 10 10 10 10 10 10 1	MORED	12268	12505	15059	11771	11586	12153	12061	12384	142140	148293	145045	145765
Detail 17 3 3 3 4 30 300 350 350 66 11 14 14 14 14 14 14	AFRICA	677	175	185	794	5917	6444	6455	6487	4004	4993	5065	5151
Description Col. Str.													180F
CAPT VISION 1 1 11 12 13 14 160 160 170 170 170 170 170 170 170 170 170 17	BURUNDI	101	15.6	93F	93F	8531	9457	989.5	9982	831	8701	920F	.928F
COURDOS 1					11	1864	5000	5000	5000	3	5 F	5F	58
CONGO CO													37F
COLINEA	CONGO	4	4.6	4F]	4F	9000	6099	6098	6220	24	· 25F	25F	26F
Video Coast	EQ GUINEA	. 7	115	LIF	1 2F	3727	3000	3009	2957	ξÌ	33F	34F	346
Septical 10 196 346 397 7733 3917 3944 3846 210 3407 3306 3457 3406 3407 3406 3407 34	GUINEA	LO-	10	105	105	8200	7500	7400	7500	82	73	74F	75F
MARICASCAM 59 78 806 406 3984 3242 3125 3125 348 407 4106 4100 1107 1100 1107 1105 32 27 7 7 8 806 407 410 1100 1107 1105 32 27 7 7 9 9 9 9 9 1 120 1100 1107 1105 32 2 7 7 9 9 9 9 9 9 1 120 1100 1100 1107 1105 32 2 7 7 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	KENYA :			38F	39F	7753	8047	8584	8846	230	340F	330F	3455
MALTERIAL 3 4.6 4.6 4.6 1225 11007 1													16F
MACHENIS 1 1 1 1 1 1 1 1 1	HALI	. 3	4.6	45		12625	11707						50F 2F
MIGGER	MAURITIUS	·		į		12667	13125	17000	10000				
RECHINGON	NIGER	1	5	· 5F	5 F	6729	5932	5957	6064	9	28	28F	29F
SERIÉGIA LÍCHE 1		15	186	19#		18000	21923	22692	23462	5	. 6F	67	. 6F
SIRÉRÉA LLONE 7 58 6 6 64 2255 2207 210 2080 200 1119 1275 134 134 134 134 134 134 134 134 134 134	RKANDA											.7F	938F 8F
SOUTH AFFICA 13 135 135 135 135 137 1290 2015 1535 1538 3697 39 A6 A6 48 48 500AM 1 1 257 227 2465 2418 2700 2400 21 445 425 460 3348 25 320 320 21 445 425 460 3348 25 320 320 320 320 320 320 320 320 320 320	STERRA LEONE					2656	2037	2105	2083	20		12F	13F
SMALLIAND 2 2 72 27 28 4013 5682 5682 5682 5882 8 107 107 107 101 101 101 101 101 101 101	SOUTH AFRICA					2900	3545	3538	3692	. 39	46	46	. 48F
1000	SHAZILAND	Ž	2F	2F)	35	4013	5882	5882	5882	8	105	105	105
Company Comp			28	35	75	2500	3000	3500	3500	3	6F	. 7F	7F
ZARIEL 53 63F 63F 63F 63F 63F 64F 67T 7704 7163 7143 7143 717 20F													680F
N. C. AMERICA 200 206 204 211 6930 6528 6061 6369 1387 1356 1237 1347 NIT 101A BHARMS BARRAMOS 1 BHARMS BARRAMOS 1 BERMUDA 15538 619 4000 4000 6000 CATMAN 15 63 800 807 807 8000 53300 5300 5000 6.000 CATMAN 15 63 800 807 807 8000 13500 5300 5300 238 3164 3255 3277 CUBB 1 600 1101C 138 8050 1200 10000 10000 10000 11 15 15 15 15 15 15 15 15 15 15 15 15	ZAIRE	53	635			4977							309F 20F
BAHARASS BARBADOS BAR				ì		,					[. [í	
BRABADOS 1 1 1F 1F 9620 6857 9278 8000 6 2 4 4 44 65 657 9 6600 6400 6400 6400 6400 6400 6400 64			·	:								ļ	
CAPPAM IS CUBA C	BARBADOS	1		1.5	· If	9650	6857	8778	8000	6	. 2	4	45
DOMINICAN DOMI	CAYMAN 15		ر م د			5000	5500	5500	5500		,,,,	3354	2270
SALVADOR	COMINICA			- [10000	10000	10000	10000	. 1	1 F	18	1.1
GUADELQIPE HAITI 1 53 63 63F 64F 5283 4235 4217 4219 220 225 260F 270F HONDURAS JAMAICA 2 2 2F 2F 2F 79701 11757 11662 12500 16 21 21 21 21 21 MINIODURAS JAMAICA 2 2 2F 2F 2F 79701 11757 11662 12500 16 21 21 21 21 21 MINIODURAS MARTINIQUE 1 1 1F 1F 1 9995 9920 11000 10000 6 5 5 6F 5 5 6F 5 5 MINIODURAS MARTINIQUE 1 1 1F 1F 1 9995 9920 11000 10000 6 5 5 6F 5 5 6F 5 5 MINIODURAS MONTSERRAT MOUNTSERRAT MOUNTSERRAT MURITIRICAL 2 2 2F 2F 28 313 1811 1313 1313 1313 1215 122 250 125 125 125 125 125 125 125 125 125 125	EL SALVADOR	10		101	. 197	5038	5000	5000	5,000		39	50	ار د
Michoduras 2 26 26 27 2790 31750	GUADELOUPE		ı	1		10000	9400	9700	10200	. 6			5F
JANALCA 2 ZE		53	63	63F	. 64F				3750	1.	โเรี	16	16
MEXICO)										6F	5
PRINTED RICO STRITTS CIC Control	HEX 1CO		4							.142	58	33	33F
SALMI LUCIA ST VINCENT 1	PUERTO ALCO	. 3	26	2	25	3496	4255	4358	4362	10	9	. 9	10
TRINIDAD TOB	SAINT LUCIA					2368	5385	3370	5357.				2F
SQUTH AKERIC 271 175 174 166 1117 8517 8365 8468 3008 1488 1455 1404	IRINIDAD TOS		ì			3366	10370	10345	10345	2	3 4	35	3F
ARGENTINA A4 34 34 34 24 10410 9527 8988 10107 457 322 302 247 BOLIVIA BOLIVIA 2 3 3 3 3F 7471 4888 5362 5000 12 14 15 15 BRAZIL 183 92 90F 90F 11775 848 8889 8889 2155 819 800F 800F CRILE 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			i l	1]		
## BOLLVIA 2 3 3 3 7471 4888 5362 5000 12 14 15 15 ## BRAZIL 183 92 90F 90F 1775 8448 8889 8889 2155 819 800F 800F ## CULADOR 3 1 1 1F 1574 5004 4398 4615 10 4 3 3 ## FR CULADOR 3 1 1 1F 3574 5004 4398 4615 10 4 3 3 ## R CULADOR 3 1 1 1F 3574 5004 4398 4615 10 4 3 3 ## FR CULADOR 10 14 14 15F 10710 7541 7785 7667 110 106 112 115 ## PERU			[]	. 1									
CHILE CHADOR CHA	BOLIVIA .	2	3	3 [3 F	7471	4888	5362	5000	12	14	15	15F 800F
FR CUIANA PARACUAY 10 14 14 15F 10710 7581 7785 7667 110 106 112 115 SURINAME URUCUAY 13 14 158 158 10710 7581 7785 7667 110 106 112 SURINAME URUCUAY 13 158 158 158 5975 4000 4000 78 60 60 60 60 60 60 60 60 60 60 60 60 60	CHILE	1	1 *	2.4	1.5	14000	7000	7000	7000	14	7 0	76	.7F
PERU 13 14 15 155 12502 10639 10000 9941 167 149 1508 151 3URINAME URUGUAY 13 158 158 158 5975 4000 4000 78 60 60 60 60 60 60 60 60 60 60 60 60 60	FR GUIANA					6570	5667	6000	5806	1	2	2 F	2#
URUGUAY VENERULLA 2 18 15 15 5975 4000 4000 78 608 608 609 VENERULLA 2 18 15 15 15 5975 4000 4000 78 608 608 609 VENERULLA 3 15 15 15 15 15 15 15 15 15 15 15 15 15	PERU					15205	10639	10000	9941				
VENEZUELA 2 18 15 15 2495 3573 3525 3583 4 4 455 45 45 45 45 45 45 45 45 45 45 4			15*			5975	4000	4000	4000		60*		60*
BAHRAIN 172 73 72 73 72 73 73 73 74 75 75 75 75 75 75 75 75 75	VENETUEL A			1		1		[4 4 4 4		i
BRUNE] BRUNE] BURNA 3 45 45 46 45 4551 4103 4154 4100 15 16F	BAHPATH					20000	10000	6667	6567				
BURNA 3 45 46 46 45 4951 4103 4154 4100 15 16F 16F 16 16 16 16 16 16 16 16 16 16 16 16 16		. 12	73	15	1 3F					830	1	16	LF.
EAST Y HOR 3 5933 10510 24353 27867 24000 15 15 15 15 15 15 15 15 15 15 15 15 15	BURHA					1951	4103	4154	4100				
INDIA 225 220 207 207F 10052 7381 6518 7246 2260 1623 1349 1500 110000051A 361 287 287 276 6131 7349 7655 7530 2215 2194 2193 2079	EAST YIMOR					5933		600		15			
15RAEL 134 64 65 65F 19356 21283 20324 20324 2590 1360 1317 1317	LHOTA					10052	7381	6518	7246	5560			1500F
	ISRAĘL .	·	ł i	1		20455	40000	40000	40000	}	į 15	1 F	15
well and the second of the second	JAPAN Kanpuchea on	134	25		- 65F 2F		21283 7368	20324 8333	20324	2590 21	1360		

Table B-6 (cont'd.)

1969-11 1979 1980 1981				AREA HAR	l v		1000 HA	AIEFO			KG/IIÁ	PRODUCT	(0)		1000 H
ROBER DOR 21 277 288 298 1326 2004 22160 2033 1387 1103 1103 1105 125 61 53 53 1387 1103 1103 125 13 54 54 54 54 54 54 54 5	,			1969-11	1979	1980	1981	1969-71	1979	1980	1981	1969-71	1979	1980	1981
KOREA DPR 122	<u> </u>			 			204	13226	13704	13393	13172				30
XOBAB REP 125 38 38 3931 3915 3955 9667 15 278 288	Κt	OREA DPR									22160	2053			110
MALAYSTA	K.	OREA REP		125								15	258	28F	i
MAL AYSIA 7 4				2 (30	365	34E	
NA DIVES				1 1	46	. 48)	***						74.174.1	1.500]
PARTSTAN 16 186 197 8000 4700 4710 4 444 468 680 1120 1040 11 51NGAPORE 51NGAPORE 51 14 12 13 1990 10502 12632 12632 4 14 17 17 17 17 17 17 17 18 18 18 18 18 18 18 18 18 18 18 18 18				{	i	ļ						141	162F	1635	1
DHILIPPINES 130 230 230 230 235 345 346 347 347 349 349 349 349 349 349				18	18F	18F									l ti
SINGAMORE SRI LANNA 17 16 14 171 3980 12604 8160 65 149 127 117 118 118 118 118 118 118 118 118 11					238	236 }	235F								1
SRILANNA 17 16 14 17 330 330 320 130 130 130 130 130 130 130 130 130 13				1				9380							
STATE CAND 1				أجذا	1.6	14 3	177	1001							
VIET NAM								8660	9254						
EUROPE 14 12 12 13 9807 10810 11234 10379 14C 135 138 1 EUROPE CREECE 1 1 1 1 117 1330 20398 21502 27 27 27 27 27 27 27 27 27 27 27 27 27			1					4977	3522	5324	6000	1108	5105	5320	43
GREECE 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	¥	KAN TSI		223	392	443	1001			(37)		ا م ، ، ا	116	129	Ι,
REECE 1 1 1 1 1 1 1 1 1	. E1	UROPE		14	12	12	13	9807	10810	11234	10379	140	132	130	•
GREECE TITALY 9 SF 9F 9F 9F 7782 7635 7794 7765 69 65 66F 90RTUGAL 9 SF 9F 9F 9782 7635 7794 7765 69 65 66F 90RTUGAL 9 SF 9F 9F 7782 7635 7794 7765 69 65 66F 90RTUGAL 9 SF 9F 9F 978 27635 7794 7765 69 65 66F 90RTUGAL 9 SF 9F 9F 978 2377 4890 13030 42 46 44 1529 16526 17480 13030 42 46 44 1529 16526 17480 13030 42 46 44 1529 13051 100000 100000 100000 100000 100000 100000 10000 100000 10000 10000 10000 100000 10000 10000 10000 10000 10				1		1		1 20 25	19005	18095	18095	2	+ř	AF	1
TRAIN	G	REECE		1. !		. !						27	21	23	[
PORTUGAL 9 9 9F 9F 9F 9F 9F 9F 1782 1780 13030 42 46 44 586 606 497 10000 10000 152 100000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 10000 152 100000 152 10000 152 100000 152 10000 152 100000 152 10000 152 100000 152 100000 152 100000 152 100000 152 100000 152 100000 152 100000 152 100000 152 100000 152 100000 152 100000 152 1000000 152 1000000 152 10000000000	- 1	TALY ·											65	68F	1
SPAIN 4 3 3 3 1122	ė	ORTUGAL		: 9											
AMER SANDA ANSTRALIA COOK ISLANDS I I IF IF IF IF IP OFFO GIAN NEW CALEDONIA NEW CALEDONIA NEW CALEDONIA SOLONDH IS SOLON				[3	3	31	11229	16250	17400	1,030			4.77	i
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^{*} Unofficial figuresF: Estimates by FAO

Table B-7 Ethylene Production

(million MT)

				<u> 1 2</u>					
	1960	1965	1970	1973	1974	1975	1976	1980	1985
EC	(0.68)	1.87	5,63	9.04	9.75	7,27	9.45		
Scandinavia	0.01	0.05	0.16	0.35	0.42	0.37	0.44	(12.0)	(15.3)
Other Euro- pean OECD	-		0.15	0.25	0.30	0.32	0.38		
Member Countries							-	and an order of the second of	
USA	2,47	4.34	8.39	10.13	10.84	9.30	9.99		
Canada	-	0.27	0.41	0.49	0.43	0,43	0.51	(13.5)	(18.0)
Japan	0.08	0.78	3.10	4.17	4.18	3.55	3.87	(4.7)	(6.1)
Other OECD Member Countries	-	~		•		<u>.</u>	· <u> </u>		- ·.
Councilies					• •	200			•
Subtotal	(3.3)	(7.3)	17.85	24.45	25.93	21.24	24.64	(30,2)	(39.4)
Others	-	-	-	•	-	(1.4)	_	(6.0)	(9.0)
Total	-	- ·	· · · · · · · · · · · · · · · · · · ·	<u>.</u>		(22.6)	 -	(36.2)	(48.4)

Note: 1) From the report by the Petrochemical Subcommittee of the Industrial Committee of OECD.

^{2) ():} estimates

2. Prices of Raw Materials

2.1 Raw Materials for Fermentation Process

As stated before, raw materials generally used for making ethyl alcohol by the fermentation process consist principally of saccharine materials like molasses and sugar cane and the starchy materials like corn, potato, cassava and sweet potato. But the prices at which these materials are made available to the ethyl alcohol breweries are not disclosed for the most part. Moreover, the prices fluctuate widely from region to region and from year to year.

According to the survey of the Economic and Social Commission for Asia and the Pacific undertaken in 1982 with respect to the prices of molasses, sugar cane and cassava, considerable differences were seen in the raw material prices in various parts of the world, as shown in Table B-8.

Table B-8 Cost of Potential Ethyl Alcohol Feedstocks (1982)

	•		(US\$/MT)
	Molasses	Sugar Cane	Cassava
Australia	65	24	34 (est.)
India	7	17	27
Indonesia	120		27
Philippines	6,490	19	40
Sri Lanka	100 (est.)	30	and the section of
Thailand	76	31	33
Brazil	et e	12	

These raw materials are all farm products or their by-products having a wide range of applications besides being used for ethyl alcohol production. Thus it seems that a big difference in raw material acquisition cost tends to be generated even within the same country depending on the location of each alcohol brewery because of the difference in such factors as:

- Mode of raw material procurement (captively operated farm or not)
- Competition with prices of the same material for other applications
- Agricultural productivity
- Mode of assembling
- National policy, etc.

In working out any concrete plant locating program, as a consequence, detailed investigations into the stable availability and prices of raw material supplies at each particular location would be necessary. Incidentally, changes in the prices of molasses, the most generally used saccharine raw material, between 1948 and 1978 are shown in Fig. B-1. Wide fluctuations are evident in the figure.

2.2 Raw Materials for the Synthesis Process

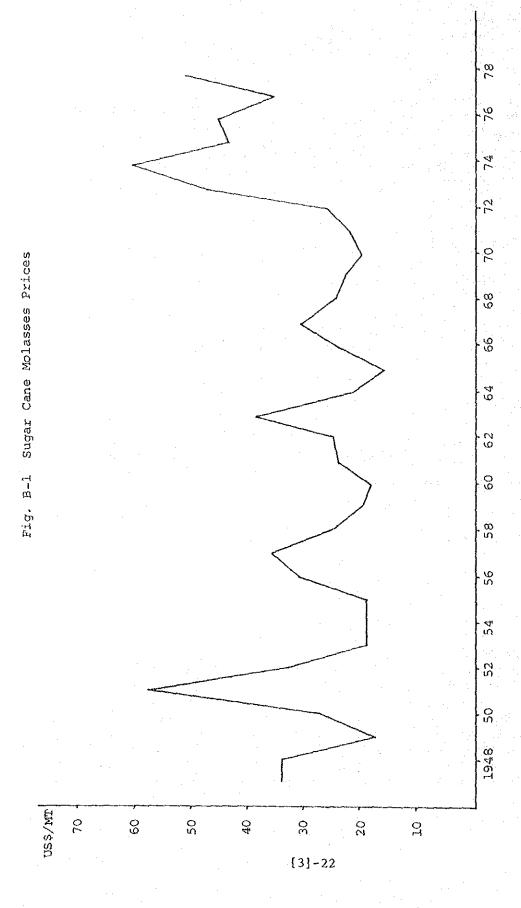
Ethylene is the raw material used for ethyl alcohol synthesis process. Ethylene, which is the basic raw material for the petrochemical industry, is produced in Japan and Europe largely from naphtha, a fraction of the crude oil. In the United States it is mostly produced from natural gas.

Fig. B-2 presents price fluctuations between 1973 and 1980. It is clear from the figure that the prices of ethylene have sharply been affected by the soaring in the crude oil price. It is also evident that the price of ethylene in the United States where natural gas is used as a raw material for the most part is lower than that in Japan and Europe.

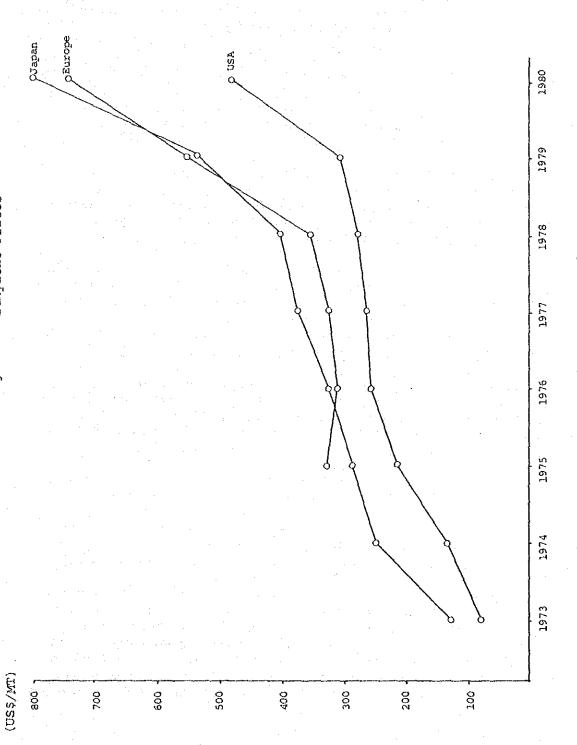
III. Production Technology

1. Principal Production Processes

Ethyl alcohol production processes are divided into the fermentation and synthesis methods. The processes by the fermentation method differ when the types of a raw material differ. Also, the production processes by the fermentation method differ somewhat according to the end-uses of ethyl alcohol produced. This means the process for manufacturing industrial ethyl alcohol differs somewhat from that for manufacturing fuel alcohol. And as stated before, the ethyl alcohol purity is about on the same level for every product specifications, standardization of fuel ethyl alcohol specifications has not made much progress yet excepting the items



Source: 1947-1969; New Orleans Market 1970-1978; Cuba-Azucar-Mincex Estimates



Source: Japan; Chemistry & Economy, Aug. 1980, Aug. 1981

Burope; Europe Chemical News
USA; Chemical Marcketing Reporter

related to purity and organic acid content. For fuel ethyl alcohol, the general practice is to simplify the distillation process with a view to reduce the manufacturing cost as much as possible. Other processes, however, are followed in much the same way both for industrial and fuel ethyl alcohol manufacture.

While the leading processes will be outlined in the following, this section will be devoted to the generally practiced processes, leaving matters connected with new technologies in particular to subsection III-3 on technological innovations.

1.1 Fermentation Process - Using Molasses as Raw Material

As total sugar content of molasses used as raw material ranges from 50 to 60%, it is diluted with hot water in the pre-treatment unit.

Molasses is also diluted with hot water when it is used for cultivating the yeast. It is then sterilized to become a sterilized mash. To this is added some nutrient like ammonium sulfate, which nourishes the yeast. Then mixing the yeast into the mash, it is cultivated as the seed at about 33°C.

Cultivated yeast is added to the molasses, the raw material, to cause fermention. This fermented raw material is known as the mash, whose total sugar content has been diluted to about 23%. When the temperature is maintained for four days at 31 to 33°C for ripening, about 12.5% ethyl alcohol is formed by the work of the yeast. This is known as the broth.

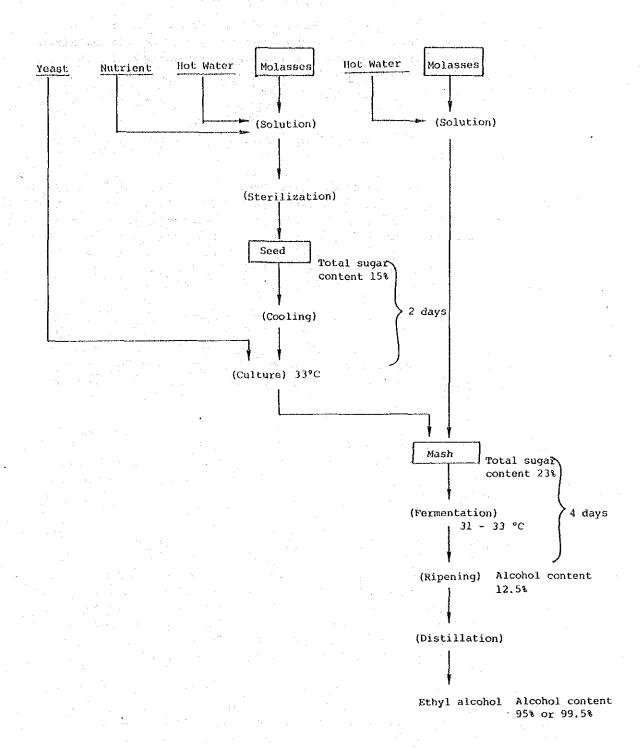
This broth is refined by the distillation process to obtain hydrous alcohol of 95% purity. To obtain anhydrous ethyl alcohol of 99.5% purity, dehydrating distillation process is executed using dehydrating agents like benzene and cyclohexane. Refer to Fig. B-3 for the process flow.

1.2 Fermentation Process - Using Sugar Cane as Raw Material

Sugar cane used as raw material is crashed to obtain juice. Milk of lime is added to the juice during the clarifying process. When it matures, it is divided into clarified juice and sludge.

To this processed clarified juice are added the nutrient and the yeast solution sterilized by means of acid, and the mixture is allowed to go through fermentation for 15 to 18 hours at a temperature of 34 to 37°C. The ethyl alcohol concentration of the fermented mash thus obtained is about 8.5 vol %. This mash is sent to the yeast separator where the yeast is recovered, and after it is treated with acid, it is used as the seed for the next fermentation process.

Fig. B-3 Ethyl Alcohol Production Process Using Molasses as Raw Material



Source: Alcohol Handbook

The mash from which the yeast has been separated is then refined by distillation and 95% hydrous ethyl alcohol is obtained. For obtaining anhydrous ethyl alcohol of 99.5% purity, further dehydration distilling using benzene and cyclohexane is executed. Refer to Fig. B-4 for the process flow.

1.3 Fermentation Process - Using Cassava and Sweet Potato as Raw Material

The raw material is washed and crushed. The crushed material is sent to the cooker in the slurry form. The slurry sent to the cooker is given liquefying enzyme as additive and is cooked while steam is applied. After this, it is cooled down and partly sent to the seed tank. Here saccharifying enzyme is added at 55°C and it is cooled down further. At 33°C yeast is added and the mixture is cultivated for 24 hours to obtain the seed.

The remainder is transferred to the fermentation tank, where it is saccharified by adding liquefying and saccharifying enzyme. After saccharification, it is cooled down to 33°C, and adding the seed to it, it is allowed to ferment. The treated mash with the alcohol content of 8.5 vol % is run through the screen filter to remove the sludge.

The treated mash is refined by means of distillation and 95% hydrous ethyl alcohol is obtained. Anhydrous ethyl alcohol is obtained by putting the hydrous ethyl alcohol through dehydration distillation using some dehydrating agent. See Fig. B-5 for the process flow.

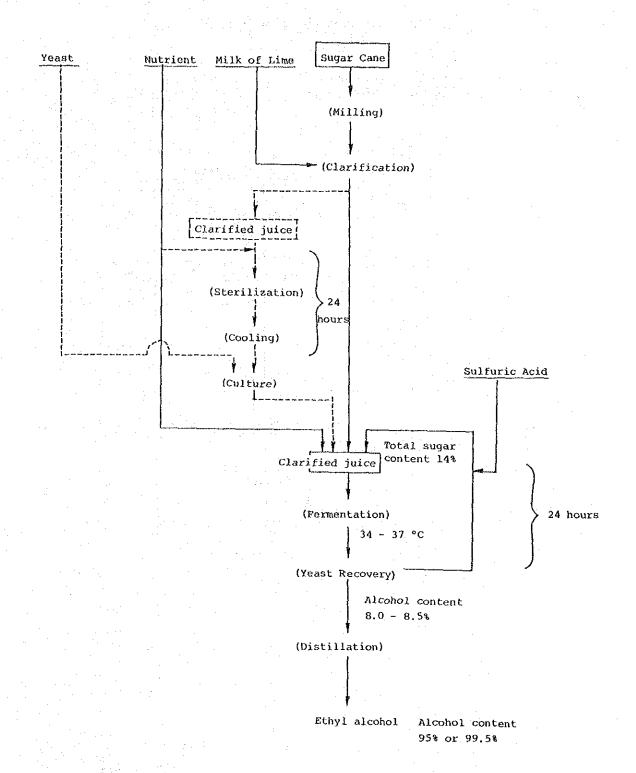
1.4 Synthesis Process Using Ethylene as Raw Material

The feed ethylene is compressed to 70 kg/cm²G with a compressor and is heated and fed to the reaction column. Water is added in the reaction column, and the phosphoric acid catalyst is used to bring about the reaction. The product of reaction is isolated as gas, while ethylene that had failed to react is sent to the preliminary stage of the reaction column for recycling.

Crude ethyl alcohol obtained by the reaction process account from 20 to 25 wt %. This crude ethyl alcohol is fractionally distilled at the fractional distilling column and is sent to the reduction column after fractional distillation.

At the reduction column, the aldehyde content of the crude ethyl alcohol is reduced under high-pressure hydrogen using nickel catalyst to become ethyl alcohol, which is fed into the refining column. Finished alcohol of 95% purity is obtained from the top of the refining column. As in the case of the fermentation

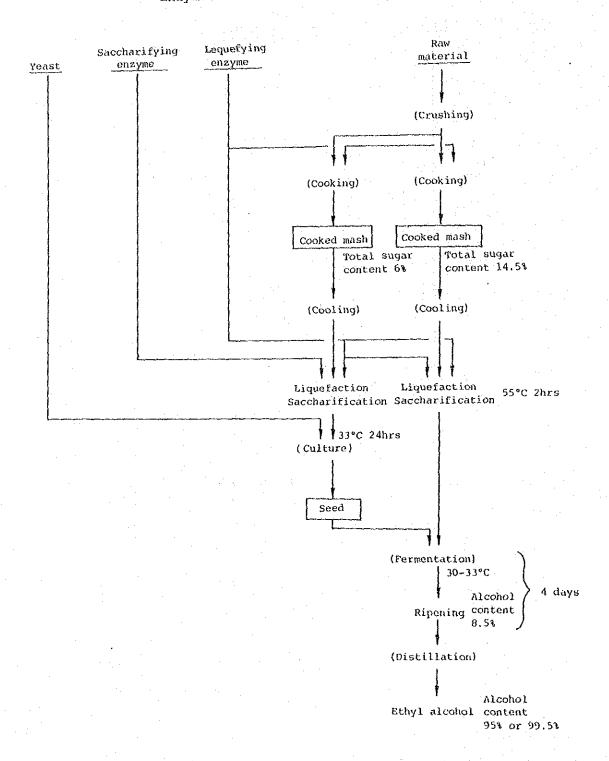
Fig. B-4 Ethyl Alcohol Production Process Using Sugar Cane as Raw Material



Source: Prepared by the Study Team

Fig. B-5 Ethyl Alcohol Production Process Using Starchy as Raw Material (Raw Sweet Potato and Cassava)

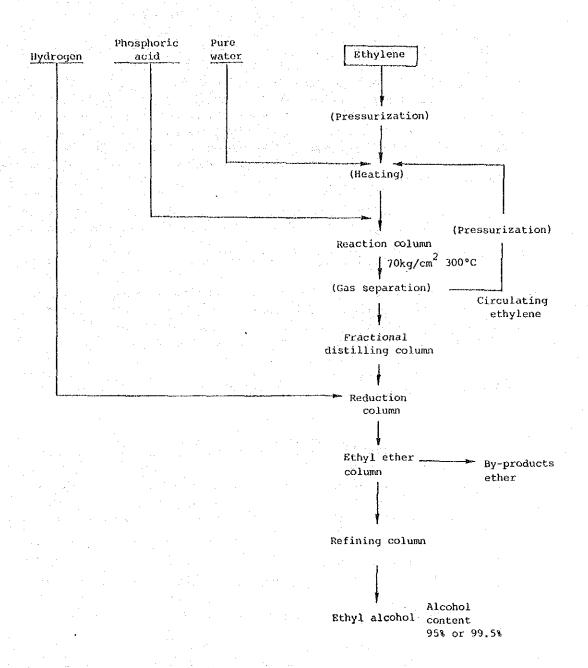
- Enzyme saccharification method -



Source: Alcohol Handbook

Fig. B-6 Ethyl Alcohol Production Process Using Ethylene as Raw Material

- Direct hydration method -



Source: Alcohol Handbook

process, 99.5% anhydrous alcohol is obatined by dehydration distillation using benzene and cyclohexane.

2. Energy Balance

Ever since ethyl alcohol began to be used as fuel, the balance of the output energy against the input energy (energy balance) has been attracting attention. The energy balance means the ratio between the energy expended in cultivating raw materials and manufacturing ethyl alcohol and the energy inherent in the ethyl alcohol produced. NER (net energy ratio) is generally used as the assessing index.

Large number of reports on energy balance involving various raw materials, such as those of Battele and da Silva (Brazil), have been made, but the results of calculations are not uniform because of differences in the premises, such as farm production, production technology and so forth, that constitute bases for calculations. An example of comparison between sugar cane and cassava as undertaken by the Study Team is given in Table B-9.

Table B-9 Energy Balance

			(kca1/l)
		Sugar Cane	Cassava
Input	Farm Production Alcohol Production Process Total	(280) (15)* 295	(415) (5,880) 6,295
Output	Alcohol	5,045	5,045
	NER	0.06	1.25

^{*} In the case of sugar cane, input energy for alcohol production process is low as the bagasse derived from the stems is usable as fuel.

NER > 1 means minus energy balance, and NER closer to zero means acquisition of greater energy. In the table, cassava NER of 1.25, which is higher than 1, stems from the need for using outside heat source like burning wood in the production process.

As NER > 1 offers no more than a qualitative conversion to liquid fuel for ease of use as merit, research and development efforts related to energy-saving processes are specially brisk as the ethyl alcohol production process requires large energy consumption.

Also, in case large quantities of fertilizer, mechanized farming and long distance haul for assembling are required in the raw material production process, energy balance deteriorates. These reasons call for detailed study of the conditions of location at the time of drafting concrete plans.

3. Technical Innovations

The areas in which technological innovations are expected in connection with ethyl alcohol production consist of raw material cultivation and ethyl alcohol production techniques.

3.1 Raw Material Cultivation

Among farm products now used as raw material for ethyl alcohol, sugar cane as material for making sugar and corn as feed claim mass cultivation history. On the other hand, it was only in recent years that cassava began to attract attention, and there is room left for improving the methods of cultivating these crops.

Principal points needing development in the cultivation of these crops include.

- Development of high-yielding varieties
- Boosting of harvests through improved fertilizer administration
- Better protection from plant diseases and harmful insects

According to the report issued by the World Bank, the average yields of cassava and sweet potato are respectively 9 - 12 tons/ha and 15 tons/ha, but improvements in the above-cited points are expected to boost the yields.

3.2 Ethyl Alcohol Production Technology

As ethyl alcohol production technology, as stated before, two processes are available, one of which is the fermentation and the other is synthesis processes. Technological innovations expected in the synthesis process include development of a technology for producing ethyl alcohol direct from natural gas, development of new catalysts in the production of ethyl alcohol from ethylene, and development of some new ethyl alcohol separation technology.

The technology for producing ethyl alcohol directly from ethylene is now in the process of research and development; it is a technology of the future as yet. Development of new catalysts in the production of ethyl alcohol from ethylene has little prospects as no research work is undertaken as of now. As for the new ethyl alcohol separation technology common to both the synthesis and fermentation process, it will be discussed under the fermentation process.

Technological innovations expected in the ethyl alcohol production under the fermentation process consist mainly of the reduction in the ethyl alcohol conversion energy and in the cost of production. In the following some of the technological developments undertaken in Japan will be discussed with respect to different raw materials and processes.

3.2.1 Bio-saccarification ethyl alcohol production technology using cellurose as raw material

Cellurose used as raw material has to be saccharified before the fermentation process. As means of facilitating saccharification, physical and chemical pre-treatments are conceivable and in Japan, various combinations of electron beam application, pulverization, blasting, ozone treatment and alkali treatment are under study for commercialization.

3.2.2 Saccharification process of starchy materials

(1) Non-cooking saccharification process

In the case of a starchy raw material, about 30% of steam needed for production is consumed as energy for cooking. For this reason, any means of saccharification without cooking will save much energy. When fresh sweet potato is used as raw material, saccharification at 80°C has successfully been achieved on the industrial level. Under the existing state of things, however, enzyme requirement for the process is somewhat larger than for the cooking process. With some room for further technological development in this respect,

this process has already reached the commercialization stage as a feasible energy conservation technique.

(2) Continuous saccharification with process immobilized enzymes

In manufacturing insomerized sugar by converting corn starch into fruit sugar (fructose), continuous saccharification by immobilized enzyme has already been commercialized. In manufacturing ethyl alcohol from uncooked sweet potato as the starting material, enzymes are still mixed into the cooked mash to bring about saccharification at present, and therefore it is possible to repeatedly use enzymes by immobilizing them. Depending on the raw material, however, clogging of fibrous substance may sometimes occur. Commercialization of continuous saccharification process by solving this problem is much awaited.

3.2.3 Saccharification process of cellulosic material

As the saccharinity of the saccharified mash of cellurosic material is low, technology for upgrading the saccharinity by combining the excess filter film and the reverse infiltration film, technology for recovering enzymes by fixation, development of new enzymes with higher saccharification potential and other technological developments are underway for commercializing ethyl alcohol production from cellulosic materials.

3.2.4 Fermentation unit

(1) Immobilized yeast process

Immobilization of yeast makes continuous fermentation process possible. And since this upgrades the capacity efficiency of the fermentation tank, cuts in the cost of fermentation tank installation as well as saving in energy can be expected. In Japan, two processes of either using calcium alginate gel or photo-curing resin film as support have been researched, with each has of them having undergone pilot operation for an extended period.

(2) Vacuum fermentation

Under normal fermentation the activity of yeast is impeded by the ethyl alcohol generated. In this process, vacuum fermentation process is promoted while removing the generated ethyl alcohol by reducing the pressure. By this process, the capacity efficiency of the fermentation tank can be improved and high-concentration feeding is made possible. As this process has further merits like energy saving in the

distillation process, technological development in the pilot stage is now being pursued.

3.2.5 Distillation unit

(1) Pressurized distillation

Pressure is applied to one column to raise the temperature of the steam generated in that column so that it can be used as the heat source for other columns. This technology makes energy saving possible. In Japan this technology is in the pilot stage and is expected to be commercialized in near future.

(2) Membrane utilization

Unlike the ethyl alcohol separation process under conventional distillation, this is a technology for promoting ethyl alcohol concentration by the use of membrance. It is still under research now, and still more time is likely to be needed for commercial use.

4. Plant Investment Cost

Estimation of plant investment cost should properly be calculated on the basis of estimates collected after choosing the location and undertaking detail design. However, because of the aims of this Study, no such calculations have been undertaken, but the reports issued by the World Bank and data published in various literatures have been used as reference. Estimations of the Study Team have also been used as part of data, as shown in Table B-10.

Table B-10 Plant Investment Cost by Raw Material

(US\$ million)

Fermentation Process	Molasses Sugar Cane Cassava	Fuel Industrial Fuel Fuel	9.8 11.0 14.3 11.5
Synthesis Process	Ethylene	Industrial	17.8

Notes:

- 1) Plant capacity of 120 kl per day.
- 2) 1982 as a base year.
- 3) Finished product will be anhydrous grade.

IV. Production Cost

Specific production costs of any operating plant have not been made public because of its nature. As production costs differ according to the raw material prices, plant investment cost and labor cost prevailing at each different location, detailed study should be made after picking specific location, collecting various estimates on facility investment cost and making concrete calculations. In this Study, however, trial calculations were undertaken on the following general and standard premises in order to grasp the trends in production cost for different raw materials.

1. Premises for Cost Estimation

1.1 Matters Examined

Molasses, sugar cane and cassava were examined as raw materials used in the fermentation process, while ethylene was assumed as material for synthesis. In view of the existing state of things, uses were divided into industrial and fuel uses with respect to molasses. Sugar cane and cassava were considered as material for making fuel ethyl alcohol only, while ethylene was considered as material for making industrial ethyl alcohol only. With regard to sugar cane, the case of Brazil was taken up as an example of a country where raw material cost was low.

Also for ethyl alcohol production facilities, a newly installed plant was assumed for all cases.

	Date Made and m3	Annliantion
and the second	Raw Material	Application
Case A	Molasses	Fuel
Case B	Molasses	Industrial
Case C	Sugar cane	Fuel
Case D	Sugar cane (Brazil)	Fuel
Case E	Cassava	Fuel
Case F	Ethylene	Industrial

1.2 Plant Capacity

The most commonly used 120 kl/day capacity is adopted.

1.3 Product Purity

The anhydrous grade of 99.5 vol % has been used both for industrial and fuel ethyl alcohol. With regard to the impurity content, specifications generally used in the developed countries

were adopted for the industrial ethyl alcohol, but no provisions were made for the impurities in the fuel entyl alcohol, for which only product purity was considered.

1.4 Operating Days

While climatic conditions in the various parts of the world may lead to different operating days with respect to farm products used as raw materials under the fermentation process, plant operating days were set at 300 days a year, assuming that the materials would be available for 300 days. With respect to the synthesis process, 330 days a year were assumed according to Japan's case.

1.5 Annual Production Rate

The annual production rate is expressed by multiplying the per day production capacity of the plant by the number of days a year operated.

- Fermentation process: Annual production rate of 36,000 kl
- Synthesis process: Annual production rate of 39,600 kl

1.6 Raw Material Cost

Unit price of each raw material is as given in the costs of raw materials.

- Raw material for fermentation process:
 Average price in various parts of the world in 1982.
- Raw material cost for synthesis process:
 Average price in Japan, Europe and the United States
 in 1980.
- Premised on the above, the unit prices will be:

Molasses: \$74 per ton
Sugar cane: \$22 per ton
Sugar cane (Brazil): \$12 per ton
Cassava: \$32 per ton
Ethylene: \$680 per ton

The unit consumption rate required for producing 1 kl of ethyl alcohol from each of the raw materials was quoted from Chemical Engineering (No. 5, Vol. 45), but the cost of cassava was estimated by the Study Team.

1.7 Utility Cost, Chemical Cost

The unit cost and quantity for calculating the utility cost and the chemical cost for each raw material are quoted from Chemical Engineering (No. 5, Vol. 45).

1.8 Labor Cost

The unit cost of labor and required number of workers have been calculated on the basis of data the Study Team has compiled for the case of Southeast Asia. The unit cost has been set at \$2,000 per man-year, and the required number of workers was set at 60 for the case of plant producing fuel ethyl alcohol and at 80 for the plant producing industrial ethyl alcohol.

1.9 Maintenance Cost

Annual maintenance cost is assumed to be 3% of the plant investment which is a normal rate for chemical plants. For the plant investment cost, values cited in the previous section III-4 have been adopted.

1.10 Taxes, Insurance

These have been estimated at 2% of the plant investment cost normally used for chemical plants.

1.11 Depreciation

Taking the residual value as 10%, the period of depreciation has been set at 12 years. As a consequence, the depreciation cost was calculated as plant investment cost \times 0.9 \times 1/12.

1.12 Profit

Estimated at plant investment cost x 0.1.

1.13 Management Cost

Labor cost x 80% has been estimated as cost needed for plant management.

2. Results of Tentative Production Cost Calculations

The results of tentative calculations of production cost on the premise cited in 1 are shown in Table B-11. Case D, production cost in Brazil, is shown in Table B-12. Details for other cases are given in Appendix Table 3.

Table B-11 Production Cost Estimate by Raw Materials

(US\$/k1)

		Variab]	Le Costs	Fixed	
Method		Material Cost	Utility Cost, etc.	Cost	duction Cost
Fermenta- tion Process	Molasses - Fuel Molasses - Industrial Sugar cane - Fuel Sugar cane - (Brazil) - Fuel Cassava - Fuel	259 (63) 259 (62) 330 (77) 184 (66) 208 (48)	73 76 4 4	77 80 95 95	402 412 429 279 427
Systhesis Process	Ethylene - Industrial	326 (63)	85	107	518

Note: Figures in () indicate the percentage share of raw material cost in the production cost.

As stated at the beginning of this section, the result of production cost estimations cited here are based on general premises. In undertaking concrete location planning, therefore, comparative study of each item of premises with actual conditions and modifications would be necessary. In particular, as the ratio of raw material cost to the total cost is very high, as is evident from the table, stable material procurement price at the chosen location becomes highly significant.

While a lagoon, for instance, has been assumed for waste water treatment in the premises for pollutant treatment facilities, the cost naturally becomes higher in a country like Japan where regulations on environmental pollution are stringent.

In any case, under the premises cited herein, the following points can be gleaned from the table as general trends:

Table B-12 Production Cost Estimate for Ethyl Alcohol by a Newly Installed Plant in Brazil

Ethyl Alcohol Production Cost

Plant capacity: Product purity: Raw Material: 120 kl/day 99.5 vol % Sugar cane

End-use:

Fuel

ro	duction Cost Item	Unit Price	Quantity	US\$/k
	Raw material cost	\$12/MT	15 MT/kl	180
(S	Utilities costs			
ง เก	Power	·		. 0
ي ص	Fuel	-		0
d Q	Water	¢20/MT	10 MT/kl	. 2
Variable Costs	Chemical cost	•		2
	Subtotal			184
	Labor cost	\$2,000/y/man x 60	men =	\$120,000/y
• • •	Maintenance cost	\$14,300,000 x 0.03	=	\$429,000/y
ง	Taxes and insurance	\$14,300,000 x 0.02	= 1	\$286,000/y
8) 8)	Depreciation charges	\$14,300,000 x 0.9	x 1/12 = \$	1,073,000/y
rıxed	Profit	\$14,300,000 x 0.1	≃ \$	1,430,000/y
T4	Management cost	Labor cost x 0.8	12	\$96,000/y
٠	Subtotal			95
	Total production cost			279

- Comparison of the fermentation and synthesis processes give the edge to the fermentation process in the aspect of production cost.
- Not much difference is observed in the costs of molasses, sugar cane and cassava used as raw materials for the fermentation process.
- In each case, the ratio of raw material cost to the total production cost is more than 50%, indicating that the total production cost is greatly affected by the raw material cost.

V. Production Policy

Policies related to ethyl alcohol production, especially to fuel ethyl alcohol production, have been lately made public by various countries as ideas or as measures for implementation. In this report policies under implementation in Brazil, the United States and the Philippines will be discussed.

1. Brazil

Findings of the interview survey conducted in march 1983 with respect to the preferential measures for promoting the "National Alcohol Program" are summarized below:

- 1.1 Low Interest Loans for Industrial and Agricultural Projects
 - a. Industrial project
 - (a) Loan ceiling 70 to 90%
 - (b) Interest rate
 Middle and southern Brazil: Inflation rate x 0.8 + 5%
 Northeastern Brazil: Inflation rate x 0.6 + 5%
 - (c) Repayment period 12 year repayment (4 year grace period)

b. Agricultural project

- (a) Loan ceiling 50 to 100%
- (b) Interest rate
 Middle and southern Brazil: Inflation rate x 0.7 + 5%
 Northeastern Brazil: Inflation rate x 0.55 + 5%
- (c) Repayment period 12 year repayment (4 year grace period)

c. Incentive for ethyl alcohol production

Producers will be guaranteed a parity purchase price.

2. The United States

The "Alcohol Fuel Policy Study Program" which the Department of Energy announced in July 1978 and the "Alcohol Fuel Policy Review" announced by the same department in June 1979 indicate the production policy as follows:

- The Department of Agriculture will provide loan guarantee for pilot plants for ethyl alcohol.
- The Department of Energy will provide conditional loan guarantees for the ethyl alcohol production plants.
- When an ethyl alcohol production plant is newly established, the 20% investment tax under the "Energy Tax Law" will be exempted.

3. The Philippines

The revised Alcogas Program which the Philippine National Alcohol Commission (PNAC) announced in 1982 provides the following policies on production:

- Acceleration of the depreciation period of the equipment in the ethyl alcohol plant will be given special approval.
- Import duties on the imported equipment used in the ethyl alcohol plant will be exempted.

VI. Changes in World Production

Although several reports are available on world production of ethyl alcohol for certain years, it is only the United Nations' statistics that offers complete data on year-to-year changes in production. However, as the following problems are associated with the United Nations statistics, the Study Team modified some of the production figures.

- As the French production was under-reported, figures estimated by Japan's Ministry of International Trade and Industry was used instead.
- As production for certain years was not surveyed, production in each of these years was assumed to be equal to each of the nearest fiscal year surveyed. For the Brazilian production

over 1977 to 1980, the figures obtained from the Comissão Nacional do Álcool of Brazil were used.

- The production volumes reported by some countries may presumably include ethyl alcohol used as material for liquors, but as such quantities are unknown, no corrections have been made.

- The United Nations statistics do not make clear distinction between industrial use and fuel use, but Brazilian production of fuel ethyl alcohol is represented by figures obtained from the institute do Acúcar e do Alcool and they are recapitulated in a graph as the component of total production. Incidentally, while countries producing fuel ethyl alcohol as of 1983 are Brazil, the United States and the Philippines, Brazil is the only country that was producing fuel alcohol before 1980. Industrial ethyl alcohol production volumes are shown in Fig. B-7 and Table B-13.

The average annual growth rates in production between 1965 and 1980 was 0.9% for industrial ethyl alcohol and 16.5% for fuel ethyl alcohol (Brazil).

VII. Fuel Ethyl Alcohol Production Programs

With regard to fuel ethyl alcohol which is attracting attention lately, a number of countries have announced production programs or future directions of development. Situations in the leading countries are as given below.

l. Brazil

The national alcohol program of Brazil was adopted and put to implementation in 1975. The production plan of the "Proalcool" calls for the following annual production schedule:

1980:	4.0	million	13
1981:	4.9	million	kl
1982:	6.0	million	kl
1983:	7.3	million	k 1
1984:	8.9	million.	k1
1985:	10.7	million	kl

The programs were implemented just about on schedule up to 1981, but according to information obtained from CENAL (the Comissão Nacional do Álcool of the Commerce and Industry Department) as of November 1982, the projected 1985 production has been revised to 8.3 million kl, so that the program is now somewhat behind schedule.

Fig. B-7 Changes in Ethyl Alcohol Production (including fuel ethyl alcohol)

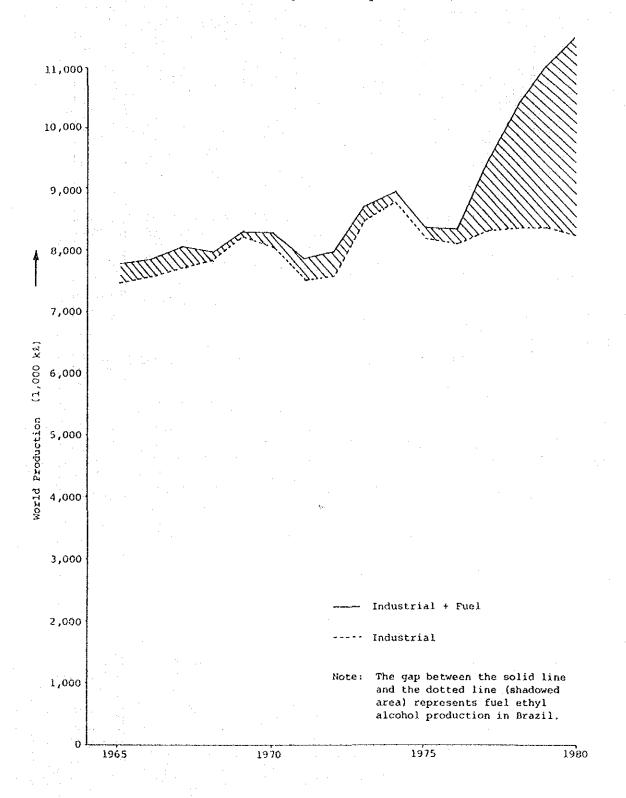


Table B-13 Changes in Production of Ethyl Alcohol (including Brazil's fuel ethyl alcohol)

															(1,000	K1)
	1965	1966	1961	1968	1969	1970	1971	1972	1973	1974	1975	1976	1977	1978	1979	1980
Developed Countries																
USA	2,632	2,602	2,571	2,590	2,818	2,619	2,158	2,154	2,573	2,451	2,199	1,891	1,886	(1,886)	(1,886)	(1,886)
[USA's fuel use]	ł		•	1	1	ŧ	1		ı	t	ı	1	1	1	[26]	[346]
Canada	19	20	25	50	30	28	40	47	27	36	23	23	24	(24)	(24)	(24)
France	(301)	(301)	(301)	(301)	(301)	(301)	(301)	(301)	301	2.70	214	241	297	(297)	(297)	(297)
Germany, FR	(195)	(195)	(195)	(195)	195	222	219	192	214	243	183	215	212	182	182	218
Italy	136	113	117	157	169	176	185	149	179	238	235	(235)	(235)	(235)	(235)	(235)
UK	177	155	143	145	165	175	167	156	186	295	98	249	272	297	280	(280)
5 EC Countries	(137)	137	131	153	164	164	170	173	192	206	196	209	216	203	(203)	(203)
Northern Europe	149	148	152	145	137	124	332	135	130	128	118	101	0.	(101)	109	66
Other European	122	158	123	147	194	102	75	164	135	162	170	(170)	(170)	(170)	310	328
Japan	193	197	230	228	661	207	229	229	240	237	208	222	222	228	213	211
Developing Countries																
Mexico	73	56	33	(33)	5		9	10	55	8	59	53	76	79	92	6
Other CA Countries	245	202	249	235	206	(206)	(206)	(206)	(206)	(206)	(306)	(206)	(206)	(206)	(206)	(206)
Brazil	572	691	584	461	450		625	684	653	615	580	642	(1,530)	(2,491)	(3,070)	(3,440)
[Brazil's fuel use]	[283]	[282]	[347]	[112]	[62]	_	[344]	[376]	[247]	[147]	[176]	(244)	[1,172]	[2,050]	[2,593]	[3,009]
Colombia	S	15	12	13	~		(21)	17	ე	33	(32)	(32)	(32)	(32)	13	(13)
Other SA Countries	❖	ιĊ	4	3	(5		(4)	(4)	(4)	(4)	(4)	(4)	(4)	(4)	7	(4)
Philippines	24	32	32	32	36		(33)	39	47	(47)	(41)	(41)	(47)	(47)	(47)	(47)
Korea, Rep. of	9₹	Ę	46	4	47		9	61	73	83	114	122	132	129	126	130
Middle East	-	-	7	(2)	(2)		m	ĸ	(3)	(3)	(3)	<u>ල</u>	(3)	(3)	<u>e</u>	(e)
Africa	36	26	26	58	36		ტ დ	37	38	(38)	(38)	(38)	(38)	(38)	(38)	(38)
Planned Economy Countries	ri es										٠.					
USSR	2,360	2,430	2,626	2,670	2,732	2,796	2,731	2,740	2,935	3,143	3.128	3,168	3,255	Ċ	3,118	M
China	30	(0)	2	24	20	<u>τ</u>	(18)	(18)	(18)	(18)	(81)	(18)	(18)	13	28	
Poland	208	183	198	206	189	190	185	181	249	250	227	262	330		327	
Others	40		163	Į.	163	170	265	273	288	243	262	284	260	259	(259)	259
Total for industrial	7,793	7,872	8,084	7,994	8,322	8,320	7,928	8,017	8,765	9,003	8,465	8,438	695'6	10,511	11,141	11,602
TOTOTO TOTOTO TOTOTO				*						:						
ethyl alcohol*	7,510	7,510 7,590	7,737	7,882	8,260	8,091	7,584	7,641	8,518	8,856	8,289	8,194	8,397	8,461	8,472	8,347

^{*} Total for industrial ethyl alcohol excludes fuel ethyl alcohol produced by Brazil and the U.S.

Source: United Nations, Statistical Yearbook

CA: Central America, SA: South America

2. The United States

In the United States, the Department of Energy announced the Alcohol Fuels Policy Study Program in July 1978, followed in June 1979 by the Department's Report of the Alcohol Fuels Policy Review, which together formed the basis for the country's full-fledged fuel ethyl alcohol production policy. The Report of the Alcohol Fuels Policy Review which envisages a rapid increase in production calls for the following fuel ethyl alcohol production program.

1979:	80 million gallons	304,000 kl
1980:	100 million gallons	379,000 kl
1982:	300 million gallons	1,136,000 kl
1985:	500 million ∿	1,893,000 ∿
	600 million gallons	2,271,000 ki

However, the 1981 production of 265,000 kl reported by SRI indicates some delay in the progress of the program.

3. The Philippines

The Philippine National Alcohol Commission (PNAC) was set up in February 1980 under the President's order. Alcogas has been positioned as non-conventional type energy in the five-year energy program or ten-year energy program annually announced by the Ministry of Energy. The Alcogas Program, in fact, made a flourishing start in 1980 as the most important item for development stipulated in the five-year program. However, the alcogas production target has been scaled down to about 1/50 in the five-year energy program drafted two years later in 1982. Comparison of the 1980 and 1982 five-year energy programs is presented in Table B-14.

Table B-14 Philippine Ethyl Alcohol Production Program

(1,000 barrels of oil equivalent)

	1980 Program	1982 Program
1981	96	
1982	576	35
1983	1,408	47
1984	1,964	47
1985	2,415	47
1986	- -	47
1987	· .	47

Source: Philippine Five-Year Energy Program

According to information obtained from the Ministry of Energy in January 1983, the curtailment of the program was made for the following reasons:

- The crude oil supply and demand situation has changed from that of the time when the program was announced.

The government subsidies (about US¢18/1 at 9.2 pesos = US\$1) have become a fiscal issue.

Various Asian Countries

4.1 Thailand

In March 1979, the Thai Government inaugurated Power Alcohol Commission (PAC) consisting of representatives of the various government departments and agencies.

The Government plans to develop technology for producing ethyl alcohol as a future energy from cassava which is now grown and exported as animal feeds. Since 1980, development efforts for ethyl alcohol production technology have been made with use of cassava and other plants at the Science and Technology Research Institute of Thailand as a joint research and development cooperation project with Japan.

4.2 Indonesia

The Indonesian Government has made it clear to vigorously implement a policy of linking ethyl alcohol production with dispersion of population to local areas with the Agency for Development and Application of Technology (BPPT) as the principal organization.

No specific production plan has been announced yet, although the State Alcohol Commission was established in June 1980.

To develop technology and provide education on ethyl alcohol production, two pilot plants were built in the southern part of Sumatra: 5,000 kl/year pilot plant constructed by indigenous technology and 8 kl/day pilot plant with technical cooperation from Japan. Both plants are scheduled to start operation one after another within 1983.

5. Various European Countries

Various countries in Europe are making plans to use either ethyl alcohol or methyl alcohol as fuel by taking advantage of their respective available resources.

Particularly, France announced the Caburol plan in January 1981 which calls for meeting 25 to 50% of total gasoline consumption with Caburol in 1990 as the target.

Federal Republic of Germany, Sweden and Italy have already been conducting road tests on automobiles with fuel that is produced by mixing entryl or methyl alcohol with either gasoline or gas oil.

C. CONSUMPTION

I. Use

Use of ethyl alcohol can broadly be divided into industrial and fuel use. For industrial use, explanation will be made about specific areas in which ethyl alcohol is used at present. Regarding fuel use, its suitability in each particular application will be discussed.

Before explanation is given on both industrial and fuel applications, denaturation, a special factor to ethyl alcohol, will be described.

Denaturation

Ethyl alcohol has special properties that it will become alcoholic drinks in addition to its industrial and fuel uses.

Since liquor taxes on alcoholic drinks provide an important source of revenue to various countries, measures are taken to make ethyl alcohol unfit for drinking purposes if it is intended for industrial uses. These measures are referred to as denaturation, and substances used for denaturation are known as denaturant.

1.1 Properties Required as Denaturant

Generally, following properties are required as a denaturant:

- It easily blends with ethyl alcohol, and its presence will not adversely affect its industrial applications at all.
- It can give ethyl alcohol sufficient obnoxious malodor that will effectively prevent its use for beverage purposes.
- It cannot easily be separated from ethyl alcohol by filtration, distillation or any other simple treatment.
- Its presence even in extremely small quantity can easily and positively be detected.
- It is low priced and readily available.

1.2 Denaturant

Specific products of denaturant will be described by taking examples in Japan.

1.2.1 General denaturation

General denaturation is a denaturation standard that is used for general purposes. Major denaturants include methyl alcohol, benzene, toluene, fragrances, etc.

1.2.2 Selective denaturation

This is a denaturation standard which allows use of denaturants other than those for general denaturation purposes when industrial ethyl alcohol is used in manufacture of acetylene derivatives, ether, ester, etc. As selective denaturants, industrial ethyl ether, kerosene, formalin, ethyl acetate, etc. are used.

1.2.3 Special denaturation

Use of special denaturants is mandatory for some of ether and ester products. Benzene, methyl ethyl ketone, etc. are used for this purpose.

1.2.4 Exceptional denaturation

In case denaturation standards described under 1.2.1 through 1.2.3 cannot be used, the reasons for it must be stated in an "application for use of exceptional denaturants" and submitted to the Minister of International Trade and Industry for approval. The approved method is referred to as exceptional denaturation.

2. Industrial Use

Many countries use ethyl alcohol in the chemical industry for solvents and other products, food industry for preservatives, and pharmaceutical industry for strilizing alcohol, etc. Some countries extensively use ethyl alcohol as basic raw materials in chemical industry for production of ethylene, acetaldehyde, etc. However, it is difficult to apply the same category to explain every country's industrial use of ethyl alcohol because classifications of application vary depending on each country. Applications also slightly differ from one country to another. For these reasons, examples in Japan will be introduced as regards applications in chemical industry, beverage and food industry, hygienic and pharmaceutical industry. Ethyl alcohol is not used in Japan as basic raw materials for chemical industry. For these applications, examples in other countries which use ethyl alcohol for such purposes will be cited. Ethylene can be produced by two processes. One is to produce from

ethyl alcohol. The other generally practiced method is thermal cracking of naphtha or natural gas (known as petrochemical process). Production costs by these methods have been calculated and compared.

Use of either fermentation process or synthesis process for particular industrial application slightly differs according to countries. However, generally, ethyl alcohol produced by fermentation method is used for human intake purposes such as food and drink additives like preservatives, vinegar, perfumery, and part of medicines under government regulations. Some countries restrict use of synthetic ethyl alcohol even for cosmetics and other products which come in contact with human body under their agricultural policy to protect farmers who produce raw materials for fermentation ethyl alcohol.

2.1 Chemical Industry Use

Ethyl alcohol is mainly used for solvents in chemical industry. It is lower in toxic effect on human body than other solvents such as acetone and benzene. For this reason, it is almost exclusively used in the field of solvents that will come in contact with human body such as soap, kitchen and clothing detergents, cosmetics, and aerosol products.

Aside from use as solvents, ethyl alcohol is used for production of ethyl esters including ethyl acrylate, ethyl malonate, ethyl silicate, ethyl phthalate, and ethyl cyanoacetate for textile processing and high-class paints as well as for production of intermediates of agricultural chemicals, medicines and dyestuff.

2.2 Beverage and Food Use

Ethyl alcohol is used as raw material for alcoholic drinks and vinegar, as well as preservatives for soybean sauce, pickles, ice cream, fresh cake, etc.

2.3 Hygienic and Medical Supplies

Ethyl alcohol is used in manufacture of vitamins, medicines for external use, antibiotics, amino acid drugs, etc. It is also used as alcohol of Japanese Pharmacopoeia.

2.4 Basic Raw Material for Chemical Industry

Another different use of ethyl alcohol besides its use in chemical industry such as for solvents described under 2.1, there

is a field that uses ethyl alcohol as basic raw material to produce a series of ethyl alcohol derivatives. For the present survey, this particular field will be referred to as basic raw material for chemical industry, to distinguish it from its use in chemical industry described in 2.1.

This field of applications is mainly found in Brazil and India because both countries are oil importers and abundantly produce agricultural products.

Production method for ethyl alcohol derivatives is usually divided into the following three types:

- To produce ethylene and its derivatives from ethyl alcohol
- To produce acetaldehyde and its derivatives from ethyl alcohol
- To produce derivatives directly from ethyl alcohol

Fig. C-1 to C-3 show a block flow chart for each of these production processes.

Fig. C-1 Production of Etylene and Its Derivatives from Ethyl Alcohol

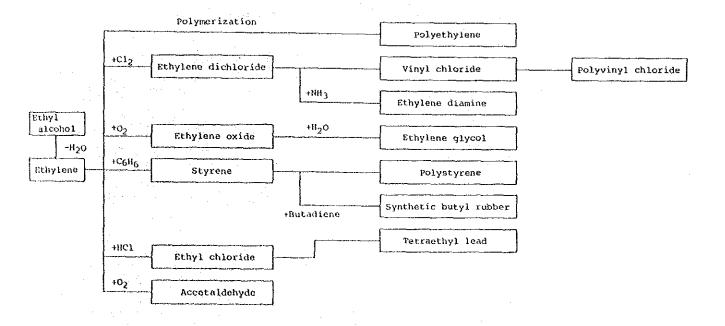


Fig. C-2 Production of Acetaldehyde and Its Derivatives from Ethyl Alcohol

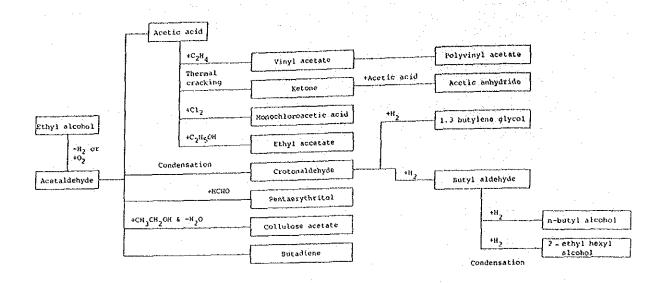
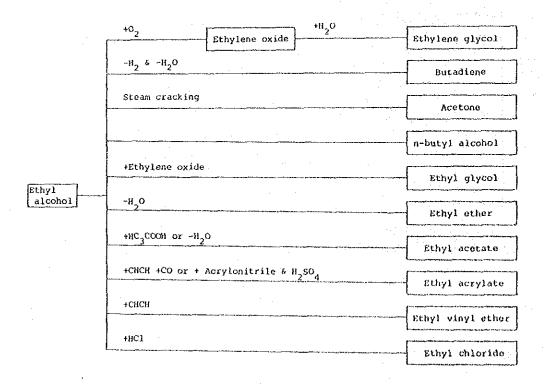


Fig. C-3 Production of Derivatives Directly from Ethyl Alcohol



2.4.1 Examples of actual applications as basic raw material for chemical industry

(1) Brazil

Brazil leads the world in this field. According to 1980 UNIDO data, with ethyl alcohol as raw material, Brazil produces ethylene, acetaldehyde, octyl alcohol, butyl alcohol, budadiene, butyl ether, ethylene glycol, diethylene glycol, etc.
For details including production capacity, refer to Appendix Table 4.

(2) India

According to 1980 UNIDO data, India produces acetic acid, acetic anhydride, butyl acetate, ethyl acetate, monochloro-acetic acid, pentaerythritol, DDT, styrene, polyethylene, acetone, butyl alcohol, butadiene, etc. For details including production capacity, refer to Appendix Table 5.

(3) Peru

Peru also produces ethylene and polyvinyl chloride.

2.4.2 Comparison of ethylene production costs

As described above, ethylene can be produced from ethyl alcohol. However, it is usually produced from naphtha, a fraction of petroleum, or from natural gas.

Ethylene is one of basic raw materials for petrochemical industry and is being consumed in large quantities. Thus, if ethylene produced from ethyl alcohol is competitive with ethylene from petrochemical process, a huge market for industrial use can be expected for ethyl alcohol.

Cost comparison will be made below between ethylene produced from ethyl alcohol and that from petrochemical process. As the method for this cost comparison, production cost of ethylene with ethyl alcohol as raw material will be calculated first. Results will then be compared with the market price of ethylene from petrochemical process as described in the section dealing with price of raw material.

(1) Premises for cost calculation

(a) Plant capacity and process
To be 60,000 tons/year and isothermal reaction fixed bed
type process.

- (b) Purity of product
 There are two grades for ethylene: polymer grade and
 chemical grade. Of these grades, stricter polymer grade
 has been employed (99.95 mol %).
- (c) Number of operating days
 To be set at 330 days a year.
- (d) Plant investment cost 8 million US dollars as estimated by Study Team.
- (e) Raw material and utilities consumption Refer to Table C-1.

Table C-1 Raw Material and Utilities Consumption (Isothermal reaction fixed bed type process)

	Polymer Grade
Raw Material 100% Ethyl alcohol per ton of Ethylene (MT)	1.75
Utilities Consumption (per ton of ethylene)	
Steam (MT) Power (kWh) Cooling Water (m³) Fuel (kcal) Catalyst (kg)	1.08 235 180 0.360 x 10 ⁶ 0.03 - 0.1

Source: Collection of Development and Technology Data on Fuel and Chemical Raw Materials by Biomass

- (f) Price of raw material Calculation was made for two cases: \$416/kl which is average cost of ethyl alcohol by the fermentation process and \$279/kl which is the production cost in Brazil.
- (g) Number of operation personnel Set at 40 persons.
- (h) Others Premises used for production cost calculation on ethyl alcohol were used.

(2) Results of cost calculation

Results of cost calculation are shown in Table C-2. For further details refer to Appendix Table 6.

Table C-2 Average Cost of Ethylene Produced from Ethyl Alcohol

		(US\$/MT)	
		World Average	Brazil
Variable costs	Raw material costs	910	611
	Utilities and chemicals costs	81	81
Fixed costs		32	32
Total		1,023	724

(3) Cost comparison

For the production cost of ethylene from ethyl alcohol, the world average and results of calculation on the Brazilian case described above were used. Production cost of ethylene from petrochemical process has not been published. Therefore, the market price was used as reference.

Table C-3 Cost Comparison

	# W. W.		(US\$/MT)
Calculation of Cost of Ethylene from Ethyl Alcohol		Cost of Ethylene from Petrochemical Process (Market Price)	
		· · · · · · · · · · · · · · · · · · ·	
World Average	1,023	Japan	800
Brazil	724	Europe	7 50
and the second second		USA	490

From Table C-3, it will be noted that the average world production cost of ethylene from ethyl alcohol is higher than the market prices in Japan, Europe and the United States.

However, when production cost calculation is made for the example in Brazil where ethyl alcohol, raw material of ethylene, is available at lower cost, we see that ethylene production cost has already reached the level of that in Japan and Europe where naphtha is used as feedstock for petrochemical process, while in the United States where natural gas is used as raw material for ethylene the cost of ethylene from ethyl alcohol is still higher than from petrochemical process. Production of ethylene from ethyl alcohol offers the following advantages:

- Raw material for fermentation process is a reproducible resource such as molasses and sugar cane.
- Ethylene is a sole product of this process and no by-products are produced as in the case of petrochemical process. This will simplify process configuration, and greatly facilitate plant operation and acquisition of operation techniques. In addition, equipment investment cost will be made smaller in scale.
- Savings on foreign exchange by substituting for petroleum, as well as promotion of agriculture, can be accomplished.

Aside from these advantages, ethylene from ethyl alcohol will become more advantageous in cost over ethylene from petrochemical process if oil price rises in future. Thus, production of ethylene from ethyl alcohol is considered an extremely significant technology in countries where molasses, sugar cane, etc. as raw materials of ethyl alcohol are available at low cost.

However, it is also required that production of ethylene from ethyl alcohol should be studied in each country under the framework of a comprehensive industrial policy which will cover use of whole products in the downstream of ethylene production.

3. Fuel Use

Use of ethyl alcohol as fuel is the latest use that emerged against a backdrop of soaring petroleum prices. Ethyl alcohol is now consumed in large volumes in Brazil, the United States, etc. by being blended with gasoline or simply as it is.

Merits and demerits of ethyl alcohol when blended with gasoline as automotive fuel are generally said to be as follows:

Merits:

- As ethyl alcohol blended is produced from reproducible

resources, it is desirable from the viewpoint of ecological environment.

- If surplus agricultural products are used as raw material, it will help expand and stabilize the market for agricultural products.
- As ethyl alcohol resources are undepletable, it will lead to conservation of oil.
- It improves octane vating of fuel. (When 10% of ethyl alcohol is blended into gasoline, the octane number improves by two or three points. In the United States where stepwise control on the use of alkyl lead is being implemented, ethyl alcohol as octane rating improver is now gaining more attention.)
- It lowers the mixed vapor temperature and improves filling efficiency (increase in blended volume).
- It lowers combustion temperature and reduces $NO_{\mathbf{X}}$ (less polution by exhaust gas than gasoline).
- Addition of anhydrous ethyl alcohol up to around 20% hardly affects engine performance so that conventional engine can be used as it is.

Demerits:

- When water content in blended fuel reaches 5 to 6%, phase separation tends to occur to make it unusable as blended fuel.
- While gasoline consists of many components with the boiling range between 20-200°C, ethyl alcohol comprises single component with low boiling point of 78.3°C. Because of this, blended fuel tends to cause vapor lock and percolation more often than gasoline (Measures to modify fuel feed system are necessary).
- As ethyl alcohol has a high latent heat of vaporization, its saturated vapor is too lean to make starting in cold area difficult (Requires equipment of preheater or the use of auxiliary fuel to support starting).
- As its heat of combustion is low at 60% compared to gasoline, the capacity of fuel tank must be made larger. Also the tank may rust because of its absorption property.
- Its exhaust gas contains acetaldehyde which has an irritating odor.
- Its price still compares unfavorably to gasoline.

Regarding its use as fuel aside from the use as automotive fuel in either the blended form with gasoline or in the form of single substance, studies are being made to use it for internal combustion engine by blending with gas oil or as boiler fuel in various countries.

We would like to introduce the interim report of the study commissioned by the Resources and Energy Agency of the Japanese Ministry of International Trade and Industry conducted in Japan with respect to the adaptability of ethyl alcohol to various applications as compared to existing fuels in the following paragraphs. It sould be noted however that this study was conducted under the particular circumstances of Japan with respect to oil, environmental pollution control regulations and meteorological conditions so that its findings may not necessarily apply to other countries with different conditions.

3.1 Substitution for Gasoline

The subject of this study was alcohol blended gasoline which is produced by blending either ethyl alcohol or methyl alcohol with gasoline. The study was directed to specific points discussed below. In the following description, alcohol is defined to mean both ethyl alcohol and methyl alcohol.

- Change in basic properties as liquid fuel when gasoline is blended with alcohol.
- Effect on human health and environment if alcohol blended gasoline is used on a large scale.
- Compatibility with Existing Systems
 Effects of properties of the blend on storage, deterioration
 of quality in the distribution stages, materials and driving
 performance of automobiles, etc.
- Effects on economy, industry and social system.

Although the study is still in the stage of interim report, the following is an outline of its findings:

3.1.1 Basic properties

(1) Octane number and heat of combustion

Octane number will improve when gasoline is blended with alcohol.

- (a) Improvement in reformed gasoline is greater than that in cracked gasoline.
- (b) Blending ratio and the rise in octane number hold a substantially linear relationship.
- (c) As far as octane number is concerned, no significant difference exists between methyl alcohol and ethyl alcohol.
- (d) In the range where alcohol content in the blend is greater, improvement in Motor octane number is greater than that in Research octane value. As for heat of combustion, alcohol is lower than gasoline and therefore the heat of combustion drops in proportion to the increase in the ratio of alcohol blended.

(2) Increase in volatility

By blending alcohol with gasoline, vapor pressure will rise due to azeotropic phenomenon. Boiling point at front-end and intermediate fractions in distillation curve will drop while vapor-liquid ratio will increase. Increase in volatility is greater for ethyl alcohol than for methyl alcohol.

As measures against this problem, studies have been made on the effect of increasing higher boiling range fractions in gasoline, etc. However, this attempt has not yet produced adequately satisfactory effect.

(3) Quality stability

No particular change is noted when alcohol is blended with reformed gasoline with smaller olefin contents. When alcohol is blended with cracked gasoline with larger olefin contents, there is the possibility of evaporation residues being formed in the mixture during storage. Oxidation stability test has also indicated reduction in stability period.

(4) Solubility, separation of water and solubilizer

Alcohol and gasoline will well mix with each other in normal temperature range but commingling of a small amount of water will separate the mixture into two phases.

The quantity of water that will cause phase separation (threshold moisture value) varies depending on type of alcohol, blended volume, temperature, etc. Generally, if content of ethyl alcohol in the mixture is less than 10%, threshold moisture value causing phase separation is more than about 0.5% at 0°C. It is considered that phase separation can be prevented by reducing water content in ethyl alcohol, or taking effective steps to prevent mixture of water into fuel in distribution and application stages.

As regards methyl alcohol, when its content in the mixture is 10%, threshold moisture value causing phase separation is about 0.1% at 0°C. Since presence of water in such an extremely small amount causes phase separation, sufficient measures against infiltration of water must be taken.

Higher alcohols are highly effective as solubilizer that will increase solubility of alcohol blended gasoline and will inhibit phase separation due to mixture of water.

3.1.2 Effect on human health and environment

To study effect on human body from labor safety hygienic standpoints, test was conducted on rats for acute and subacute toxicity, as well as for mutagenicity. Test results showed no difference from those on conventional gasoline.

For effect on environment, tests were conducted to study decomposition by microorganisms, intake by and accumulation in fish, and effect on aquatic plants. However, no difference from conventional gasoline was noted.

3.1.3 Compatibility with existing systems

(1) Effect on distribution equipment

Among distribution equipment currently in use, there are some that are not suited for alcohol blended gasoline. However, this problem can be technically solved. Thus there is fundamentally no problem in this area.

(2) Effect on driving performance of automobiles

Automobiles fueled by alcohol blended gasoline presented no particular problems in driving performance. However, under environment control and meteorological conditions in Japan, some deterioration in driving comfort was noted. When changes in fuel properties such as phase separation occur, this affects driving performance.

(3) Effect on functions of automobiles

When tests with alcohol blended gasoline were conducted on vehicles now on market without any adjustments or modifications for this fuel, results showed following problems: some vehicles cannot completely satisfy the emission standards in Japan. On some cars, low temperature starting characteristics are adversely affected, as well as normal temperature and high temperature drivability. And there is the possibility that faults or failures will occur in fuel system parts and material, thereby lowering reliability and stability.

Based on these results, study was made on the possibility of technical countermeasures.

3.1.4 Effect on economic, industrial and social systems

With technical study on use of alcohol blended gasoline as a base, effect on social and economic systems was assessed by means of a simulation model.

3.2 Substitution for Kerosene, Gas Oil, and Fuel Oil

Use of alcohol not only as substitute for gasoline, but as substitute for middle distillate petroleum products, i.e. kerosene, gas oil and fuel oil, is drawing attention worldwide. However, its study has just begun and technical data on this study has not been accumulated as much as on the study of alcohol blended gasoline.

Investigation in Japan is directed to the understanding of basic properties resulting from blending of alcohol with middle distillates. The following is an outline of the results of this investigation:

3.2.1 Basic properties of ethyl alcohol blended kerosene

(1) Basic properties

- (a) Ethyl alcohol's boiling point is 78.3°C which is much lower than kerosene's. As a result, ethyl alcohol alone tends to evaporate as liquid temperature of blended kerosene rises.
- (b) Blended kerosene's flashing point is equal to ethyl alcohol's at 13°C so that blended kerosene will be more easily ignited than kerosene alone.
- (c) Heat of combustion of ethyl alcohol blended kerosene will drop by about 3.5% per every 10% increase in ethyl alcohol content.
- (d) There are no problems in oxidation stability and storage stability compared with kerosene.

(2) Phase separation and solubilizer

With increase in ethyl alcohol content in the mixture, threshold moisture value causing phase separation will increase. When ethyl alcohol content is changed from 10, to 20 and to 30 vol %, threshold moisture value causing phase separation changes from 0.10 to 0.15 and 0.22 vol %, respectively. Higher alcohols exhibited excellent effect as solubilizer.